



# Article Enhanced Oil Recovery and CO<sub>2</sub> Storage Performance in Continental Shale Oil Reservoirs Using CO<sub>2</sub> Pre-Injection Fracturing

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Abstract: CO<sub>2</sub> pre-injection fracturing is a promising technique for the recovery of continental shale oil. It has multiple advantages, such as oil recovery enhancement, CO<sub>2</sub> geological storage and water consumption reduction. Compared with conventional CO<sub>2</sub> huff and puff and flooding, CO<sub>2</sub> pre-injection features higher injection rates and pressures, leading to EOR and improved CO<sub>2</sub> storage performance. Combining physical experiments and numerical simulation, this research systematically investigated the EOR and storage performance of CO<sub>2</sub> pre-injection in continental shale reservoirs. The results showed that CO<sub>2</sub> pre-injection greatly improved the oil recovery; after seven cycles of soaking, the average oil recovery factor was 39.27%, representing a relative increase of 31.6% compared with that of the conventional CO<sub>2</sub> huff and puff. With the increasing pressure, the CO<sub>2</sub> solubility grew in both the oil and water, and so did the CO<sub>2</sub> adsorption in shale. Numerical simulation indicated that the average CO<sub>2</sub> storage ratio of the production stage was 76.46%, which validated the effectiveness of CO<sub>2</sub> pre-injection in terms of CO<sub>2</sub> geological storage.

Keywords: carbon neutrality; shale oil; CO<sub>2</sub> pre-injection; enhanced oil recovery (EOR); CO<sub>2</sub> storage

# 1. Introduction

The Intergovernmental Panel on Climate Change (IPCC) of the United Nations reports that limiting global warming to 2 °C requires the global artificial carbon dioxide (CO<sub>2</sub>) emissions to be reduced by a quarter compared with those in 2010 before 2030 at the latest and achieve net zero CO<sub>2</sub> emissions by 2075; for a more rigorous goal of 1.5 °C, net zero CO<sub>2</sub> emissions need to be achieved globally by 2050 [1]. Besides promoting applications of techniques of energy-saving, carbon emission reduction and renewable energy, CO<sub>2</sub> capture, utilization and storage (CCUS) is also one of the effective approaches to deliver net zero CO<sub>2</sub> emissions [2–4]. In the context of China's clearly stated goals of carbon peak and neutrality, CCUS becomes increasingly important for China.

Geological sites applicable to  $CO_2$  storage include saline aquifers, unrecoverable coal seams, and oil and gas reservoirs [5–9].  $CO_2$ -EOR projects performed in oil and gas reservoirs deliver both the social benefits of carbon reduction and economic benefits to stimulated hydrocarbon production. Moreover, such projects are highly feasible due to the available existing well sites and equipment, and thus, become preferred choices for the geological storage of  $CO_2$  [10–14]. By far, the  $CO_2$ -EOR in conventional oil and gas reservoirs has reached the stage of industrial demonstration, where demonstration projects with a capacity of up to 100,000 tons have been successively accomplished in the Daqing, Jilin and Shengli oilfields.  $CO_2$  features unique physical and chemical properties. Under the



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**Copyright:** © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). formation temperature and pressure, it is typically supercritical, with an extremely strong diffusion capacity to enter micro-nano pores of rocks and harvest crude oil beyond the reach of conventional EOR methods. Furthermore, supercritical  $CO_2$  can extract light components of crude oil and improve oil mobility [15–17]. As  $CO_2$  and crude oil mix, plenty of  $CO_2$  dissolves into oil, which considerably reduces the oil viscosity and enhances oil recovery via various mechanisms, such as the resultant crude oil volume expansion and miscibility between oil and  $CO_2$ . Besides the above-stated,  $CO_2$  is clean and environment-friendly. It has great application potential in the exploration and development of unconventional hydrocarbon resources [18–22].

