



# **Review Recent Advances in Biomass Pretreatment Technologies for Biohydrogen Production**

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Abstract: Hydrogen is an economical source of clean energy that has been utilized by industry for decades. In recent years, demand for hydrogen has risen significantly. Hydrogen sources include water electrolysis, hydrocarbon steam reforming, and fossil fuels, which emit hazardous greenhouse gases and therefore have a negative impact on global warming. The increasing worldwide population has created much pressure on natural fuels, with a growing gap between demand for renewable energy and its insufficient supply. As a result, the environment has suffered from alarming increases in pollution levels. Biohydrogen is a sustainable energy form and a preferable substitute for fossil fuel. Anaerobic fermentation, photo fermentation, microbial and enzymatic photolysis or combinations of such techniques are new approaches for producing biohydrogen. For cost-effective biohydrogen production, the substrate should be cheap and renewable. Substrates including algal biomass, agriculture residue, and wastewaters are readily available. Moreover, substrates rich in starch and cellulose such as plant stalks or agricultural waste, or food industry waste such as cheese whey are reported to support dark- and photo-fermentation. However, their direct utilization as a substrate is not recommended due to their complex nature. Therefore, they must be pretreated before use to release fermentable sugars. Various pretreatment technologies have been established and are still being developed. This article focuses on pretreatment techniques for biohydrogen production and discusses their efficiency and suitability, including hybrid-treatment technology.

Keywords: biohydrogen; feedstock; fermentation; fossil fuel; global warming; substrate pretreatment

# 1. Introduction

Currently, the world is facing severe energy problems. Global energy demands have been met through the use of non-renewable energy sources. Excessive burning of fossil fuels has caused global climate change with disastrous effects. Biohydrogen could strategically serve as an alternative to non-renewable energy because it is a clean energy carrier with low emissions. It is renewable, clean, has high energy content, and does not contribute to the accumulation of greenhouse gases (GHGs) [1]. Due to its eco-friendly nature, hydrogen production through biological processes utilizing biomass is one of the most remarkable techniques [2]. Anaerobic fermentation processes are desirable since hydrogen production is enhanced and obtained from organic wastes or sewage water emended with carbohydrates as substrate. The rate of production of biohydrogen depends upon the kind of substrates and conversion technologies utilized. It is imperative to use lowcost substrates such as glycerol, glucose, and cellulose for cost-effective biohydrogen production [3]. Identifying and selecting sustainable fuels derived from biomass is the most promising option for sustainable energy generation. Biomass from industries and



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**Copyright:** © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). algae could be a viable source for biohydrogen production, especially in cases where waste treatment and energy production are combined (Table 1).

Table 1. Biohydrogen production using various substrates.

Substrates	H <sub>2</sub> Yield	References
Olive mill wastewater	13.9 L H <sub>2</sub> /L OMW	[4]
Chlamydomonas reinhardtii	311.1 mL $H_2/g$ monosaccharides	[5]
Dairy Wastewater	3.6 L H <sub>2</sub> /L dairy wastewater	[6]
Arthrospira platensis	337.0 mL $H_2/g$ DW	[7]
Microalgae	25.1 mL $H_2/g$ dry biomass powder	[8]
Cheese whey	$2.04 \text{ mol H}_2/\text{mol lactose}$	[9]
Sugarcane bagasse	1753 mL H <sub>2</sub> /L	[10]

Note: DW: Dry weight; MW: Olive mill wastewater.

Carbohydrate-rich, nitrogen-deficient solid waste such as starch residues may be used for hydrogen production using suitable bioprocess technologies. Sugarcane bagasse is the most readily available fibrous material obtained after juice extraction from cane. In most cases, sugarcane bagasse is discarded or fed into mill boilers, neither of which is beneficial to the long-term use of resources and the environment. Sugarcane bagasse contains high cellulose, hemicellulose, and lignin, making it a suitable substrate for producing value-added chemicals and fuels via a biorefinery approach. Because of the complicated structure of cellulose present in bagasse, pretreatment is carried out to improve enzymatic hydrolysis. Thus, substrate pretreatment is essential before it is subjected to hydrolysis or fermentation [3].

Algal biomass can be used as a substrate for biohydrogen production. The utilization of algal biomass has dual benefits as algae sequestered carbon dioxide, which contributes towards the global reduction of harmful greenhouse gases concerning climatic changes, and secondly, it releases oxygen [3]. Substrate pretreatment is one of the significant steps in the successful utilization of biomass for biohydrogen production. Sometimes, the pretreatment of the inoculum also enhances the rate and yield of the biohydrogen. The pretreatment of anaerobic sewage sediment by various methods such as acid, alkali, chloroform, heatshock, ethylenediaminetetraacetic acid (EDTA), ammonia, ozonolysis, and microwave were carried out in batch experiments to test and compare their hydrogen-producing efficiency utilizing various sources of carbon. Among the various pretreatment methods, heat-shock and base treatment showed more H<sub>2</sub> generation at 35  $^{\circ}$ C and 6.5 pH using 5 g/L substrate concentration. The combined action of base and heat pretreatment was found to be the most suitable method for efficient hydrogen production [11]. Several other methods have been reported to enhance hydrogen yield in the recent past through mutation, genetic manipulation, metabolic engineering, and optimization of substrates' pretreatments [3]. In the future, due attention must be paid to identifying optimal conditions and suitable techniques for substrates' pretreatments for cost-effective biohydrogen. The overview of various pretreatment methods employed for efficient biohydrogen production is summarized in (Figure 1). This review focuses on technologies used for pretreatment of substrate and inoculum for biohydrogen production, discussing their advantages and disadvantages.

## 1.1. Algae Biomass-Based Biohydrogen Production

Algae can be either prokaryotic or eukaryotic, performing photosynthesis such as higher plants and contributing to approximately 50% of global carbon fixation depending on the amount of light captured and photosynthesis [8,12]. Algal carbohydrates constitute 15–70% of dry cell mass and have a unique basic composition in every algal strain. As algae are rich in proteins, free amino acids can be obtained by hydrolysis of proteins. The cultivation medium and environment also affect the composition of the algal biomass. The cultivation of *N. oleobundans* in freshwater raises the carbohydrate content of the cell wall, whereas glucose content rises when grown in seawater [13]. *Arthrospira platensis* total carbohydrate content was reported to be 64.3%, mainly in the form of glucan [7]. The

 $\alpha$ -glucan is made up primarily of glucose (>90%), with negligible other simple sugars [14]. To obtain useful raw material to produce biohydrogen, suitable pretreatment must be used to extract both structural and storage microalgal carbohydrates from biomass after harvesting [15].



