# Mathematical Modeling and Stability Analysis of a Two-Phase Biosystem

#### Authors:

Milen Borisov, Neli Dimitrova, Ivan Simeonov

Date Submitted: 2020-10-26

Keywords: numerical simulation, stability analysis, equilibrium points, mathematical model, Hydrogen, methane, anaerobic digestion, two-phases process

#### Abstract:

We propose a new mathematical model describing a biotechnological process of simultaneous production of hydrogen and methane by anaerobic digestion. The process is carried out in two connected continuously stirred bioreactors. The proposed model is developed by adapting and reducing the well known Anaerobic Digester Model No 1 (ADM1). Mathematical analysis of the model is carried out, involving existence and uniqueness of positive and uniformly bounded solutions, computation of equilibrium points, investigation of their local stability with respect to practically important input parameters. Existence of maxima of the input?output static characteristics with respect to hydrogen and methane is established. Numerical simulations using a specially elaborated web-based software environment are presented to demonstrate the dynamic behavior of the model solutions.

Record Type: Published Article

Submitted To: LAPSE (Living Archive for Process Systems Engineering)

Citation (overall record, always the latest version):LAPSE:2020.1060Citation (this specific file, latest version):LAPSE:2020.1060-1Citation (this specific file, this version):LAPSE:2020.1060-1v1

DOI of Published Version: https://doi.org/10.3390/pr8070791

License: Creative Commons Attribution 4.0 International (CC BY 4.0)





Article

# Mathematical Modeling and Stability Analysis of a Two-Phase Biosystem

Milen Borisov <sup>1</sup>, Neli Dimitrova <sup>1,\*</sup> and Ivan Simeonov <sup>2</sup>

- Institute of Mathematics and Informatics, Bulgarian Academy of Sciences, Acad. G. Bonchev Str., Block 8, 1113 Sofia, Bulgaria; milen\_kb@math.bas.bg
- Institute of Microbiology, Bulgarian Academy of Sciences, Acad. G. Bonchev Str., Block 26, 1113 Sofia, Bulgaria; issim@abv.bg
- Correspondence: nelid@math.bas.bg

Received: 24 April 2020; Accepted: 2 July 2020; Published: 6 July 2020



**Abstract:** We propose a new mathematical model describing a biotechnological process of simultaneous production of hydrogen and methane by anaerobic digestion. The process is carried out in two connected continuously stirred bioreactors. The proposed model is developed by adapting and reducing the well known Anaerobic Digester Model No 1 (ADM1). Mathematical analysis of the model is carried out, involving existence and uniqueness of positive and uniformly bounded solutions, computation of equilibrium points, investigation of their local stability with respect to practically important input parameters. Existence of maxima of the input–output static characteristics with respect to hydrogen and methane is established. Numerical simulations using a specially elaborated web-based software environment are presented to demonstrate the dynamic behavior of the model solutions.

**Keywords:** anaerobic digestion; two-phases process; hydrogen; methane; mathematical model; equilibrium points; stability analysis; numerical simulation

#### 1. Introduction

Anaerobic digestion (AD) is a multi-step biotechnological process widely applied for the treatment of wastes and wastewater, where usually hydrogen ( $H_2$ ) is not accumulated as an intermediary product [1–3]. Recently, the interest in  $H_2$  production by modifying operating conditions of the AD has increased. This process has been denominated as dark fermentative  $H_2$  production in contrast with light derived  $H_2$  producing processes [4–8]. In traditional AD,  $H_2$  is not detected as it is consumed immediately e.g., by hydrogenotrophic methanogens to produce methane  $CH_4$  and carbon dioxide  $CO_2$  [9]. Conversely,  $H_2$  can be produced as main component in biogas by engineering the process conditions in a way that favors Clostridium and Enterobacter metabolism against the predominance of archaeal microflora [10,11]. However, the main limitation of dark fermentative  $H_2$  production is the rather low energy recovery, cf. [7,8,12]. In order to completely utilize the organic acids produced during dark fermentation and improve the overall energy conversion efficiency, a two-stage AD (TSAD) concept consisting of a hydrogenic process followed by a methanogenic process has been suggested (see e.g., [12–14]).

A lot of models describing separately the fermentative hydrogen production (see, e.g., [15,16]) and the AD for methane production (cf. e.g., [17–20]) are known. However, only a few models of TSAD processes are available; see, for example, [21–24] and the references therein. In [25], both bio-hydrogen and bio-methane productions are optimized in two bioreactors regarding two operating parameters, organic loading rate, and dilution rate; however, the first bioreactor is running in semi-continuous and the second—in batch operation mode because the working volumes of both bioreactors are

Processes 2020, 8, 791 2 of 27

equal. In [26], steady-state analysis for the TSAD process is presented using a very simple model without hydrolysis and without taking into account the most important fact for this scheme—the energy recovery.

In this paper, we propose a mathematical model, describing the process of simultaneous  $H_2$  and  $CH_4$  production by AD of organic wastes in a cascade of two connected continuously stirred bioreactors. The biochemical processes in the first bioreactor include disintegration of organic wastes, hydrolysis, and acidogenesis with hydrogen production. The methane production from acetate (methanogenesis) is separated in the second bioreactor. The proposed model is developed by reducing the well known Anaerobic Digester Model No 1 (ADM1) basic structure elaborated by the International Water Association (IWA) [17]. Modeling of TSAD using ADM1 is also presented in [27], where, however, methane is obtained from both bioreactors.

Our mathematical model is presented for the first time in the conference paper [21]. The present article represents an extension of the latter, providing (i) more detailed assumptions used to construct the model, which are based on practical experiments (Section 3); (ii) more precise and deeper theoretical investigations of the model dynamics such as existence and uniqueness of positive and uniformly bounded solutions, more precise computation of practically important equilibrium points of the model and study of their local asymptotic stability (Section 4); (iii) more detailed investigation concerning the optimization of hydrogen and methane production on steady state operation (Section 5), and showing the existence of maxima, which is important for the applications; (iv) numerical simulations in Section 6, based on a newly developed, so called one-page application "(H2,CH4) Bioreactors ver. 1.2" in the SmoWeb platform.

The main goal of this article is to present rigorous mathematical analysis of the proposed model dynamics, to detect and provide some model-based predictions, which could be useful in designing and engineering real laboratory experiments. The theoretical studies are supported and complimented by a variety of computer simulations.

# 2. Process Description

In the TSAD system, relatively fast growing acidogens and  $H_2$  producing microorganisms are developed in the first-stage hydrogenic bioreactor ( $BR_1$  with working volume  $V_1$ , see Figure 1) and are involved in the production of volatile fatty acids (VFA–mainly acetate and butyrate, [8]) and  $H_2$ . Some recent studies on the microbial community in the hydrogen producing  $BR_1$  show that dominant are microorganisms of the genus Clostridium [28]. In  $BR_1$ , pH is kept in the range 5.0–5.5, and acetate degrading methanogens are eliminated. On the other hand, the slow growing acetogens and methanogens are developed in the second-stage methanogenic bioreactor ( $BR_2$  with working volume  $V_2$ , and pH in the range 6.5–8.5), in which the produced VFA are further converted to  $CH_4$  and  $CO_2$  by methane-forming bacteria from the genus Methanobacterium, Methanococcusect, see, for details, [9] and the references therein.