 $CO_2$  huff and puff demands a miscible state between  $CO_2$  and formation crude oil to maximize the recovery factor [23–26]. However, the shale oil plays of China are mainly continental [27–29], the miscibility pressure of crude oil is high and miscible flooding cannot be delivered; in some shale oil reservoirs, the minimum miscibility pressure (MMP) is even higher than the formation pressure. Even in shale oil reservoirs with the MMP lower than the bottomhole pressure,  $CO_2$  flooding is still faced with the challenge that miscibility can only be delivered in the near-wellbore zone and the consequently restrained EOR performance due to the limits of the reservoir's physical properties, crude oil properties, equipment capacity and technical schemes [30,31]. Given the aforementioned, the EOR technique based on  $CO_2$  pre-injection was proposed to take full advantage of  $CO_2$  in EOR. Specifically,  $CO_2$  is injected into the reservoir at high injection pressures and high pump rates during fracturing to keep the formation pressure above the miscibility pressure over a rather long period, expand the swept zone of  $CO_2$  and ultimately enhance the recovery factor of shale oil.

Extensive studies have been carried out to investigate the mechanisms and sequestration effects of  $CO_2$ -EOR with low injection pressures and rates. However, the EOR and sequestration mechanisms of  $CO_2$  pre-injection during fracturing are less researched [32–34]. Therefore, this research systematically investigated the interaction mechanisms between  $CO_2$  and crude oil in the case of high-pressure fracturing of continental shale oil reservoirs via laboratory experiments. It shall be noted that the experiments stated below were all performed at high pressure. Furthermore, the storage performance of  $CO_2$  pre-injection at the field scale was analyzed via numerical simulation. The findings of this research validate the effectiveness of EOR and sequestration of  $CO_2$  pre-injection during fracturing in continental shale oil reservoirs and provide theoretical support for applications of this technique in shale oil recovery.

#### 2. Experimental Section

#### 2.1. Sample Preparation

The shale samples used in this research were collected from the Qingshankou Formation in the Songliao Basin, China. The Qingshankou Formation is predominated by mudstone-shale, with localized occurrence of interbeds of sandstone, limestone and limy mudstone. Mudstone-shale is mainly composed of quartz, clay minerals, feldspar and carbonate minerals, and clay minerals account for about 30–60%. The content of brittle minerals is relatively high, and micropores and microfractures are well developed. This area is one of China's major plays of continental shale and possesses geological resources of shale oil of about 54.6 × 10<sup>8</sup> t, which shows extremely high potential for exploration and development [35,36].

The depth of the Qingshankou Formation shale reservoir is about 1800–2400 m. The first section of the Qingshankou Formation is the current focus of shale oil recovery and has a formation temperature of 90 °C–110 °C. The experimental temperature was set to 90 °C for all experiments of this research as per the comprehensive consideration of the formation conditions and experimental apparatus capacity. The formation water salinity was determined via tests of produced water (Table 1).

**Table 1.** Ionic analysis of produced water.

Ions	Na <sup>+</sup> and K <sup>+</sup>	Ca <sup>2+</sup>	Mg <sup>2+</sup>	Cl-	$SO_4^{2-}$	$HCO_3^-$
Concentration (mg/L)	2320	39	5	1450	307	3420

The drill cores were made into cylinders with the dimensions of 5 cm  $\times$   $\Phi$ 2.5 cm via a wire-cutting process. Then, the cylindrical specimens with intact appearance and no notable cracking were selected and polished using abrasive paper to ensure smooth end and side surfaces. The polished cylinders were then put through an oil removal process, which was performed with caution to avoid damage to the core specimens. The prepared core specimens are presented in Figure 1.



Figure 1. Photo of the shale core specimens.

#### 2.2. Experimental Apparatus and Methods

Analysis of porosity and permeability: The porosity and permeability of shale were measured using a high-precision high-pressure porosimeter–permeameter. This apparatus can apply a confining pressure of 0–3000 psi and measure the porosity and permeability within the ranges of 0.1–40.0% and 0.001–10,000 mD, respectively. In this experiment, the specimens were first dried for 24 h and placed in the core holder for the measurement of porosity and permeability at the confining pressure of 3000 psi. The initial porosity and permeability of tested specimens are shown in Table 2.