Figure 1. Schematic representation of various pretreatment methods.

Pretreatment is obligatory for releasing fermentable sugars from algal biomass, as hydrogen-producing bacteria (HPB) typically have less hydrolytic enzymatic activity. To improve the efficiency of the fermentative biohydrogen production from algal biomass, pretreatment is imperative to liberate simple fermentable sugars from complex substrates and make them readily biodegradable. Methods employed for algal biomass pretreatment are physical (mechanical, heat, microwave irradiation, and ultrasonic treatment), chemical (acid, alkali, chloroform, ozonolysis), and biological (enzymatic treatment). These pretreatment methods are supposed to break down the cell wall and facilitate organic substances to be released from the cells [12]. After appropriate pretreatment, the HPB could consume simple substrates to produce hydrogen.

The production of biohydrogen primarily requires algae with high carbohydrate content. To avoid carbohydrate loss during pretreatment processes, the pretreatment method should be chosen very carefully, keeping in mind energy, environmental and economic indicators to achieve maximum efficiency with maximum hydrogen yield. Before selecting a pretreatment method for algal biomass, the following factors should be considered [15]. The efficiency of the pretreatment method is one of the major factors, depending in part on cell disruption efficiency and product yield. Energy demand is also important, as mechanical methods have high energy demands. Hence, to reduce the energy demand and improve performance, a combination of chemical and thermal processes may be used in combination with mechanical methods. Such an approach may reduce environmental impact. Cost is also an important factor. Microwaves or pulsed electric fields have high associated costs, while continuous systems with wet microalgal biomass may reduce costs by eliminating drying and dewatering processes. Product quality is another relevant factor. Although the selected pretreatment method should not ideally influence the quality of the product, it may be compromised in methods involving cavitation or oxidation, where free radicals are generated, or in chemical pretreatment, where excess salts are produced due to neutralization reactions. Another factor is the pollution generated from the solvents used in chemical pretreatment methods [10]. Specific criteria should be followed for selecting the pretreatment technology; the chosen approach should avoid biomass size reduction; the proportion of hemicellulose must be conserved and should be less energy-intensive [16].

#### 1.2. Substrates for Biohydrogen Production and Their Pretreatment

Various pretreatment strategies have been studied to use waste materials for biohydrogen production. Pretreatment of the substrate is essential for efficient bio-hydrogen production. Pretreatment facilitates the substrate breakdown into simple sugars, a prerequisite for microbial growth. Fermentation at high operating temperatures (65–70  $^{\circ}$ C) improves the substrate degradation rate and thus increases  $H_2$  production rate [17]. Various wastes or wastewaters, such as olive mill effluent [4], are ideal for photo-biological  $H_2$  production as they contain ethanol, acetate, butyrate, and propionate. Sewage sludge and tofu wastewater are regarded as the most promising substrates for the production of  $H_2$ . Before being utilized as a substrate, sewage sludge is pretreated for 1 h at 150 °C temperature under 10 atm after alkali treatment. During photo-fermentation,  $0.17-0.28 \text{ L H}_2/\text{L}$  broth/day is constantly produced [18]. Cassava pulp is a by-product of the starch industry that contains up to 60% starch as dry matter and contains carbohydrates, including cellulose and hemicelluloses. Cassava pulp's cellulose and hemicellulose may be acid hydrolyzed and utilized as a substrate for fermentative hydrogen production by anaerobic mixed cultures, under optimum hydrolysis conditions; 0.5% H<sub>2</sub>SO<sub>4</sub> at a ratio of 1:15 (dry w/v) for 30 min, resulting in 27.4 g/L of total sugar yield [14]. Using various organic-rich domestic, agricultural, and industrial waste products for biohydrogen production decreases the cost of scaling up while also effectively removing the organic load [19]. Cheese whey is an excellent substrate for fermentative H<sub>2</sub> generation since it comprises 4.6% lactose, 1.2% crude protein, 0.6% ash, 0.3% fat, 5–8% total solids, and 92.7% water [9,10]. Dairy wastewaters contain polysaccharides, proteins, and lipids, which undergo hydrolysis to sugars, amino acids, and fatty acids [18]. During fermentation, they are transformed into volatile fatty acids (VFAs), which are then degraded by acetogens to produce acetate,  $CO_2$ , and  $H_2$  [20]. Pretreatment of complex biomass containing lignocelluloses is prevalent, as most complex substrates are unsuitable and must be broken down into simple components for easier access during fermentation [21] (Figure 2).

After selecting a microbial consortium enriched in hydrogen-producing ability, the accessibility of easily decomposable substrate is essential for the growth of microbes and hydrogen production. Algal biomass is enriched in large sugar residues and is considered an ideal substrate for  $H_2$  production [22]. Algal biomass (such as other organic waste) can be used for  $H_2$  generation due to its high protein, fat, and carbohydrate content [5,14]. Nguyen et al. (2010) used accumulated starch present within the green algae C. reinhardtii as a substrate. Later, the hyperthermophilic eubacterium T. neapolitana was used to convert that accumulated starch into  $H_2$  gas [5]. The algal starch could be directly fermented into  $H_2$ by a bacterium with amylase activity  $(1.8-2.2 \text{ mol } H_2/\text{mol})$  [22]. These wastes' cellulose and hemicellulose percentages must be hydrolyzed to degrade carbohydrates to achieve high  $H_2$  production. The resultant treated waste may be regarded as the potential substrate for fermentative hydrogen production. The main purpose of substrate pretreatment technology is to break down the complex structure of less degradable organic compounds and improve their solubility, thereby upgrading  $H_2$  yield. The type of substrate and pretreatment method used may affect the yield of hydrogen and the characteristics of the effluents. Ideally, the best substrate pretreatment method should be selected, with high hydrogen yield, cost efficiency, process sustainability, and low energy requirements.

#### 1.3. Inoculum Pretreatment

Inoculum pretreatment technologies are also utilized to enhance the H<sub>2</sub> production yield (Figure 3). These pretreatment technologies help to improve H<sub>2</sub> yield by suppressing the hydrogen-consuming bacteria (HCB) and methane production. Inoculum pretreatment is proven to select micro-flora for better hydrogen production [23] (Table 2). Many studies have reported that pretreatment of sludge to avoid methane production and suppress the action of HCB leads to higher H<sub>2</sub> production [24–27]. Pretreatment of inoculum or substrate may result in enhanced hydrogen content in the range of 43–69% [28]. Pretreatment of anaerobic inoculum accelerates hydrolysis, reducing the effect of the rate-limiting step and improving process stability, thus enhancing anaerobic digestion for effective H<sub>2</sub> production [29–31]. Various inoculum pretreatment methods exist to enhance hydrogen production, including heat pretreatment, acid/alkali-treatment, ultrasonic waves, chloroform, and iodopropane [23,29,30]. Among these methods, the first three are more effective.