It is known that in the two-stage  $H_2 + CH_4$  system the energy yields are up to 43% more in comparison to the traditional one-stage  $CH_4$  production process (cf. [8,25]).

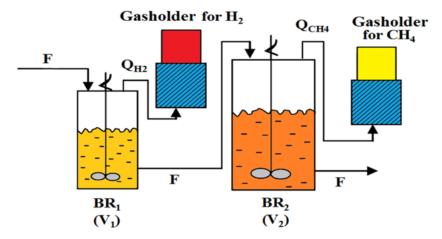
Assume that the volumes  $V_1$  and  $V_2$  of the bioreactors are constant. Let  $F_1$  and  $F_2$  be the inflows in the first and second bioreactor, respectively, and let  $F_1 = F_2 = F$  be valid. It is well known that the dilution rates  $D_1$  and  $D_2$  are defined as

$$D_1 = \frac{F}{V_1}, \quad D_2 = \frac{F}{V_2}.$$

Then,

$$D_2 = \frac{V_1}{V_2} D_1 := \gamma D_1 \text{ within } \gamma := \frac{V_1}{V_2}.$$
 (1)

Processes 2020, 8, 791 3 of 27



**Figure 1.** Two-phases process of AD with production of hydrogen  $(H_2)$  and methane  $(CH_4)$ .

Later on in the paper (see Section 5), by analyzing points of maximum biogas production in both bioreactors, biohydrogen in  $BR_1$  and biomethane in  $BR_2$ , we find theoretical values for the hydraulic retention time (HRT)  $1/D_{1,\max}$  and  $1/D_{2,\max}$ , which show that the volume  $V_2$  of the second bioreactor for methane production should be larger than the volume  $V_1$  of the first bioreactor. Therefore,  $\gamma < 1$  should be valid. We determine the constant  $\gamma$  using the proposed model equations.

#### 3. Model Description

In this section, we present the mathematical model describing the process of simultaneous  $H_2$  and  $CH_4$  production by AD of organic wastes in a cascade of two continuously stirred bioreactors. The model is derived on the basis of the ADM1 basic structure as well as on our experience with the two-phase AD process with hydrogen and methane production, see [29]. The ADM1 is a complex model, describing all known (till its creation) microbiological and biochemical reactions occurring in AD of organic wastes. ADM1 is suitable mainly for process knowledge and simulation, but not appropriate for theoretical studies and model-based control design due to the plenty of input parameters which are difficult to obtain. The reduction of ADM1 is done here in order to simplify the latter and to provide an opportunity for studying the asymptotic stability of the dynamics with respect to two essential indicators: (i) the separation of the model equations into two groups, corresponding to the reactions taking place in each one of the two bioreactors, and (ii) the type of the organic wastes used.

For simplification of the model, the following assumptions have been accepted:

- In  $BR_1$  only the main VFA (acetate) is formed, whereas propionate and butyrate are omitted. In this way, the slow acetogenic phase has been omitted and consequently in  $BR_2$  only methanogenesis occurs.
- In both bioreactors, the existence of inhibition phenomena is not taken into account.
- Balance equations of the hydrogen and methane in the liquid phases have been neglected because they are practically not dissolved in liquids.
- Hydrogenotrophic bacteria, producing methane from hydrogen, are eliminated previously and do not exist in this process.
- Equations describing balances of inorganic components and some biochemical equations have been neglected in view of simplifying the model.
- pH is considered in the range between 5.0 to 5.5 in the first bioreactor  $BR_1$ , and in the interval 6.5–8.5 in the second bioreactor  $BR_2$ . Within these ranges, it is assumed that the negative influence of pH could be disregarded.

These ranges of pH were obtained during our own laboratory experiments with TSAD of waste from maize treatment (not published yet).

Processes 2020, 8, 791 4 of 27

The proposed models of the two bioreactors are of balancing type for continuously stirred tank reactors (CSTR) like the original ADM1 model. For each bioreactor, the ADM1 equations are adapted to describe the dynamics of the most important variables for the AD of organic waste.

Following the above assumptions, we present now the mathematical model for hydrogen and methane production in the two bioreactors. The definitions of the phase variables and parameters in the model equations are given in Tables 1 and 2 below.

The dynamics in the first bioreactor are described by the following set of 10 nonlinear ordinary differential equations:

$$\frac{d}{dt}S_{su}(t) = D_1(S_{su}^{in} - S_{su}) + k_{hyd,ch}X_{ch} + f_{su,li}k_{hyd,li}X_{li} - \mu_{suaa,su}X_{suaa}$$
 (2)

$$\frac{d}{dt}S_{aa}(t) = D_1(S_{aa}^{in} - S_{aa}) + k_{hyd,pr}X_{pr} - \mu_{suaa,aa}X_{suaa}$$
(3)

$$\frac{d}{dt}S_{fa}(t) = D_1(S_{fa}^{in} - S_{fa}) + f_{fa,li}k_{hyd,li}X_{li} - \mu_{fa}X_{fa}$$
(4)

$$\frac{d}{dt}S_{ac}(t) = -D_1S_{ac} + (1 - Y_{suaa}) [f_{ac,su}\mu_{suaa,su} + f_{ac,aa}\mu_{suaa,aa}] X_{suaa} + 0.7 (1 - Y_{fa})\mu_{fa}X_{fa}$$

$$\frac{d}{dt}X_c(t) = D_1(X_c^{in} - X_c) - k_{dis}X_c \tag{6}$$

(5)

$$\frac{d}{dt}X_{ch}(t) = -(D_1 + k_{hyd,ch})X_{ch} + f_{ch,xc}k_{dis}X_c$$
(7)

$$\frac{d}{dt}X_{pr}(t) = -(D_1 + k_{hyd,pr})X_{pr} + f_{pr,xc}k_{dis}X_c$$
(8)

$$\frac{d}{dt}X_{li}(t) = -(D_1 + k_{hyd,li})X_{li} + f_{li,xc}k_{dis}X_c$$
(9)

$$\frac{d}{dt}X_{suaa}(t) = [-D_1 + Y_{suaa}(\mu_{suaa,su} + \mu_{suaa,aa})]X_{suaa}$$
(10)

$$\frac{d}{dt}X_{fa}(t) = -(D_1 - Y_{fa}\mu_{fa})X_{fa}, (11)$$

with gaseous output

$$Q_{h2} = (Y_{h2,su} \mu_{suaa,su} + Y_{h2,aa} \mu_{suaa,aa}) X_{suaa} + Y_{h2,fa} \mu_{fa} X_{fa},$$

where  $Q_{h2}$  is the hydrogen flow rate [L/h]. In (2)–(11), the biomass specific growth rates  $\mu_{suaa,su}$ ,  $\mu_{suaa,aa}$  and  $\mu_{fa}$  are of the form

$$\begin{array}{lcl} \mu_{suaa,su} & = & \mu_{suaa,su}(S_{su},S_{aa}) = k_{m,suaa} \frac{S_{su}}{k_{s,suaa} + S_{su}} \cdot \frac{S_{su}}{S_{su} + S_{aa}}, \\ \mu_{suaa,aa} & = & \mu_{suaa,aa}(S_{su},S_{aa}) = k_{m,suaa} \frac{S_{aa}}{k_{s,suaa} + S_{aa}} \cdot \frac{S_{aa}}{S_{su} + S_{aa}}, \\ \mu_{fa} & = & \mu_{fa}(S_{fa}) = k_{m,fa} \frac{S_{fa}}{k_{s,fa} + S_{fa}}. \end{array}$$