Table 2. Physical properties of shale specimens.

Specimen No.	Diameter (cm)	Length (cm)	Porosity (%)	Permeability (mD)
1—Huff and puff	2.51	5.01	5.49	0.0314
2—Huff and puff	2.51	5.00	5.17	0.0224
3—Pre-injection	2.50	5.01	5.81	0.0392
4—Pre-injection	2.51	5.02	4.96	0.0271

 $CO_2$  huff and puff experiment and  $CO_2$  pre-injection experiment: First, the specimens were dried in the oven for 24 h and then placed in an intermediate container for crude oil saturation at a given pressure and temperature, during which the saturated oil volume was recorded. The specimen was then placed in the core holder, and  $CO_2$  was continuously injected at a rate of 0.1 mL/min until the pressure reached a pre-specified value, during which the confining pressure was set at the tracking mode and kept at 2 MPa higher than the injection pressure. For conventional  $CO_2$  huff and puff, the target injection pressure was set as 20 MPa, and for  $CO_2$  pre-injection, 30 MPa. Subsequently, the inlet valve was closed for  $CO_2$  soaking for 24 h. Once the soaking was completed, the inlet valve was



opened for production, and the oil production was recorded. The experimental apparatus is shown in Figure 2.

Figure 2. Schematic diagram of the EOR physical model simulation apparatus.

Microscopic morphology analysis of shale after CO<sub>2</sub> treatment: The Apreo highperformance field-emission scanning electron microscope (FE-SEM), manufactured by Thermo Scientific, was used to capture the microscale morphology variation of the shale surface after CO<sub>2</sub> treatment. This FE-SEM is equipped with an energy-dispersive spectroscopy (EDS) system and can deliver resolutions to 1 nm at an accelerating voltage of 1 kV. In this experiment, each sample with a flat and clean surface was prepared via wire-cutting, mechanical polishing and argon ion polishing. The prepared sample was then treated with CO<sub>2</sub> and observed in the FE-SEM under the conditions of an accelerating voltage of 10 kV and beam current of 1.6 nA to characterize the microscale pore structure variation of the shale surface attributed to CO<sub>2</sub>.

Analysis of CO<sub>2</sub> solubility: A high-temperature high-pressure visualization reactor for PVT analysis was used to measure the CO<sub>2</sub> solubility via the PVT method. First, a given amount of gas was injected into the visualization reactor, and the equilibrium temperature and pressure ( $T_1$  and  $P_1$ ) of gas inside the reactor were recorded after the readings of temperature and pressure became stable. Then, a fixed volume of liquid was injected into the visualization reactor via the intermediate container. The temperature and pressure ( $T_2$  and  $P_2$ ) inside the reactor were recorded after the re-balancing between gas and liquid in the reactor. Finally, the solubility of the injected gas in the injected liquid was calculated using the equation of state for gas. The experimental apparatus is shown in Figure 3.



**Figure 3.** Schematic diagram of the high-temperature, high-pressure visualization reactor for PVT analysis.

 $CO_2$  adsorption in shale: First, each shale sample was ground into micron-scale powder. The sample powder was then placed in the sample cell, where it was vacuumed for 24 h at a given temperature. Subsequently, the reference cell was pressurized to the specified pressure using  $CO_2$  and sealed, after which  $CO_2$  was injected into the sample cell. The pressure readings of both the reference and sample cells were recorded continuously until they reached equilibrium. The adsorbed  $CO_2$  in the shale was estimated according to the equilibrium pressure. The experimental apparatus is shown in Figure 4.



Figure 4. Schematic diagram of CO<sub>2</sub> adsorption experiment.