Pretreatment steps vary, depending on the inoculum and substrate utilized and their unique properties that affect the H<sub>2</sub> yield.



**Figure 2.** Schematic representation of biomass hydrolysis for biofuel production using physical, chemical, and biological pretreatment methods.



Figure 3. Inoculum pretreatment approaches for biohydrogen production by mixed microbial culture.

Substrate	Pretreatment	Pretreatment Conditions	Hydrogen Yield	Increase in H <sub>2</sub> Yield (%)	References
Rice mill wastewater	Thermal	121 °C temp Time: 15 min	Control: 90 mL of hydrogen Pretreated: 200 mL of hydrogen	55	[32]
Beetroot pulp	Microwave	700 W; 170 °C temp Time: 30 min	Control: 95.7 mL of hydrogen Pretreated: 108.7 mL of hydrogen	12	[33]
Palm oil mill effluent (POME)	Acid	0.8% $(w/v)$ phosphoric acid and $1\%$ $(w/v)$ nitric acid. Time: 10 min	Control: 0.64 mol H <sub>2</sub> /mol Pretreated: 1.24 mol H <sub>2</sub> /mol	48.4	[34]
Bio-methenated distillery wastewater	Ozone	Ozone dose: 4.6 g/h Time: 160 min	Control: 0.37 mL/h of hydrogen yield with 44.19 mL/g COD Pretreated: 1.18 mL/h of H <sub>2</sub> yield with 185.5 mL/g COD	68.6	[35]
Cassava wastewater	Enzymatic	<ul> <li>α-amylase (Incubation temperature: 37 °C)</li> <li>OPTIMASH BG<sup>®</sup></li> <li>(Incubation temperature: 60 °C)</li> <li>Incubation period: 10 day</li> </ul>	Control: 2.06 mol/g COD Pretreated: 5.02 mol/g COD (α-amylase) 4.24 mol/g COD (OPTIMASH BG <sup>®</sup> )	58.9 51.4	[36]
Rice mill wastewater	Combined (acid followed by enzymatic)	Acid -1.5% H <sub>2</sub> SO <sub>4</sub> Reaction time: 60 min Enzyme: <i>A. niger</i> Incubation temperature: 29 °C	Control: 0.128 mol/L of hydrogen Pretreated: 5.34 mol/L of hydrogen.	97.6	[37]
Textile desizing wastewater	Combined pretreatment (flocculation and coagulation)	Coagulant GGEFloc-653 –1 g/L Rapid mixing 100 rpm (3 min), Slow mixing 30 rpm (20 min), Sludge settling time: 1 h	Control: 0.88 L/L-d H <sub>2</sub> production rate Pretreated: 3.8 L/L-d H <sub>2</sub> production rate	76.8	[38]

#### Table 2. List of various substrate pretreatment methods used for biohydrogen production.

## 2. Pretreatment Technologies

Various pretreatment methods for biomass and inoculum pretreatment, and their advantages and disadvantages are listed in Table 3.

## 2.1. Physical

Physical pretreatment methods include thermal treatment, freezing and thawing, microwave radiation, ultrasonication, radiation, and milling.

## 2.1.1. Thermal

Thermal pretreatment facilitates HPB growth, ultimately suppressing methanogenic activity. The formation of spores shows the presence of HPB while non-spore-forming bacteria (methanogens and non-sporulating HPB) are killed during this pretreatment method [29]. Heat-shock is used to assess hydrogen generation by the mixed and anaerobic communities. Compared to untreated culture, thermally shocked or acid-treated culture shows no methanogenic activity [39]. The ability of anaerobic seed sludge to produce hydrogen has been studied using soluble starch. Different heat pretreatment techniques have been reported that result in efficient hydrogen production [40]. The pH is also a critical factor in methanogenic activity during fermentative biohydrogen production [29]. Two essential parameters that should be optimized for effective pretreatment are the pretreatment time and temperature. Most hydrogen consumers are effectively inhibited by short pretreatment time and temperature, as high values of these factors lead to a decrease in hydrogen production [41]. Heat-shock pretreatment of the inoculum for dark

fermentation hydrogen production uses various temperature and pretreatment times and gives impressive results [29].

## 2.1.2. Freezing and Thawing

Physical pretreatment is a novel approach involving freezing and thawing. When microalgal biomass is freeze-dried in a lyophilizer, large ice crystals disrupt the cell and release intracellular components at room temperature [43]. Various freezing and thawing procedures are used. These include: high-pressure freezing, dehydro-freezing, the application of anti-freeze protein, ice nucleation (in freezing), high-pressure microwave thawing, and ohmic thawing (in thawing). Freezing applies two processes: change in phase from liquid to solid and lowering the temperature. When water freezes at atmospheric pressure, the increased volume leads to the formation of ice that has a lower density than liquid water [44]. The freezing procedure is primarily utilized for large-scale food preservation where crystal formation is necessary, and a significant heat gradient exists [45,46].

Thawing is usually slower than freezing. In the food industry, thawing is a new application. According to some studies, it can preserve food quality and reduce necessary thawing time [47,48]. Thawing has various effects that have been investigated for a long time, including protein denaturation, gelatinization of starch, and inactivation of enzymes. It can be used to sterilize microorganisms and preserve food without heating [47]. Thawing helps in releasing soluble intracellular components from the substrate at 25 °C in 12 h [49].

## 2.1.3. Microwave Radiation

Microwave radiation is generally used as a pretreatment method for feedstock with lignocellulosic components. This pretreatment method may be preferred due to factors such as the ease with which it degrades the cellulose fraction structure, ease of operation, minimal inhibitor formation, high heating capacity in a short time, and low energy consumption (Table 3). Switchgrass hydrolyzed with cellulase enzyme, soaked in water, and microwave irradiated had a total sugar production of 34.5 g/100 g biomass [50,51]. Microwave irradiation is an economical and cost-effective way of removing microorganisms and does not cause any harm to the environment. A previous study suggested that microwave heating (at 850 W for 3 min) increased soluble chemical oxygen demand COD in sludge [52].