Processes **2020**, *8*, 791

Table 1. Model Variables

	Definition of the Model Variables
$S_{su}$	concentration of monosacharides $[gCOD/L]$
$S_{aa}$	concentration of amino acids (AA) $[gCOD/L]$
$S_{fa}$	concentration of fatty acids (LCFA) $[gCOD/L]$
$S_{ac}$	concentration of total acetate in $BR_1$ [ $gCOD/L$ ]
$S_{ac,ch4}$	concentration of total acetate in $BR_2$ [ $gCOD/L$ ]
$X_c$	concentration of composites $[gCOD/L]$
$X_{ch}$	concentration of carbohydrates $[gCOD/L]$
$X_{pr}$	concentration of proteins $[gCOD/L]$
$X_{li}$	concentration of lipids $[gCOD/L]$
$X_{suaa}$	concentration of sugar and AA degraders $[g/L]$
$X_{fa}$	concentration of LCFA degraders $[g/L]$
$X_{ac}$	concentration of acetate degraders $[g/L]$
$D_1$	dilution rate in the first bioreactor $[h^{-1}]$
$D_2$	dilution rate in the second bioreactor $[h^{-1}]$

Table 2. Model Parameters

	<b>Definition of the Model Parameters</b>	Values
$S_{su}^{in}$	input concentration of $S_{su}$ [ $gCOD/L$ ]	0.01
Sin Sin Sin Sin	input concentration of $S_{aa}$ [gCOD/L]	0.001
$S_{fa}^{in}$	input concentration of $S_{fa}$ [ $gCOD/L$ ]	0.001
$X_c^{in}$	input concentration of $X_c$ [ $gCOD/L$ ]	50
$f_{ch,xc}$	stoichiometric parameter [–]	0.2
$f_{pr,xc}$	stoichiometric parameter [–]	0.2
$f_{li,xc}$	stoichiometric parameter [–]	0.3
$f_{su,li}$	stoichiometric parameter [–]	0.05
$f_{fa,li}$	stoichiometric parameter [–]	0.95
fac,su	stoichiometric parameter [–]	0.41
fac,aa	stoichiometric parameter [–]	0.4
$Y_{suaa}$	stoichiometric parameter [–]	0.1
$Y_{ac}$	stoichiometric parameter [–]	27.3
$Y_{fa}$	stoichiometric parameter [–]	0.06
$Y_{h2,su}$	physicochemical parameter $[L^2/g]$	0.7
$Y_{h2,aa}$	physicochemical parameter $[L^2/g]$	0.7
$Y_{h2,fa}$	physicochemical parameter $[L^2/g]$	0.7
$Y_{ch4,ac}$	physicochemical parameter $[L^2/g]$	75
$k_{dis}$	biochemical parameter $[h^{-1}]$	0.0208
$k_{hyd,ch}$	biochemical parameter $[h^{-1}]$	0.417
$k_{hyd,pr}$	biochemical parameter $[h^{-1}]$	0.417
$k_{hyd,li}$	biochemical parameter $[h^{-1}]$	0.417
$k_{m,suaa}$	biochemical parameter $[h^{-1}]$	1.25
$k_{s,suaa}$	biochemical parameter $[g/L]$	0.5
$k_{m,fa}$	biochemical parameter $[h^{-1}]$	0.15
$k_{m,ac}$	biochemical parameter $[h^{-1}]$	0.0167
$k_{s,fa}$	biochemical parameter $[g/L]$	0.67
$k_{s,ac}$	biochemical parameter $[g/L]$	0.4

We extend the model (2)–(11) by the following two equations describing the process in the second bioreactor (cf. [30]):

$$\frac{d}{dt}S_{ac,ch4}(t) = D_2(S_{ac} - S_{ac,ch4}) - Y_{ac} \mu_{ac,ch4} X_{ac}$$
(12)

$$\frac{d}{dt}X_{ac}(t) = (-D_2 + \mu_{ac,ch4})X_{ac}$$
(13)

with gaseous output

$$Q_{ch4} = Y_{ch4,ac} \mu_{ac,ch4} X_{ac}$$

Processes 2020, 8, 791 6 of 27

where  $Q_{ch4}$  is the methane flow rate [L/h], and

$$\mu_{ac,ch4} = \mu_{ac,ch4}(S_{ac,ch4}) = k_{m,ac} \frac{S_{ac,ch4}}{k_{s,ac} + S_{ac,ch4}}$$

All model coefficients in the Equations (2)–(13) are assumed to be positive. The dilution rates  $D_1$  and  $D_2$  are the decision or control variables. The numerical values in the rightmost column of Table 2 are taken from [31]. In the model variables and parameters, the subscripts h2 and ch4 indicate hydrogen ( $H_2$ ) and methane ( $CH_4$ ), respectively.

### 4. Investigation of the Model Solutions

This section presents a mathematical analysis of the dynamics (2)–(13). First, we establish existence, positivity, uniform boundedness, and uniqueness of the model solutions. The results are presented in Theorem 1. The interested reader can find a detailed proof of Theorem 1 in Appendix A.1 of the Appendix A.

**Theorem 1.** If  $0 < Y_{suaa} < 1$  and  $0 < Y_{fa} < 1$ , then all solutions of the dynamics (2)–(13) are positive and uniformly bounded, and thus exist for all time  $t \in [0, +\infty)$ .

Note that the assumptions  $0 < Y_{suaa} < 1$  and  $0 < Y_{fa} < 1$  in Theorem 1 are not restrictive. Since  $Y_{suaa}$  and  $Y_{fa}$  are yield coefficients, their values are always less than 1 in practice; see last column in Table 2.

In the next two subsections, we find the simplest solutions of the model, namely the equilibrium points, and investigate their local asymptotic stability.