# 3. Results

3.1. EOR Performance Analysis of CO<sub>2</sub> Pre-Injection

The EOR performances of conventional huff and puff and pre-injection of  $CO_2$  were compared via laboratory experiments. The four groups of experiments were all carried out at 90 °C. The soaking pressure for conventional  $CO_2$  huff and puff was 20 MPa, while that for  $CO_2$  pre-injection was 30 MPa. The oil recovery for seven injection cycles is shown in Figure 5.



Figure 5. EOR performance comparison during laboratory experiments.

Clearly,  $CO_2$  pre-injection resulted in considerable growth in oil recovery. This was because the formation pressure after high-pressure, high-rate injection of  $CO_2$  during fracturing was much higher than that after conventional  $CO_2$  huff and puff. During injection with high pressures and high pump rates,  $CO_2$  had stronger diffusion and could enter the smaller nanoscale reservoir space to extract oil and improve the oil recovery.

In addition, previous research states that the MMP between the Qingshankou Formation crude oil and  $CO_2$  is about 27.45 MPa [31]. Hence, the miscibility of  $CO_2$  with the formation of crude oil cannot be achieved during the soaking phase of conventional  $CO_2$  huff and puff. In contrast, it can be achieved during  $CO_2$  pre-injection to enhance crude oil mobility and recovery. Also, the increased pressure leads to more EOR-effective injection–production cycles. Compared with conventional  $CO_2$  huff and puff that presents a notably reduced increment of oil recovery after six cycles, the  $CO_2$  pre-injection managed to deliver considerable growth in oil recovery even after seven cycles.

After mechanical and argon ion polishing, the shale sample surface was seen to have a flat tight matrix with only some small primary pores (1–5  $\mu$ m in diameter). The pore structure of the shale surface was greatly modified by the CO<sub>2</sub>, as shown in Figure 6. The primary pores of shale were considerably expanded via dissolution, with pore diameters growing from 1–5  $\mu$ m to 5–25  $\mu$ m. Meanwhile, large-area dissolution occurred in zones enriched with minerals that react with CO<sub>2</sub>, such as albite, potassium feldspar and calcite, which left large dissolution pores in the shale surface. The possible reactions between CO<sub>2</sub> and shale that occurred on the shale surface are shown in Equations (1)–(5). These incremental dissolution pores improved the fluid flow capacity of the shale and provided new channels for oil migration. For CO<sub>2</sub> pre-injection, the higher injection pressure and rate effectively expanded the swept volume of CO<sub>2</sub>. This meant a larger stimulated volume with enhanced pore structures and, consequently, higher oil recovery.

$$2NaAlSi_{3}O_{8} + 2CO_{2} + 3H_{2}O = 2Na^{+} + 2HCO_{3}^{-} + 4SiO_{2} + Al_{2}Si_{2}O_{5}(OH)_{4}$$
(1)

$$2KAlSi_{3}O_{8} + 2H^{+} + 9H_{2}O = 2K^{+} + 4H_{4}SiO_{4} + Al_{2}Si_{2}O_{5}(OH)_{4}$$
(2)

$$CaCO_3 + H^+ = Ca^{2+} + HCO_3^-$$
 (3)

$$CaMg(CO_3)_2 + 2H^+ = Ca^{2+} + Mg^{2+} + 2HCO_3^-$$
(4)

# $[Fe/Mg]_{5}Al_{2}Si_{3}O_{10}(OH)_{8} + 5CaCO_{3} + 5CO_{2} = 5Ca[Fe/Mg](CO_{3})_{2} + Al_{2}Si_{2}O_{5}(OH)_{4} + SiO_{2} + 2H_{2}O$ (5)



Figure 6. SEM analysis of shale samples: (a,c) before CO<sub>2</sub> treatment and (b,d) after CO<sub>2</sub> treatment.

The photos of shale core specimens before and after  $CO_2$  pre-injection are shown in Figure 7. Multiple highly developed horizontal fractures occurred in the shale specimen, which were all gradually generated with the increasing injection cycles; the fractures became exceedingly notable after four injection-production cycles. Such induced fractures expanded the contact area between the  $CO_2$  and shale, which effectively enlarged the affected zone of  $CO_2$  and enhanced the oil recovery.