#### 2.1.4. Ultrasonication

Ultrasound irradiation reduces treatment time and chemical or enzyme requirements. Overall, this approach is regarded as a unique and eco-friendly technique that provides a high degree of intensification [53]. It works on the activated sludge's physical, chemical, and biological properties (Table 3). Ultrasound irradiation is the most favorable inoculum pretreatment method for the utilization of activated sludge as an inoculum in hydrogen production by dark fermentation. The effectiveness of ultrasonic irradiation pretreatment for improving hydrogen production before anaerobic fermentation from complex food waste has been investigated [54,55]. Hydrogen production was improved by an increase in total solids (TS) concentration and ultrasound time. The significant enhancement of hydrogen yield—up to 75% and a rate of 104%—implies that ultrasound irradiation could be used for enhanced biohydrogen production from food waste [54,55].

# 2.1.5. Radiation

The radioisotope Cobalt-60 emits gamma radiation, which has been used in substrate treatment for biohydrogen production. Gamma radiation easily penetrates lignocellulose, damaging lignin and cellulose structure through the formation of free radicals, which cause deterioration of the biomass. This approach was assessed and compared for long-term viability in enriching hydrogen-producing bacteria in dark fermentation with glucose as a substrate in batch experiments [56]. The results showed that among different pretreatment methods, seed sludge pretreated with ionizing radiation produced the maximum hydrogen

rate and yield, and a substrate degradation rate of 525.6 mL, 37.2 mL/h, 267.7 mL/g glucose (2.15 mol/mL glucose), and 98.9%, respectively [56]. The fermentation process was studied under ideal conditions using gamma-irradiated sludge as inoculum. Another approach that could be used as inoculum pretreatment for dark fermentation is infrared radiation. For the first time, the efficient conversion of wheat-straw waste into biohydrogen gas was reported using cow-dung compost [57]. The maximum cumulative hydrogen yield of 68.1 mL H<sub>2</sub>/g total volatile solids was found after 12.65 h of fermentation utilizing raw wheat straw and HCl.

**Mechanical Methods** Pretreatment **Cell Disruption Method** Advantages Disadvantages High energy required. Applicable for wet microalgal Heat generated may cause pastes and components that Cell wall grinding with thermal degeneration. Bead milling can be easily separated. spinning solid beads. High installation cost. Reduced cellulose Beads should be crystallinity. replaced periodically. High disruption efficiency. High energy required. Cavitation effect and acoustic Brief pretreatment time. Cooling required for the excess heat generated. streaming from ultrasonic Components such as Ultrasonication Oxidative free radicals formed waves create high pressure solvent/beads are and heat for disrupting cell not required. cause damage. Minimum maintenance costs. Not applicable to all algae. Component structure may Interaction between High disruption efficiency. change due to changes in high-frequency shock waves Short pretreatment time. non-covalent interactions. Microwaves and charged molecules creates Low energy demand. High maintenance cost. high-pressure and heat to Easy to scale up. Cooling required for excess disrupt cells. heat generated. Thermal Methods Biomass is exposed to steam High temperature at high T and P followed by High cell disruption efficiency. (180–240 °C) and pressure Steam explosion depressurization that causes Low maintenance. (1.03-3.45 MPa). an explosion to rupture the Resistant to corrosion. Disruption efficiency varies cell membrane. with species. Mild operating conditions, Biomass freezing to form large ice crystals that expand and suitable for extracting Time-consuming. disrupt the cell wall. Excess Freezing and thawing sensitive components. Energy and cost-intensive. disruption with repeated Biomass blending is not needed. freeze-thaw cycles. Chemical methods Acid- hydrolysis of cell Low energy demand Highly prone to corrosion. Low temperature. wall polymers. Formation of fermentation Acid/alkali treatment Short reaction time. Alkali- saponification of cell inhibitors (furfurals). wall lipids for cell disruption. Easy to scale up. Protein denaturation **Biological methods** Hydrolysis of the cell wall by Mild reaction conditions. Enzyme is costly. enzymes from bacteria and High product quality Enzyme specificity requires a Enzymatic hydrolysis/ fungi destroys cell wall and purity. cocktail of enzymes. Fungal Treatment polymers to access Low energy demand. Long reaction time intracellular glucans. No chemicals required. (2-4 weeks).

Table 3. Pretreatment methods and their advantages and disadvantages [10,42].

2.1.6. Milling

Milling is the initial step of the pretreatment process and is done to increase the surface area of the substrate molecules. Milling techniques utilized in biofuel production include two-roll milling, bead milling, hammer milling, disc milling, and colloid milling [58]. After chipping, milling, and grinding, the final particle size of the feedstock may be reduced to a few mm (0.2–10 mm) (Table 3). A disadvantage of dry milling is the consumption of a high amount of energy that could bead dressed by wet milling [59]. Because of its low energy usage, wet disc milling has also become a common pretreatment method. Compared to hammer milling, which creates finer bundles, disc milling improves cellulose hydrolysis by producing more fibers [60,61]. Compared to ball milling, attrition and planetary mills are more effective at reducing biomass size. Other milling methods produce the lowest quantity of glucose and galactose, whereas planetary milling produces the most. It should be emphasized that none of the mill pretreatment procedures yield hazardous chemicals such as hydroxyl methyl furfuraldehyde (HMF) or linolenic acid (LA).

Milling is used before enzymatic hydrolysis and other pretreatment procedures such as chemical or physicochemical approaches. Methods of pretreatment, such as chemical or mechanical milling, are classified as either dry or wet, and the type of biomass used determines which method should be employed. Dry biomass such as napier grass and maize stover is often processed using extruders, roller mills, cryogenic mills, and hammer mills. Colloid milling, fibrillation, and dissolving are also appropriate for pretreatment of wet biomass such as energy cane, wheat bran, or wheat straw [62].

#### 2.2. Chemical Pretreatment Technologies

Various chemical pretreatment technologies have been reported to improve biohydrogen production for either inoculum or substrate. Chemical pretreatment agents include acids, alkalis, methanogen inhibitors such as 2-bromoethane sulfonate (BES) or 2-bromoethane sulfonic acid (BESA) ozonolysis, chloroform, iodopropane, and acetylene.