### 4.1. Equilibrium Points of the Model

The equilibrium points of the model are solutions of the algebraic equations obtained from (2)–(13) by setting the right-hand sides equal to zero:

$$D_1(S_{su}^{in} - S_{su}) + k_{hyd,ch}X_{ch} + f_{su,li}k_{hyd,li}X_{li} - \mu_{suaa,su}X_{suaa} = 0$$
(14)

$$D_1(S_{aa}^{in} - S_{aa}) + k_{hyd,pr} X_{pr} - \mu_{suaa,aa} X_{suaa} = 0$$
(15)

$$D_1(S_{fa}^{in} - S_{fa}) + f_{fa,li}k_{hyd,li}X_{li} - \mu_{fa}X_{fa} = 0$$
(16)

$$D_1 S_{ac} - (1 - Y_{suaa}) \left[ f_{ac,su} \mu_{suaa,su} + f_{ac,aa} \mu_{suaa,aa} \right] X_{suaa} - 0.7(1 - Y_{fa}) \mu_{fa} X_{fa} = 0$$
 (17)

$$D_1(X_c^{in} - X_c) - k_{dis}X_c = 0 (18)$$

$$(D_1 + k_{hyd,ch})X_{ch} - f_{ch,xc}k_{dis}X_c = 0 (19)$$

$$(D_1 + k_{hyd,pr})X_{pr} - f_{pr,xc}k_{dis}X_c = 0 (20)$$

$$(D_1 + k_{hyd,li})X_{li} - f_{li,xc}k_{dis}X_c = 0 (21)$$

$$[D_1 - Y_{suaa}(\mu_{suaa,su} + \mu_{suaa,aa})] X_{suaa} = 0$$
(22)

$$(D_1 - Y_{fa}\mu_{fa})X_{fa} = 0. (23)$$

$$D_2(S_{ac} - S_{ac,ch4}) - Y_{ac} \,\mu_{ac,ch4} X_{ac} = 0 \tag{24}$$

$$(D_2 - \mu_{ac.ch4})X_{ac} = 0. (25)$$

We are looking for nonnegative solutions of (14)–(25) since only these equilibrium points correspond to the physical and biological meaning of the model variables. The equilibrium points are computed as functions of the control variables, the dilution rates  $D_1$  and  $D_2$ .

Processes 2020, 8, 791 7 of 27

Using Equation (18), we find the equilibrium component  $\bar{X}_c$  with respect to the phase variable  $X_c$ , which is obviously positive:

$$\bar{X}_c = \frac{D_1 X_c^{in}}{D_1 + k_{dis}}.$$

Furthermore, Equations (19)–(21) imply the equilibrium components  $\bar{X}_{ch}$ ,  $\bar{X}_{pr}$  and  $\bar{X}_{li}$  respectively:

$$\bar{X}_{ch} = \frac{f_{ch,xc} \, k_{dis} \, \bar{X}_c}{D_1 + k_{hvd,ch}}, \quad \bar{X}_{pr} = \frac{f_{pr,xc} \, k_{dis} \, \bar{X}_c}{D_1 + k_{hvd,vr}}, \quad \bar{X}_{li} = \frac{f_{li,xc} \, k_{dis} \, \bar{X}_c}{D_1 + k_{hvd,li}}. \tag{26}$$

Equations (16) and (23) are used to determine the steady state components  $\bar{S}_{fa}$  and  $\bar{X}_{fa}$  with respect to the phase variables  $S_{fa}$  and  $X_{fa}$ . Obviously, Equation (23) suggests that the trivial solution  $X_{fa} \equiv 0$  always exists, but is not of practical interest. Assuming that  $X_{fa} \not\equiv 0$  and using the expression of  $\mu_{fa}$ , we obtain from (23) the following steady state component  $\bar{S}_{fa}$ 

$$\bar{S}_{fa} = \frac{D_1 k_{s,fa}}{Y_{fa} k_{m,fa} - D_1}. (27)$$

It is clear that  $\bar{S}_{fa}$  exists and  $\bar{S}_{fa} > 0$  if and only if

$$D_1 < Y_{fa} k_{m,fa}. \tag{28}$$

Equation (16) then delivers the equilibrium component for  $X_{fa}$ ,

$$\bar{X}_{fa} = \frac{D_1(S_{fa}^{in} - \bar{S}_{fa}) + f_{fa,li}k_{hyd,li}\bar{X}_{li}}{\mu_{fa}(\bar{S}_{fa})}.$$
(29)

Obviously,  $\bar{X}_{fa}$  from (29) satisfies  $\bar{X}_{fa} > 0$  if and only if

$$\bar{S}_{fa} < S_{fa}^{in} + \frac{1}{D_1} f_{fa,li} k_{hyd,li} \bar{X}_{li}.$$
 (30)

**Remark 1.** The quantity  $S_{fa}^{in} + \frac{1}{D_1} f_{fa,li} k_{hyd,li} \bar{X}_{li}$  on the right-hand side of (30) can be considered as a worst-case upper bound of the total concentration of fatty acids (LCFA)  $S_{fa}$ , which accumulation in BR<sub>1</sub> might occur as a result of some imbalances in the degradation phase, leading to wash-out of the LCFA degraders  $X_{fa}$  [32].

The inequality (30) is equivalent with

$$D_{1}^{3}\left(k_{s,fa}+S_{fa}^{in}\right)+D_{1}^{2}\left[\left(k_{s,fa}+S_{fa}^{in}\right)\left(k_{hyd,li}+k_{dis}\right)-Y_{fa}k_{m,fa}S_{fa}^{in}\right] +D_{1}\left[f_{fa,li}k_{hyd,li}f_{li,xc}k_{dis}X_{c}^{in}-Y_{fa}k_{m,fa}S_{fa}^{in}\left(k_{hyd,li}+k_{dis}\right)+k_{s,fa}+S_{fa}^{in}\right] -Y_{fa}k_{m,fa}k_{hyd,li}k_{dis}\left(S_{fa}^{in}+f_{fa,li}f_{li,xc}X_{c}^{in}\right)<0.$$

$$(31)$$

The left hand-side cubic polynomial in (31) always possesses one positive real root  $D_1^{(1)}$ , so the inequality (30) is satisfied if  $D_1 \in (0, D_1^{(1)})$ . Therefore, the equilibrium components  $\bar{S}_{fa}$  and  $\bar{X}_{fa}$  exist and satisfy  $\bar{S}_{fa} > 0$  and  $\bar{X}_{fa} > 0$  if  $D_1 \in (0, \min\{Y_{fa}k_{m,fa}, D_1^{(1)}\})$  holds true.

If  $D_1 = D_1^{(1)}$ , then  $\bar{X}_{fa}$  becomes equal to zero; this means that  $D_1 = D_1^{(1)}$  is a (transcritical) bifurcation value of the parameter  $D_1$ , leading to wash-out of the biomass  $X_{fa}$  for  $D_1 > D_1^{(1)}$ . We shall

Processes 2020, 8, 791 8 of 27

pay special attention to the case  $\bar{X}_{fa}=0$  motivating our considerations later on in the paper (see Section 5). If  $\bar{X}_{fa}=0$ , then we obtain the following steady state component for  $S_{fa}$ 

$$\bar{S}_{fa}^{0} = S_{fa}^{in} + \frac{1}{D_{1}} f_{fa,li} k_{hyd,li} \bar{X}_{li}, \tag{32}$$

which always exists and is positive if  $D_1 > D_1^{(1)}$ .