Figure 7. Photos of a shale core before and after CO<sub>2</sub> pre-injection.

The performed physical simulation experiment using shale cores demonstrated the advantages of  $CO_2$  pre-injection for EOR. In the case of the conventional  $CO_2$  huff and puff, the average oil recovery was 29.84% after seven cycles, while that of  $CO_2$  pre-injection reached 39.27%, which was a relative growth of 31.6%. At higher pressure,  $CO_2$  was able to penetrate deeper into the cores, create induced fractures to expand its contact with the shale and deliver miscible mixing with oil to improve the oil mobility, all of which jointly enhanced the oil recovery.

## 3.2. Analysis of CO<sub>2</sub> Solubility in Formation Fluids of Shale Oil Reservoirs during Pre-Injection

The CO<sub>2</sub> solubility in the formation fluids was experimentally measured at 90 °C with different pressures to characterize the effects of pressure on CO<sub>2</sub> solubility in the formation water. The phase equilibrium data in the PVT reactor were recorded and used to calculate CO<sub>2</sub> solubility, as shown in Figure 8. Clearly, at the formation temperature, the CO<sub>2</sub> solubility in the formation water grew with the increasing pressure, though this growth slowed down gradually. The solubility at the formation temperature presented greater variations in cases of lower pressure, and an inflection point occurs around 15 MPa, which represented the start of the slowing down of the solubility growth.



Figure 8. CO<sub>2</sub> solubility in formation water vs. formation pressure.

Also, the variation in  $CO_2$  solubility with formation water salinity was analyzed at the formation temperature and pressure (Figure 9). At a given temperature and pressure, the  $CO_2$  solubility declined with the increasing water salinity. This is explained below:  $CO_2$  generally exists in solvation shells of water, and ions alter the  $CO_2$  solubility by changing the properties of water (the solvent), such as the hydrogen bond and translation and rotation activities. For instance, chloride ions considerably slow down water molecules and strengthen hydrogen bonds, which endows NaCl with the salting-out effect and leads to the degradation of  $CO_2$  solubility.



Figure 9. CO<sub>2</sub> solubility in formation water vs. salinity.

The effects of pressure on  $CO_2$  solubility in crude oil were investigated. The corresponding experiments were performed at 90 °C and different pressure, during which the phase equilibrium data of the PVT reactor were recorded to compute the  $CO_2$  solubility (Figure 10). At the formation temperature, the solution gas–oil ratio of  $CO_2$  in oil increased with the increasing pressure, while the growth amplitudes gradually reduced. This was because the pressurization reduced the oil volume and increases the oil density, and the narrower gaps between oil molecules were unfavorable for the  $CO_2$  solution.



Figure 10. CO<sub>2</sub> solubility in crude oil vs. formation pressure.

To sum up, with the increasing formation pressure,  $CO_2$  solubility grew in the formation fluids. The CO<sub>2</sub> solubility in crude oil is about 90 m<sup>3</sup>/m<sup>3</sup> in the case of conventional CO<sub>2</sub> huff and puff (20 MPa). Due to the tightness of shale reservoirs, the oil recovery factor is typically below 10%, which means a tremendous volume of CO<sub>2</sub> can be stored in such reservoirs by dissolving it into formation fluids. As stated above, the injection pressure and rate of CO<sub>2</sub> pre-injection were far higher than those of conventional CO<sub>2</sub> huff and puff. Moreover, more extensive and more effective contact between CO<sub>2</sub> and formation fluids was expected at the higher injection pressure (30 MPa and above). These jointly increased the dissolved CO<sub>2</sub>, and thus, improved the storage performance of CO<sub>2</sub>.