#### 2.2.1. Acid and Alkali Pretreatment

Acid and alkali pretreatments are the most widely used methods for enriching or suppressing HCB, considering the susceptibility of non-sporulating hydrogen consumers to pH changes [63]. The most commonly used acids are HCl,  $H_2SO_4$ , and HNO<sub>3</sub>, having a concentration ranging from 0.1 to 6.0 M [30,64–66] and pH stretching from 2.0 to 4.0 [39,64,67]. In the case of alkali pretreatment, NaOH, KOH, and  $Ca(OH)_2$  are widely used, with concentrations ranging from 1.0 to 8.0 M and pH-value from 10 to 12 [43,65,66,68]. Acid and alkali pretreatments are most commonly used in selective hydrogen-producing cultures for dark fermentation [43]. Acid pretreatment may improve the solubilization of hemicellulose but is ineffective in delignification [69,70]. Alkaline solutions saponify the ester bonds in lignocellulosic substrates, disintegrating their crystalline structure and subsequently enhancing hydrolysis. Dilute alkali pretreatment is suitable for the subsequent hydrolysis by enzymes and enhances residual polysaccharide reactions. Pretreatment with  $Ca(OH)_2$ increased the crystallinity index by removing amorphous substances [71]. Another study revealed that  $Ca(OH)_2$  can remove the group of acetyls from hemicellulose to increase the digestibility of cellulose and reduce enzyme steric obstruction [72]. Pretreatment using CaO<sub>2</sub>/CaO enhanced both biohydrogen production and biodegradation of wheat straw. At 6% CaO and 121 °C, a maximum H<sub>2</sub> yield of 114 mL per g of total solids (TS) was attained. Moreover, CaO<sub>2</sub> pretreatment under identical conditions gave a maximum H<sub>2</sub> yield of (71.8 mL/g TS), higher than that of the control group (43.2 mL/g TS) treated with hot water at 12 °C [72]. Considering the pretreatment costs and the capability of  $H_2$  production, CaO proved to be a better pretreatment agent than  $CaO_2$  [73]. Among acids, HCl and H<sub>2</sub>SO<sub>4</sub> are the most widely employed, while NaOH is a commonly used alkali for pretreatment [74]. Pretreating complex substrates using acids is an economical and efficient method, but it has the drawback of corroding the reactors. Another drawback of both pretreatments (acid/alkali) is the probability of the formation of inhibitory compounds [63]. Further, adjusting pH after the pretreatment complicates the process and increases operational costs (Table 3).

## 2.2.2. BES and BESA

Using chemicals such as 2-bromoethane sulfonate (BES) or 2-bromoethane sulfonic acid (BESA), hydrogen-consuming methanogens can be removed selectively and efficiently. BESA is a coenzyme-M analog found mainly in methanogens and is reported to inhibit the process of acetate production by methanogens [75]. BESA (25–100 mmol) also efficiently destroys methanogenic bacteria present in sludge [76,77]. In mixed microalgal biomass, BESA addition at pH 5.5 results in maximum hydrogen production rate (210 mL/L/day) and yield (29.5 mL/g), nearly 3-fold higher than the control [78].

#### 2.2.3. Ozonolysis

Ozonolysis is an efficient oxidative pretreatment method to treat lignocellulosic substrates such as wheat straw, bagasse, and green hay. Ozonolysis reduces lignin in the lignocellulosic biomass, thereby promoting enzymatic hydrolysis to increase the levels of fermentable sugars [79]. Ozonolysis pretreatment is carried out at optimum temperature and pressure. Furthermore, it does not give rise to any noxious inhibitors, making the process sustainable [80]. The main aspect that influences pretreatment with ozone is the biomass moisture content; the higher the moisture content in biomass, the lower the lignin oxidation [63]. Wheat straw ozonolysis produced 4.14 g/100 g dry matter of oxalic acid and 0.98 g/100 g dry matter of acetic acid at 40% moisture, 60 L/h air/ozone flow rate, 2.7% ozone concentration, and 2 h pretreatment time [81]. Several process parameters such as ozone intake, pretreated biomass composition, sugar yield, and generation of inhibitory compounds capable of inhibiting subsequent phases of enzymatic hydrolysis and hydrolysate fermentation significantly affect process efficiency. Therefore, process parameters must be optimized to attain a viable commercial alternative [82]. Ozonolysis pretreatment was found to be successful due to lignin polymer degradation and minimal effects on lignocellulosic biomass cellulose and hemicellulose content (Figure 4).

#### 2.2.4. Chloroform

Chloroform (CHCl<sub>3</sub>) has been used to limit methanogenic activities and therefore enhance biohydrogen yield during dark fermentation [30,43]. Chloroform is used in a similar way to BES or BESA in principle. It inhibits the conversion of methyl-CoM to methane by inhibiting methyl-com reductase (MCR), thereby restricting corrinoid enzymes function. Considering the inhibition of hydrogenotrophic methanogenesis, hydrogen generated during dark fermentation is not consumed, resulting in increased hydrogen yield [83]. Various authors have studied the use of chloroform to suppress HCB in mixed microflora. Chloroform concentration has ranged between 0.0005% and 5%, with time intervals between 17 h and 24 h at ambient temperature [30,43,84–86]. Furthermore, several studies also revealed that the use of chloroform above certain concentrations leads to inhibition of HPB due to adverse effects of chemicals [30,87,88]. Luo et al. found that pretreatment with chloroform does not significantly affect hydrogen yield [89].

## 2.2.5. Iodopropane

Iodopropane is an inhibitor used in dark fermentation (DF) to suppress methanogenic activity [29,68]. The effects of iodopropane and chloroform on methanogens are more or less similar. As a corrinoid antagonist, iodopropane denatures B12 enzyme activity and functions as a methanogen methyl carrier [90]. As a result, the formation of methyl-CoM is inhibited, suppressing methanogenesis and methane formation [87]. The inhibition of hydrogenotrophic methanogenesis prevents the consumption of produced hydrogen, thus resulting in a higher hydrogen yield [83]. Microflora obtained from sludge pretreated with iodopropane was reported to inhibit methanogenic activity, whereas no impact on hydrogen production was found. Iodopropane has also been able to suppress non-methanogens [29].





Particle

Figure 4. Schematic representation of microalgal cell disruption through ozonolysis.

## 2.2.6. Acetylene

Polymer

Acetylene is regarded as a non-specific methanogenesis inhibitor. Valdez-Vazquez et al. (2006) reported that inoculum pretreatment with acetylene resulted in higher hydrogen yield [91]. As methanogenic organisms lose their ability to sustain a transmembrane pH gradient, acetylene exposure reduces methanogenic processes such as methanogenesis and ATP generation [63]. There are several advantages of using acetylene in pretreatment technology for enhanced biohydrogen generation since it is a cheap molecule for inoculum pretreatment and has positive methanogenic inhibitory effects [60]. Methanogenic inhibition was also reported when the inoculum was treated with 1% acetylene (v/v) using milled paper as a substrate [92].