Here, and in what follows, the superscript 0 indicates  $\bar{X}_{fa} = 0$  in the steady state components.

Furthermore, Equations (14), (15) and (22) are used to compute the steady state components  $\bar{S}_{su}$ ,  $\bar{S}_{aa}$ , and  $\bar{X}_{suaa}$  in case positive solutions do exist. It is clear from Equation (22) that the trivial solution  $X_{suaa} \equiv 0$  always exists, but we exclude this case for biological evidence. Let  $X_{suaa} \not\equiv 0$ . Then, Equation (22) reduces to

$$D_1 - Y_{suaa} \left[ \mu_{suaa,su}(S_{su}, S_{aa}) + \mu_{suaa,aa}(S_{su}, S_{aa}) \right] = 0.$$
 (33)

Using Equation (15), we find

$$X_{suaa} = \frac{D_1(S_{aa}^{in} - S_{aa}) + k_{hyd,pr}\bar{X}_{pr}}{\mu_{suaa,aa}(S_{su}, S_{aa})}.$$
(34)

The equilibrium component with respect to  $X_{suaa}$  should be positive and this will be fulfilled if

$$S_{aa} < S_{aa}^{in} + \frac{1}{D_1} k_{hyd,pr} \bar{X}_{pr}. \tag{35}$$

A similar remark to Remark 1 can be made to the above inequality (35) concerning the concentrations of the amino acids  $S_{aa}$  and of the degraders  $X_{suaa}$ .

Substituting  $X_{suaa}$  from (34) into Equation (14) and using Equation (33), we obtain the following nonlinear algebraic system with respect to  $S_{aa}$  and  $S_{su}$ :

$$\left[D_{1}(S_{su}^{in} - S_{su}) + k_{hyd,ch}\bar{X}_{ch} + f_{su,li}k_{hyd,li}\bar{X}_{li}\right] \mu_{suaa,aa}(S_{su}, S_{aa}) 
- \left[D_{1}(S_{aa}^{in} - S_{aa}) + k_{hyd,pr}\bar{X}_{pr}\right] \mu_{suaa,su}(S_{su}, S_{aa}) = 0$$

$$D_{1} - Y_{suaa} \left[\mu_{suaa,su}(S_{su}, S_{aa}) + \mu_{suaa,aa}(S_{su}, S_{aa})\right] = 0.$$
(36)

It is straightforward to see that system (36) is equivalent to the following one:

$$A_1 S_{aa} (S_{aa} + S_{su}) - A_2 S_{su}^2 (1 + S_{aa}) - A_3 S_{aa} S_{su} (S_{aa} + S_{su}) = 0$$

$$B_1 (S_{aa}^2 + S_{su}^2) + B_2 S_{aa} S_{su} (S_{aa} + S_{su}) - B_3 S_{aa} S_{su} - B_4 (S_{aa} + S_{su}) = 0,$$
(37)

where

$$A_{1} = D_{1}S_{su}^{in} + \frac{D_{1}k_{dis}X_{c}^{in}}{D_{1} + k_{dis}} \left[ \frac{k_{hyd,ch}f_{ch,xc}}{D_{1} + k_{hyd,ch}} + \frac{f_{su,li}k_{hyd,li}f_{li,xc}}{D_{1} + k_{hyd,li}} \right],$$

$$A_{2} = D_{1}S_{aa}^{in} + \frac{D_{1}k_{hyd,pr}f_{pr,xc}k_{dis}X_{c}^{in}}{(D_{1} + k_{dis})(D_{1} + k_{hyd,pr})}, \quad A_{3} = D_{1}k_{s,suaa};$$

$$B_{1} = k_{s,suaa}(Y_{suaa}k_{m,suaa} - D_{1}), \quad B_{2} = Y_{suaa}k_{m,suaa} - D_{1},$$

$$B_{3} = 2D_{1}k_{s,suaa}, \quad B_{4} = D_{1}k_{s,suaa}^{2}.$$

All coefficients  $A_i$ , i = 1, 2, 3, and  $B_j$ , j = 1, 2, 3, 4, depend on  $D_1$  (the dilution rate in the first bioreactor  $BR_1$ ). Assume that there exists a value  $D_1^{(2)} > 0$  of  $D_1$  such that system (37) possesses

Processes 2020, 8, 791 9 of 27

positive solutions  $\bar{S}_{aa}$  and  $\bar{S}_{su}$  for all  $D_1 \in (0, D_1^{(2)})$ . Denote by  $D_1^{(3)}$  a third possible upper bound for  $D_1$ , defined by

$$D_1^{(3)} = \min\{D_1 > 0 : \bar{S}_{aa} < S_{aa}^{in} + \frac{1}{D_1} k_{hyd,pr} \bar{X}_{pr}\}.$$
(38)

Then, the equilibrium component  $\bar{X}_{suaa}$  will exist and will be positive for  $D_1 \in$  $(0, \min\{D_1^{(2)}, D_1^{(3)}\})$ :

$$\bar{X}_{suaa} = \frac{D_1(S_{aa}^{in} - \bar{S}_{aa}) + k_{hyd,pr}\bar{X}_{pr}}{\mu_{suaa,aa}(\bar{S}_{su}, \bar{S}_{aa})}.$$
(39)

Note that an equality  $\bar{S}_{aa} = S_{aa}^{in} + \frac{1}{D_1} k_{hyd,pr} \bar{X}_{pr}$  for some value of  $D_1$  will lead to  $\bar{X}_{suaa} = 0$  and may cause wash-out of the biomass  $X_{suaa}$ , which is practically undesirable, and we shall exclude this case.

Equation (17) is used to find the steady state component

$$\bar{S}_{ac} = \frac{1}{D_1} \left[ (1 - Y_{suaa}) \left( f_{ac,su} \mu_{suaa,su} (\bar{S}_{su}, \bar{S}_{aa}) + f_{ac,aa} \mu_{suaa,aa} (\bar{S}_{su}, \bar{S}_{aa}) \right) \bar{X}_{suaa} + 0.7 (1 - Y_{fa}) \mu_{fa} (\bar{S}_{fa}) \bar{X}_{fa} \right].$$
(40)

The latter is positive if  $Y_{suaa} < 1$  and  $Y_{fa} < 1$  are fulfilled.