# 3.3. Analysis of CO<sub>2</sub> Adsorption in Shale Oil Reservoirs during Pre-Injection

The adsorption isotherms of  $CO_2$  in shale at different pressures were analyzed to characterize the effects of pressure on the adsorbed  $CO_2$  volume in shale (Figure 11). With the growing formation pressure, more  $CO_2$  molecules were adsorbed onto the shale surface. It should be noted the adsorption growth was much faster at medium–low pressure, while the adsorbed volume gradually became stable at a higher pressure and presented no notable further growth. For  $CO_2$  pre-injection during fracturing,  $CO_2$  was injected at high pressure and high pump rates, after which the formation pressure was considerably higher than the resultant formation pressure of conventional  $CO_2$  huff and puff. It was this higher formation pressure that led to the growth in  $CO_2$  adsorbed onto the shale surface and improved the  $CO_2$  geological storage performance.



Figure 11. CO<sub>2</sub> adsorption in shale vs. formation pressure.

In the case of  $CO_2$  pre-injection, the formation temperature was considerably reduced, as a massive volume of liquid  $CO_2$  was injected into formations at high pressures and pump rates during fracturing. This phenomenon cannot be overemphasized, as previous research demonstrated that temperature has great effects on the physical gas adsorption capacity of shale. In other words, the adsorbed quantity of gas changes with temperature. Given this, the  $CO_2$  adsorption isotherms in shale at different temperatures were measured to characterize the effects of temperature on  $CO_2$  adsorption (Figure 12). The adsorbed  $CO_2$  reduced with elevated temperature. Higher pressure led to easier desorption and harder adsorption of gas in shale. Accordingly, the  $CO_2$  adsorption in shale could be effectively enhanced by the sufficient cooling of formations due to  $CO_2$  pre-injection, which further improved the  $CO_2$  storage performance.



Figure 12. CO<sub>2</sub> adsorption in shale vs. formation temperature.

#### 3.4. Numerical Simulation of CO<sub>2</sub> Storage by Pre-Injection

To further validate the CO<sub>2</sub> storage performance of CO<sub>2</sub> pre-injection in shale oil reservoirs, the numerical simulation was performed for the Qingshankou Formation shale oil reservoir in the Songliao Basin. Specifically, block-wide geological modeling was done for the H block, in which fractures were simulated via localized mesh refinement—the properties of the near-wellbore refined meshes, such as permeability and flow conductivity, were modified to resemble fractures. The constant half-length of the main fracture was 175 m, and that of the branch fracture was 25 m. The spacing of the branch fracture was 10 m, and the main fracture permeability was set to 100 mD, as per the convergence performance of computation. Such mesh refinement was performed for three treatment wells, namely, H-4, H-3 and H-7. The presented simulation covered an area of 4.45 km<sup>2</sup> and involved 46,852 meshes. All oil producers and water injectors within the study area were incorporated, and the oil producers were operated at a constant bottomhole pressure.

The simulated variations of  $CO_2$  recovery percentage with time of the three wells are shown in Figure 13. The ultimate  $CO_2$  storage ratios of wells H-4, H-3 and H-7 were 83.87%, 71.57% and 73.77%, respectively, with an average of 76.46%, which represents excellent storage performance. In addition, the produced  $CO_2$  could be collected at the wellhead and re-injected during the next treatments after a simple separation.

It was demonstrated that  $CO_2$  pre-injection featured higher injection pressure and pump rates, which enhanced the reservoir rock penetration and pore diffusion of  $CO_2$ . Under such circumstances, the swept zone of  $CO_2$  was greatly expanded, and  $CO_2$  could enter micro-nano pores beyond the reach of conventional  $CO_2$  huff and puff. After soaking, such  $CO_2$  was trapped in the pores and formation fluids of shale reservoirs due to the interfacial tension, dissolution and adsorption, which produced effective  $CO_2$  geological storage.



Time/D

Figure 13. Recovery percentage of CO<sub>2</sub> vs. time.