## 2.3. Biological Technologies

Biological pretreatment technology is more effective and environment-friendly than conventional physical and chemical pretreatment methods due to its lower energy consumption [63]. Plenty of cellulolytic and hemicellulolytic microorganisms exist in nature, and these are targeted for effective biomass pretreatment [93]. Biological pretreatment technology breaks cross-linked structures of lignocellulose residues by increasing the rate of hydrolysis. Numerous biological pretreatment methods can be considered as natural methods of pretreatment, including enzymatic hydrolysis and fungal treatment. Different strains of fungus are used, such as white, brown, and soft-rot fungi that break down hemicellulose, lignin, and also a small amount of cellulose [94]. Lignin degradation with the help of white-rot fungi is mainly attributed to lignin-degrading enzymes, such as peroxidases and laccases [95]. Enzymatic hydrolysis is an alternative for fungal pretreatment, which directly utilizes enzymes to carry out hydrolysis. Poplar leaves containing lignocellulose are considered a substrate to produce biohydrogen [96]. Batch experiments have been

of enzymes (VicozymeL-2%) was equal to 44.92 mL/g of dry leaves, which is far better (1.34-fold) than the yield obtained by acid/alkali pretreatment. According to [36], with the help of batch production, biohydrogen was obtained from cassava wastewater using seed sludge as a substrate after different pretreatment conditions. Wastewater treated with sonication for 45 min at pH 7 yields 0.913 mol H<sub>2</sub>/g of COD. OPTIMASH BG, 0.2% enzyme pretreatment at pH 7 yields 4.24 mol H<sub>2</sub>/g of COD. Alpha-amylase (0.2%) pretreatment at pH 7 yields 5.02 mol H<sub>2</sub>/g COD. The major drawback of biological treatment technology that limits its use industrially is the sluggish hydrolysis rate of lignocellulose [82].

## 2.3.1. Hybrid Technologies

Numerous experiments have been conducted to enhance the yield of hydrogen using a combination of two or more pretreatment techniques. The methods mentioned above are combined for substrate pretreatment, such as physical and chemical methods [97].

# 2.3.2. Combined Technologies for Substrates and Inoculum

Combined technologies widely used for the pretreatment of substrates involve heat with acid and alkali [33,98,99]. Pretreatment of activated sludge with 0.5 percent w/v HCl acid followed by 15–60 min of heating at 110 °C showed the maximum hydrogen yield (155.6% in 45 min) compared to control [99]. Furthermore, combining alkali, acid, and heat pretreatment with sonication gave 180.4% hydrogen yield compared to untreated food waste. In addition, the combination of ozone and sonication for pretreating activated sludge led to an increase of H<sub>2</sub> by 686.4% compared to the untreated sludge [100,101]. Despite this, BESA alone released higher hydrogen than the acid and BESA combined [23].

## 2.4. Other Advanced Pretreatment Methods

According to Zabed et al., 2019, several factors need to be considered for a sustainable pretreatment method [102]. These include cellulose content modification for better downstream enzyme hydrolysis, decreasing cellulose and hemicellulose content loss, possible inadequate formation of unwanted inhibitors, reducing energy consumption, and reducing by-product formation and the exhaustion of chemicals. Present pretreatment approaches are incapable of fulfilling the industrial suitability criterion. As a result, developments of various other advanced approaches such as hydrothermal carbonization (HTC), super-critical fluid, ammonia fiber explosion (AFEX), ionic fluids (IFs), low-temperature steep delignification (LTSD), or co-solvent enhanced lignocellulosic fractionation (CELF), have been used for the industrial utilization of biomass.

## 2.4.1. Hydrothermal Carbonization (HTC)

Hydrothermal carbonization (HTC) is an eco-friendly approach for transforming biomass containing high lignocellulose into various value-added products [103,104]. Temperature is one of the main variables in hydrothermal carbonization, which significantly influences biomass conversion by structural component degradation. The combined action of hydrothermal reactions disintegrates the chemical bonds of the structural components [105]. This pretreatment is established at a high temperature of subcritical water (less than 374 °C). Each component is individually affected by degradation at different temperature ranges. Microalgae biomass is also treated with HTC in the following ways: directly on microalgal biomass, pre-digested microalgal biomass effluents by recycling, and lipid-extracted microalgal cells [106,107]. Natural microalgae are the most widely studied raw material for their HT conversion into products such as fuel, organic chemicals, and hydrochar [107]. HTC is an energy-saving process that is more sustainable at low temperatures (below 300 °C). For heating at high temperatures, processes such as pyrolysis or microwave heating are preferred [108]. Hydrothermal carbonization was found suitable

for converting water hyacinth to hydrochar [109]. Hydrochar, HTC's solid, hydrophobic and nontoxic product, is employed as adsorbent/activated carbon in treating wastewater [110–112]. Direct conversion of biomass into a carbon product and the formation of gas and liquid phases is an example of recent advancements in HTC. Furthermore, many functional groups that involve oxygen are retained over the hydrochar surface [113,114]. The content of cellulose, hemicellulose, and lignin affects the amount and yield of hydrochar [112]. An autoclave reactor at high pressure in the range of 0.6 to 1.5 bar, with temperature varying from 160 to 240 °C, was used for performing the HTC experiment. The residence time in the experimental process was in the range of 30–120 min [113].

## 2.4.2. Supercritical Fluid

When lignocellulosic material is pretreated in a supercritical manner that involves high temperature and pressure conditions, we obtain biomass with an increased surface area at enhanced mass transfer rates along with lignin getting separated from the raw materials (e.g., wood chips, rice straw, switchgrass, etc.) [114,115]. SC-CO<sub>2</sub> is most widely used for extracting lignin and lipids as it is relatively nontoxic, readily available, cheap, and has lower critical values of temperature and pressure [116]. The diffusion of  $CO_2$  into biomass results in the formation of carbonic acid that hydrolyzes hemicellulose. In the next step, the sudden release of pressure disrupts biomass structure resulting in increased availability of cellulose [117]. SC-CO<sub>2</sub> is operable at low temperatures and high pressure compared to thermal treatments that reduce inhibitors formed by lignin and hemicellulose degradation. Owing to its high diffusivity and low viscosity, SC-CO<sub>2</sub> is more efficient for wet biomass, and it can also be combined with other pretreatments for further improvement in cellulose digestibility. In the case of sugarcane bagasse pretreated with  $SC-CO_2$ , the glucose yield of enzymatic hydrolysis could be up to 72%, whereas 20% higher glucose yield could be achieved by combining SC-CO<sub>2</sub> with alkali pretreatment as compared to alkali pretreatment alone [118].