If  $\bar{X}_{fa} = 0$ , then the equilibrium component with respect to  $S_{ac}$  becomes

$$\bar{S}_{ac}^{0} = \frac{1}{D_{1}} (1 - Y_{suaa}) \left( f_{ac,su} \mu_{suaa,su} (\bar{S}_{su}, \bar{S}_{aa}) + f_{ac,aa} \mu_{suaa,aa} (\bar{S}_{su}, \bar{S}_{aa}) \right) \bar{X}_{suaa}; \tag{41}$$

 $\bar{S}_{ac}^0$  exists and is positive if  $Y_{suaa} < 1$  and  $D_1 \in \left(D_1^{(1)}, \min\{D_1^{(2)}, D_1^{(3)}\}\right)$  hold true. The above calculations deliver the following equilibrium points of the model (2)–(11):

$$\begin{split} E_1(D_1) &= (\bar{S}_{su}, \bar{S}_{aa}, \bar{S}_{fa}, \bar{S}_{ac}, X_c, X_{ch}, X_{pr}, \bar{X}_{li}, X_{suaa}, X_{fa}) \\ & \qquad \text{if } 0 < D_1 < \min\{Y_{fa}k_{m,fa}, D_1^{(1)}, D_1^{(2)}, D_1^{(3)}\}; \\ E_1^0(D_1) &= (\bar{S}_{su}, \bar{S}_{aa}, \bar{S}_{fa}^0, \bar{S}_{ac}^0, \bar{X}_c, \bar{X}_{ch}, \bar{X}_{pr}, \bar{X}_{li}, \bar{X}_{suaa}, \bar{X}_{fa} = 0) \\ & \qquad \text{if } D_1^{(1)} < D_1 < \min\{D_1^{(2)}, D_1^{(3)}\}. \end{split}$$

Consider now Equations (12) and (13) to compute the equilibrium components with respect to  $X_{ac}$  and  $S_{ac,ch4}$ . Thereby, we assume that  $\bar{S}_{ac}$  and  $\bar{S}_{ac}^0$  are known; see (40) and (41), respectively.

Equation (25) has the obvious solution  $X_{ac} \equiv 0$ , which is not of practical interest. Assuming that  $X_{ac} \neq 0$ , the equality  $\mu_{ac,ch4}(S_{ac,ch4}) = D_2$  delivers the following steady state component for  $S_{ac,ch4}$ 

$$\bar{S}_{ac,ch4} = \frac{k_{s,ac}D_2}{k_{m,ac} - D_2}. (42)$$

Obviously, there exists  $\bar{S}_{ac,ch4} > 0$  if  $D_2 < k_{m,ac}$  holds true. Then, the corresponding equilibrium component  $\bar{X}_{ac}$  is

$$\bar{X}_{ac} = \frac{1}{Y_{ac}} (\bar{S}_{ac} - \bar{S}_{ac,ch4}).$$
 (43)

The latter is positive if

$$\bar{S}_{ac} > \bar{S}_{ac\,ch4}.\tag{44}$$

Processes 2020, 8, 791 10 of 27

If we consider  $\bar{S}_{ac}^0$  instead of  $\bar{S}_{ac}$  we obtain

$$\bar{X}_{ac}^{0} = \frac{1}{Y_{ac}} (\bar{S}_{ac}^{0} - \bar{S}_{ac,ch4}), \tag{45}$$

which is positive if

$$\bar{S}_{ac}^0 > \bar{S}_{ac\,ch4}.\tag{46}$$

The inequalities (44) and (46) give relationships between  $D_1$  and  $D_2$  because  $\bar{S}_{ac}$  and  $\bar{S}_{ac}^0$  depend also on  $D_1$ ; see (40) and (41), respectively.

Therefore, the extended model (2)–(13) can possess the following two equilibrium points, parameterized on  $D_1$  and  $D_2$ :

$$E(D_1, D_2) = E_1(D_1) \cup (\bar{S}_{ac,ch4}, \bar{X}_{ac}),$$
  
 $E^0(D_1, D_2) = E_1^0(D_1) \cup (\bar{S}_{ac,ch4}, \bar{X}_{ac}^0).$ 

We summarize the above calculations in the following theorem.

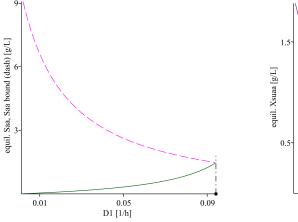
**Theorem 2.** Let  $0 < Y_{suaa} < 1$  and  $0 < Y_{fa} < 1$  be fulfilled in the model Equations (2)–(13). Assume that there exists a value  $D_1^{(2)} > 0$  such that positive solutions  $\bar{S}_{aa}$  and  $\bar{S}_{su}$  of system (37) do exist for  $D_1 \in (0, D_1^{(2)})$ . Then,

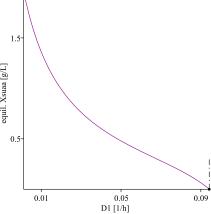
- (i) the equilibrium point  $E(D_1, D_2)$  exists if  $0 < D_1 < \min\{Y_{fa}k_{m,fa}, D_1^{(1)}, D_1^{(2)}, D_1^{(3)}\}$  and  $0 < D_2 < \min\{k_{m,ac}, \min\{D_2 > 0 : \bar{S}_{ac} > \bar{S}_{ac,ch4}\}\};$
- (ii) the equilibrium point  $E^0(D_1, D_2)$  exists if  $D_1^{(1)} < D_1 < \min\{D_1^{(2)}, D_1^{(3)}\}$  and  $0 < D_2 < \min\{k_{m,ac}, \min\{D_2 > 0 : \bar{S}_{ac}^0 > \bar{S}_{ac,ch4}\}\}$ .

Theorem 2 can be specified using the coefficient values in Table 2. Numerical calculations produce the following values:

$$Y_m k_{m,fa} = 0.009, \ D_1^{(1)} \approx 0.00843, \ D_1^{(2)} \approx 0.0942.$$
 (47)

The left plot in Figure 2 visualizes the equilibrium component  $\bar{S}_{aa}$ , which exists for  $D_1 \in (0, D_1^{(2)})$ , as well as the upper bound  $S_{aa}^{in} + \frac{1}{D_1} k_{hyd,pr} \bar{X}_{pr}$  of  $\bar{S}_{aa}$  according to (35). Obviously, (35) is satisfied for  $D_1 \in (0, D_1^{(2)})$  so that  $D_1^{(2)} \leq D_1^{(3)}$ . The right plot in Figure 2 shows the equilibrium component  $\bar{X}_{suaa}(D_1)$ , which vanishes at  $D_1 = D_1^{(2)}$ .





**Figure 2.** Left plot: the equilibrium point  $\bar{S}_{aa}$  (solid line) and the upper bound of  $\bar{S}_{aa}$  (dash line) according to (35) as functions of  $D_1$ . Right plot: the graph of the equilibrium point  $\bar{X}_{suaa}$ . The vertical dash-dot lines pass through  $D_1^{(2)}$ .