#### 4. Discussion

Unconventional oil and gas reservoirs, unlike conventional ones, sequestrate  $CO_2$  mainly via four mechanisms: the storage of free  $CO_2$  in microfractures and micro-nano pores of rocks, the  $CO_2$  adsorption onto surfaces of minerals and organic matter, the  $CO_2$  solution in formation fluids and the  $CO_2$  mineralization precipitation. The contributions of these four mechanisms to  $CO_2$  storage change considerably, as  $CO_2$  constantly migrates in shale (via flow and diffusion) and repeatedly contacts the shale surface, and so does the main occurrence of  $CO_2$  in shale.

In CO<sub>2</sub> pre-injection, the complex fracture network, which is derived from fracturing, provides more microfractures and macropores for CO<sub>2</sub> accommodation and increases the stored free-state CO<sub>2</sub>. Meanwhile, the high injection pressure pushes CO<sub>2</sub> toward the unstimulated shale matrix far from the well, which increases the contributions of the adsorption and solution storage mechanisms. In the subsequent soaking and production, CO<sub>2</sub> interacts with the shale matrix and formation fluids, during which it dissolves into formation fluids and adsorbs onto the surfaces of the shale matrix and organic matter. On the one hand, owing to the extremely low recovery factor of shale oil (in most cases, 4–8%), plenty of crude oil remains in shale reservoirs, even in the late stage of development. On the other hand, a good deal of CO<sub>2</sub> can be adsorbed onto the surfaces of the shale matrix and organic matter due to the extremely intensive adsorption between CO<sub>2</sub> and these materials. Therefore, CO<sub>2</sub> pre-injection fracturing can greatly enhance shale oil recovery and, in the meantime, achieve effective geological storage of CO<sub>2</sub>.

The field pilot program showed after fracturing with  $CO_2$  pre-injection, the oil well cumulatively produced oil over 1000 days, with a cumulative oil production of about 11,000 tons and the  $CO_2$  storage ratio of this stage surpassing 80%. However,  $CO_2$  preinjection is associated with a series of challenges, such as the crude oil emulsion, pipeline corrosion and the collection of produced  $CO_2$ . Furthermore, the injection of a large amount of  $CO_2$  depends on a mature  $CO_2$  pipeline network. All of these may cause extra costs of operation and maintenance [37,38]. Despite these issues needing to be solved,  $CO_2$ pre-injection is an effective reservoir stimulation method for the high-efficiency green development of shale oil.

# 5. Conclusions

This research involved a systematic experimental investigation of the EOR and  $CO_2$  geological storage of high-pressure  $CO_2$  injection in shale oil reservoirs. Moreover, nu-

merical simulation was carried out to comprehensively analyze the advantages of  $CO_2$  pre-injection concerning  $CO_2$  storage in shale oil reservoirs. The following conclusions were drawn:

- (1) After seven cycles of conventional CO<sub>2</sub> huff and puff, the average oil recovery was 29.84%, while that of CO<sub>2</sub> pre-injection was 39.27%, showing a relative growth of 31.6%. At higher injection pressures and pump rates, CO<sub>2</sub> could diffuse deeper into the cores, create induced fractures to expand its contact area with the shale and mix with crude oil at the miscible state, all of which jointly led to considerable enhancement of shale oil recovery.
- (2) With increasing pressure, the CO<sub>2</sub> solubility grew considerably in both oil and water and so did the quantity of the CO<sub>2</sub> molecules adsorbed onto the shale. With the high injection pressure and rate of CO<sub>2</sub> pre-injection, more extensive and more effective contact between the CO<sub>2</sub> and formation fluids was expected, and this improved the CO<sub>2</sub> storage performance.
- (3) Numerical simulation validated the effectiveness of CO<sub>2</sub> pre-injection, regarding CO<sub>2</sub> geological storage. The simulation results showed that CO<sub>2</sub> could flow further into the reservoirs (leading to a larger swept zone) and enter micro-nano pores that were beyond the reach of the conventional CO<sub>2</sub> huff and puff. Excellent storage performance was seen in the production stage, as the average storage ratio of CO<sub>2</sub> reached 76.46%. CO<sub>2</sub> pre-injection has tremendous potential for the geological storage of CO<sub>2</sub>.

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