## 2.4.3. Ammonia Fiber Explosion and Ammonia Pretreatment

Ammonia fiber explosion (AFEX) involves physical effects such as high temperature and pressure, and chemical effects such as  $NH_{3}$ , to reduce the biomass's unresponsiveness towards hydrolytic enzymes. In the AFEX method, the biomass is reacted with liquid anhydrous NH<sub>3</sub> for 5 to 60 min in a temperature ranging from 60 to 100  $^{\circ}$ C and in pressure ranging from 1.7 to 2.1 MPa with water-load variations [119]. AFEX generally operates on dry biomass (w/w) with a dosage ratio of (1:1/2:1); this aqueous biomass is then given heat treatment in a temperature range from 60 to 120 °C and pressure of 3 MPa for a duration of 30 min [102]. At present, AFEX has less application in microalgae biomass hydrolysis, but recent research has shown that free ammonia in pretreating algal biomass remarkably enhances the production of  $CH_4$  and  $H_2$  under anaerobic conditions [120,121]. In the latest study, co-digestion of sugarcane bagasse (SCB) and cow manure by AFEX enhanced the  $N_2$  content of the biomass and enhanced the biodegradation capability SCB. As a result, the methane yield increased to 292–299 L/kg of volatile solids, and the methane content of biogas increased to 57–59%, v/v [122]. Since AFEX-treated biomass is a dry-to-dry process, it produces negligible significant inhibitors during pretreatment. Thus, water-washing or further detoxification is not required for biomass treated with AFEX, as it can be easily hydrolyzed and fermented [123].

## 2.4.4. Ionic Liquid-Based Pretreatment

Ionic Liquids (ILs) are solvents with high thermal stability and they are environmentally safe, relatively nontoxic, and less volatile. ILs notably remove hemicellulose and lignin from lignocellulosic biomass while releasing cellulose [124–126]. Acidic ILs have been used to solubilize lignocellulosic biomass partially or entirely. After the dissolution of wood (either partially or completely), the dissolved fractions can be distinguished from the undissolved ones using four techniques. When there is no requirement for biomass fractionation, the first strategy is usually applied. After the biomass' partial/complete dissolution, an anti-solvent is employed in the regeneration of the reaction mixture entirely without fractionation. The second technique involves the complete dissolution of lignocellulosic biomass and selective fractionation into polysaccharides and lignin. The third technique is used to separate dissolved and undissolved biomass by filtration/centrifugation before regeneration. The fourth technique fractionates lignin, a polysaccharide-rich part, and undissolved biomass into three parts [127]. The lignin and hemicellulose can be dissolved partially with ILs. ILs based on imidazolium pretreatment eliminated lignin in the range of 15–92% from softwood biomass (e.g., pine, switchgrass, willow, etc.) and agricultural residue [128]. Pretreatment of biomass with ionic liquids and hydrolysis with enzymes gave 90% glucose and 25% hemicellulose. Acid/alkali pretreatment methods produce less reducing sugar than ILs for rice and barley straw. To reduce the cost of the consolidated process, IL-tolerant enzymes have been created to saccharify lignocellulosic biomass. For using biomass pretreatment industrially, factors such as recovery, cost, reusing ionic liquids, etc., become significant [129].

## 2.4.5. Low-Temperature Steep Delignification (LTSD)

In low-temperature steep delignification (LTSD), oxygen, nontoxic chemicals, and bases are used in small concentrations for lignocellulosic biomass conversion. This process uses the lowest possible concentrations of chemicals that are nontoxic under mild operating conditions, resulting in high yield and a high rate of conversion. The chemicals and enzymes used may also be recovered. Mild operating conditions segregate the biomass into lignin, hemicellulose, and cellulose. In this process, lignocellulosic biomass is separated into three fractions: lignin, cellulose, and hemicellulose, with lignin being extracted 90% from the biomass. The steps involved in the process include enzymatic hydrolysis, fermentation, and distillation. In one study, this method was used on hardwood chip conversion, where glucan was obtained with a yield of 78% after the enzymatic hydrolysis. The chemicals used were sodium chlorite,  $H_2O_2$ , and  $O_2$  [130].

# 2.4.6. Co-Solvent Enhanced Lignocellulosic Fractionation (CELF)

In this pretreatment method, acetylated lignin is dissolved from lignocellulosic biomass with enhanced hydrolysis of cellulose in water by combining organic solvents, water (H<sub>2</sub>O), dilute sulphuric acid (H<sub>2</sub>SO<sub>4</sub>), and tetrahydrofuran to achieve a high yield of reducing sugar with minimal use of enzyme [131]. Additionally, CELF uses co-solvents such as 1,4-dioxane, acetone, and organic solvents such as methanol and ethanol for effective lignin fractionation from biomass. The process operates at high temperatures (160–180 °C) and generates several by-products converted to other valuable products [132].

## 3. New Approaches to Biomass-Based Biohydrogen Production

Three processes have been studied based on the pretreatment method, hydrolysis, and fermentation process being carried out in the same/separate reactor.

#### 3.1. Separate Hydrolysis and Fermentation (SHF)

Separate Hydrolysis and Fermentation (SHF) is an effective approach that involves sequential enzymatic hydrolysis and fermentation (Figure 5). However, to obtain a better biohydrogen yield, the conditions for hydrolysis should be optimal [133]. The sugars hexose and pentose are found in lignocellulosic biomass hydrolysate; to further enhance H<sub>2</sub> yield, microbes are essential to ferment both sugars. The bacterium *Thermoanaerobacterium thermosaccharolyticum* W16 uses fermentable sugars obtained from hydrolysis to produce biohydrogen. Biomass pretreatment procedures produce inhibitors, which should be removed during the detoxification step. Biological pretreatment and saccharification methods have been reported to avoid the production of inhibitors. In the case of enzymatic saccharification, *Trichoderma viride* crude cellulase was utilized to produce reducing sugar [134]. When steam-exploded switchgrass was used for H<sub>2</sub> production, the yield

obtained by using the hydrolysate was 99.8 mL/g of substrate [135]. Pretreatment methods involving acid and alkali have a significant influence on grass saccharification for producing H<sub>2</sub>. In the case of grass with acid pretreatment, the H<sub>2</sub> produced was 72.2 mL/g of the substrate.