Processes 2020, 8, 791 11 of 27

Furthermore, we have  $k_{m,ac} = 0.0167$  and

$$\bar{S}_{ac} > \bar{S}_{ac,ch4} \iff D_2 \in (0, 0.0156) =: (0, D_2^{(2)}).$$
 (48)

This relation is demonstrated in the left plot of Figure 3. The steady state component  $\bar{S}_{ac} = \bar{S}_{ac}(D_1)$  (solid line) exists for  $D_1 \in (0, D_1^{(1)})$ . The dash line represents the equilibrium component  $\bar{S}_{ac,ch4} = \bar{S}_{ac,ch4}(D_2)$ . The horizontal dot line passes through the points  $(D_1^{(1)}, \bar{S}_{ac}(D_1^{(1)}))$  and  $(D_2^{(2)}, \bar{S}_{ac,ch4}(D_2^{(2)})$ . The vertical dash-dot lines pass through  $D_1^{(1)}$  and  $D_2^{(2)}$ . This implies the following relation:

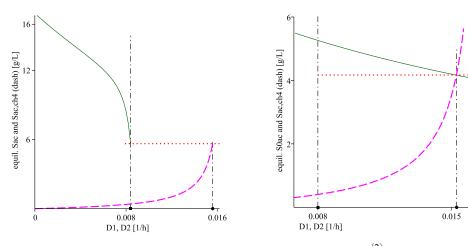
$$E(D_1, D_2) \text{ exists } \iff D_1 \in (0, D_1^{(1)}) = (0, 0.00843), \quad D_2 \in (0, D_2^{(2)}) = (0, 0.0156).$$

Similarly, the right plot in Figure 3 visualizes the relation

$$\bar{S}_{ac}^0 > \bar{S}_{ac,ch4} \iff D_2 \in (0,0.0152) =: (0,D_2^{(1)})$$

because the horizontal dot line passes through the points  $(D_1^{(1)}, \bar{S}_{ac}^0(D_1^{(1)}))$  and  $(D_2^{(1)}, \bar{S}_{ac,ch4}(D_2^{(1)}))$ . This leads to the following relation:

$$E^0(D_1,D_2) \text{ exists } \iff D_1 \in (D_1^{(1)},D_1^{(2)}) = (0.00843,0.0942), \quad D_2 \in (0,D_2^{(1)}) = (0,0.0152).$$



**Figure 3.** Existence of the equilibrium component  $\bar{S}_{ac,ch4}$  for  $D_2 \in (0, D_2^{(2)})$  (left plot) and for  $D_2 \in (0, D_2^{(1)})$  (right plot).

The equilibrium  $E(D_1, D_2)$  corresponds to coexistence of all microorganisms in both bioreactors  $BR_1$  and  $BR_2$ , whereas  $E^0(D_1, D_2)$  is related to wash-out of the LCFA degrader  $X_{fa}$  in  $BR_1$  for hydrogen production.

# 4.2. Local Asymptotic Stability of the Equilibrium Points

We shall first investigate the local asymptotic stability of the coexistence equilibrium  $E_1(D_1)$  for  $D_1 \in (0, D_1^{(1)})$  and of the wash-out steady state (within  $\bar{X}_{fa} = 0$ )  $E_1^0(D_1)$  for  $D_1 \in (D_1^{(1)}, D_1^{(2)})$ , i.e., of the equilibrium points corresponding to the model (2)–(11) of the first bioreactor  $BR_1$  for hydrogen production.

It is well known that an equilibrium point is locally asymptotically stable, if all eigenvalues of the Jacobi matrix evaluated at this equilibrium have negative real parts, cf. e.g., [33]. The eigenvalues of this Jacobi matrix coincide with the roots of the corresponding characteristic polynomial.

Processes 2020, 8, 791 12 of 27

Detailed calculations of the characteristic polynomials  $P(E_1; \lambda)$  and  $P(E_1^0; \lambda)$  with respect to the equilibrium points  $E_1 = E_1(D_1)$  and  $E_1^0 = E_1^0(D_1)$  are given in Appendix A.2 of the Appendix A. We present them here for reader's convenience:

$$\begin{split} P(E_1;\lambda) &= (-D_1 - \lambda)(-D_1 - k_{dis} - \lambda)(-D_1 - k_{hyd,ch} - \lambda) \\ &\times (-D_1 - k_{hyd,pr} - \lambda)(-D_1 - k_{hyd,li} - \lambda) \\ &\times \left[\lambda^2 + \left(D_1 + \frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right)\lambda + D_1\frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right] \\ &\times R(E_1;\lambda), \quad D_1 \in (0,D_1^{(1)}); \\ P(E_1^0;\lambda) &= (-D_1 - \lambda)^2(-D_1 - k_{dis} - \lambda)(-D_1 - k_{hyd,ch} - \lambda) \\ &\times (-D_1 - k_{hyd,pr} - \lambda)(-D_1 - k_{hyd,li} - \lambda)(-D_1 + Y_{fa}\mu_{fa}(\bar{S}_{fa}^0) - \lambda) \\ &\times R(E_1^0;\lambda), \quad D_1 \in (D_1^{(1)},D_1^{(2)}), \end{split}$$

where  $R(E_1;\lambda)$  is the characteristic polynomial of the sub-matrix  $\Delta^{(3)}$  evaluated at the steady state components  $(\bar{S}_{su},\bar{S}_{aa},\bar{X}_{suaa})$  for  $D_1 \in (0,D_1^{(1)})$ , and  $R(E_1^0;\lambda)$  is the characteristic polynomial of the sub-matrix  $\Delta^{(3)}$  evaluated at the steady state components  $(\bar{S}_{su},\bar{S}_{aa},\bar{X}_{suaa})$  for  $D_1 \in (D_1^{(1)},D_1^{(2)})$ , see Appendix A.2 in the Appendix A.

The linear quantities in  $P(E_1; \lambda) = 0$  and  $P(E_1^0; \lambda) = 0$  produce negative real eigenvalues of the Jacobi matrices. In the first polynomial  $P(E_1; \lambda)$ , the quadratic equation

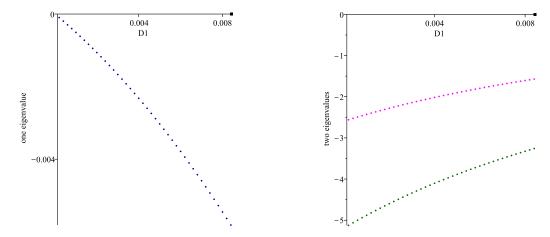
$$\lambda^{2} + \left(D_{1} + \frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right)\lambda + D_{1}\frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa} = 0$$

possesses two real negative roots. The multiplier  $-D_1 + Y_{fa}\mu_{fa}(S_{fa}^0) - \lambda$  in the second polynomial  $P(E_1^0;\lambda)$  delivers a negative root  $\lambda = -D_1 + Y_{fa}\mu_{fa}(\bar{S}_{fa}^0)$  because, in this case,  $D_1 > Y_{fa}\mu_{fa}(\bar{S}_{fa}^0)$  holds true.