**Figure 5.** Schematic representation of separate hydrolysis and fermentation (SHF) for biohydrogen production.

#### 3.2. Simultaneous Saccharification and Fermentation (SSF)

Simultaneous Saccharification and Fermentation (SSF) is carried out in the same reactor by utilizing different enzymes for saccharification and  $H_2$  fermentation (Figure 6). SSF can improve process time and cost while also removing product inhibition during cellulose hydrolysis [136]. Using SSF, Clostridium butyric ASI 209 was selected to produce  $H_2$  from steam-exploded corn, and the effect of variables such as pH or enzyme-loading on the productivity of  $H_2$  was noted. Some other studies noted how acetic acid and enzymeloading affected  $H_2$  production. When the acetic-acid steam-exploded procedure was used to increase the production of  $H_2$ , *Ethanoligenes harbinese* B49 resulted in high  $H_2$  yield and production [133]. SSF produced biohydrogen from fermentable sugars derived from fungal-pretreated cornstalk using enzymes from *T. viride* and *T. thermosaccharolyticum* W16. After studying various process parameters, a maximum yield of 89.3 mL  $H_2$ /gm substrate was reported [134].



**Figure 6.** Schematic process of simultaneous saccharification and fermentation (SSF) for biohydrogen production.

#### 3.3. Consolidated Bioprocessing (CBP)

CBP is an integrated and economically feasible single-step process (Figure 7). Compared with fermentation, the pH and temperature required here for enzymatic hydrolysis are different. Compared to SHF and SSF technologies, the CBP operating technique is more straightforward. No enzyme is added, the chances of contamination are less, the energy requirement is low, and minimal investment is required. CBP encompasses a mixed-culture activity producing a synergistic effect, thus giving a high H<sub>2</sub> yield. Reduced time for fermentation, high substrate conversion, and eliminated inhibitory compounds are some of the benefits of using mixed culture in this process [137].



Enzyme production, hydrolysis and fermentation

Figure 7. Process outlook of consolidated bioprocessing (CBP) for hydrogen production.

## 3.4. Integrated Strategy for Alternative Bioenergy Resources in Addition to $H_2$

Along with hydrogen, several byproducts such as lactic acid, ethanol, and lactic acid are also produced during fermentation. Recovery of these products is desirable to make the process more feasible and economical.

## 3.4.1. Co-Production of Bio Alcohols and VFAs

Complex lignocellulosic biomass is prefered to be utilized as a substrate to minimize the impact on the environment. In aerobic bacteria, extracellular enzymes break down cellulose, while in anaerobic bacteria, hydrolysis of cellulose is mediated by the cellulosome, a multi-enzyme extracellular complex. Alcohols, VFAs, and  $H_2$ , can be produced simultaneously using mixed culture [138]. Cellulolytic *E. coli* have been utilized on bovine rumen to produce 0.36 g/L of ethanol and an  $H_2$  yield of 4.41 mL/gm from corn straw. Sweet sorghum stalk, plant-based biomass, contains structural carbohydrates, e.g., cellulose, hemicellulose, and nonstructural sugars, e.g., fructose, glucose, sucrose, that affect the yield of  $H_2$ . A two-step dark fermentation method has been used with *Clostridium thermosaccharolyticum* to utilize carbohydrates (structural and nonstructural) of sweet sorghum stalk to produce VFAs and hydrogen. In the first step, nonstructural sugars were used by *C. thermosaccharolyticum* to give 3.27 mmol  $H_2/g$  substrate, 1.18 g/L of acetic acid, and 0.97 g/L of butyric acid. The remaining biomass, when treated with dilute  $H_2SO_4$  before being refermented with *C. thermosaccharolyticum*, produced 2.5 mmol  $H_2/gm$  of the substrate, 0.99 g/L acetic acid, and 1.09 g/L butyric acid [139].

### 3.4.2. Co-Production of Methane and Hydrogen

The energy density of hydrogen is high (142 kJ/g), followed by methane having 55.5 kJ/g [140]. Acidogenic microorganisms utilize lignocellulosic biomass to produce CO<sub>2</sub>, H<sub>2</sub>, and VFAs after hydrolyzing them into free sugars. Hydrogenotrophic methanogens use CO<sub>2</sub> and H<sub>2</sub> dissolved in the medium to generate CH<sub>4</sub>, whereas acetogenic bacteria do conversion of acetic acid into methane and CO<sub>2</sub>. H<sub>2</sub> is scavenged by hydrogenotrophic methanogens, which reduces H<sub>2</sub> production. Dark fermentation is preferred over anaerobic fermentation to prevent the formation of methane and methanogens, resulting in increased H<sub>2</sub> and CO<sub>2</sub> production. Lignin amount was decreased on pretreatment with alkaline H<sub>2</sub>O<sub>2</sub>, accompanied by an increase in substrate conversion efficiency to around 86 percent, resulting in H<sub>2</sub> yields of 303 mL/g COD. In recirculated-two-phase anaerobic digestion (R-TPAD), both dark fermentation and methane-producing processes are combined. Hydrogen is produced in the first stage, and the used material of 1st stage that contains byproducts

is then subjected to methane production in the second stage. After this, the effluent from methanogenesis is recirculated to the dark fermentation unit [141].

## 4. Conclusions

Biohydrogen production is eco-friendly compared to other methods of hydrogen production. It has the potential to replace conventional fossil fuels without releasing greenhouse gases. Hydrogen production from biomass depends upon the type of feedstock utilized. The substrate pretreatment is an important step that affects biohydrogen production. The utilization of substrates such as food, agriculture, and industrial waste after pretreatment could enhance hydrogen yield. So far, no single pretreatment method hydrolyzes biomass efficiently in an eco-friendly way. Many reports suggest that the combination of thermal and enzymatic pretreatments could improve substrate hydrolysis and thus enhance biohydrogen production. However, biological methods may be more eco-friendly and cost-effective compared to other physical and chemical methods. More extensive research is needed to exploit the potential of combined approaches of pretreatment. The pretreatment method employed should be cost-effective, should not be energy-intensive, and should be quicker to hydrolyze substrates. Pretreatment methods such as AFEX and steam explosion require specialized and costly equipment in comparison to simple methods such as milling. The integration of effective and cheap pretreatment methods will improve the hydrolysis of the substrate and thus biohydrogen production, with better hydrogen yields. Most of the studies being carried out on the pretreatment process are at the laboratory scale. Sound technical knowledge and technologies are needed to shift substrate pretreatment processing from the laboratory to commercial scale.

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