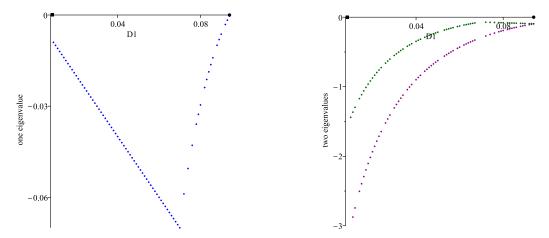
It remains to show that  $R(E_1;\lambda)=0$  and  $R(E_1^0;\lambda)=0$  have roots with negative real parts. These two problems are solved numerically using the coefficient values in Table 2. The real parts of the eigenvalues are presented graphically in Figures 4 and 5, and are obviously negative. Therefore, the steady states  $E_1=E_1(D_1),\,D_1\in(0,D_1^{(1)})$  and  $E_1^0=E_1^0(D_1),\,D_1\in(D_1^{(1)},D_1^{(2)})$ , are locally asymptotically stable.

Consider again the polynomial  $P(E_1; \lambda)$ . We know that at  $D_1 = D_1^{(1)}$  the equilibrium component  $\bar{X}_{fa}$  becomes equal to zero, and thus  $P(E_1; \lambda) = 0$  has a root  $\lambda = 0$ . This means that a transcritical bifurcation at  $D_1 = D_1^{(1)}$  occurs, leading to wash-out of the LCFA degraders  $X_{fa}$ . As a result, the steady state  $E_1(D_1)$  no longer exists for  $D_1 > D_1^{(1)}$ , and  $E_1^0(D_1)$  remains the locally asymptotically stable equilibrium for  $D_1 \in (D_1^{(1)}, D_1^{(2)})$ .

Processes 2020, 8, 791 13 of 27



**Figure 4.** Eigenvalues of  $\Delta^{(3)}$  evaluated at  $(\bar{S}_{su}, \bar{S}_{aa}, \bar{X}_{suaa})$  for  $D_1 \in (0, D_1^{(1)})$ . The solid box on the horizontal axis denotes  $D_1^{(1)}$ .



**Figure 5.** Eigenvalues of  $\Delta^{(3)}$  evaluated at  $(\bar{S}_{su}, \bar{S}_{aa}, \bar{X}_{suaa})$  for  $D_1 \in (D_1^{(1)}, D_1^{(2)})$ . The solid box and solid circle on the horizontal axis denote  $D_1^{(1)}$  and  $D_1^{(2)}$ , respectively.

The equilibrium points of the model (12)–(13) of the second bioreactor  $BR_2$  for methane production are known to be locally asymptotically stable for any admissible value of  $D_2$  (see, e.g., [20]). Indeed, the characteristic polynomials of the Jacobi matrices of (12)–(13) evaluated at the equilibrium points  $(\bar{S}_{ac,ch4}, \bar{X}_{ac})$  and  $(\bar{S}_{ac,ch4}, \bar{X}_{ac}^0)$ , have the form

$$\lambda^2 + A \lambda + B = 0$$
 and  $\lambda^2 + A_0 \lambda + B_0 = 0$ ,

where

$$A = D_2 + Y_{ac} \frac{d\mu_{ac,ch4}}{dS_{ac,ch4}} (\bar{S}_{ac,ch4}) \bar{X}_{ac} > 0, \quad B = Y_{ac} \mu_{ac,ch4} (\bar{S}_{ac,ch4}) \frac{d\mu_{ac,ch4}}{dS_{ac,ch4}} (\bar{S}_{ac,ch4}) \bar{X}_{ac} > 0,$$

$$A_0 = D_2 + Y_{ac} \frac{d\mu_{ac,ch4}}{dS_{ac,ch4}} (\bar{S}_{ac,ch4}) \bar{X}_{ac}^0 > 0, \quad B_0 = Y_{ac} \mu_{ac,ch4} (\bar{S}_{ac,ch4}) \frac{d\mu_{ac,ch4}}{dS_{ac,ch4}} (\bar{S}_{ac,ch4}) \bar{X}_{ac}^0 > 0.$$

The positive signs of A, B and  $A_0$ ,  $B_0$  imply that the above quadratics possess two roots with negative real parts. This proves the local asymptotic stability of the equilibrium points  $E(D_1, D_2)$  for  $D_1 \in (0, D_1^{(1)})$ ,  $D_2 \in (0, D_2^{(2)})$  and  $E^0(D_1, D_2)$  for  $D_1 \in (D_1^{(1)}, D_1^{(2)})$ ,  $D_2 \in (0, D_2^{(1)})$ .

Processes 2020, 8, 791 14 of 27

#### 5. Optimization of Hydrogen and Methane Flow Rates

In this section, we shall find values of the dilution rates  $D_1$  and  $D_2$ , so that the hydrogen and methane flow rates  $Q_{h2}$  and  $Q_{ch4}$  evaluated at the model equilibrium points achieve maximal values.

We compute  $Q_{h2}$  on the two sets of steady states  $E_1(D_1)$ ,  $D_1 \in (0, D_1^{(1)})$  and  $E_1^0(D_1)$ ,  $D_1 \in (D_1^{(1)}, D_1^{(2)})$ :

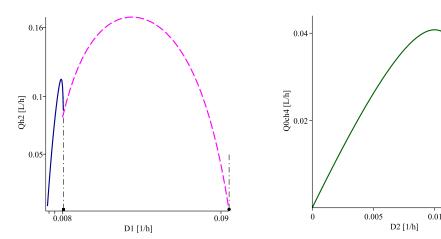
$$\begin{split} Q_{h2} &= Q_{h2}(D_1) = Y_{h2,su} \; \mu_{suaa,su}(\bar{S}_{su},\bar{S}_{aa}) \bar{X}_{suaa} \\ &+ Y_{h2,aa} \mu_{suaa,aa}(\bar{S}_{su},\bar{S}_{aa}) \bar{X}_{suaa} + Y_{h2,fa} \; \mu_{fa}(\bar{S}_{fa}) \bar{X}_{fa}, \quad D_1 \in \left(0,D_1^{(1)}\right); \\ Q_{h2}^0 &= Q_{h2}^0(D_1) = Y_{h2,su} \; \mu_{suaa,su}(\bar{S}_{su},\bar{S}_{aa}) \bar{X}_{suaa} \\ &+ Y_{h2,aa} \mu_{suaa,aa}(\bar{S}_{su},\bar{S}_{aa}) \bar{X}_{suaa}, \quad D_1 \in \left(D_1^{(1)},D_1^{(2)}\right) \end{split}$$

Recall that the superscript 0 in the equilibrium  $E_1^0(D_1)$  indicates  $\bar{X}_{fa} = 0$ .

The functions  $Q_{h2}(D_1)$  and  $Q_{h2}^0(D_1)$  depend on the decision (manipulated) input  $D_1$  and are called input–output static characteristics with respect to the hydrogen production.

The important (from practical point of view) question is whether the functions  $Q_{h2}(D_1)$  and  $Q_{h2}^0(D_1)$  are unimodal with respect to  $D_1$  in the admissible intervals for  $D_1$ . For example, the function  $Q_{h2}(D_1)$  is called unimodal if there exists a unique (admissible) point  $D_{1,\max}$ , such that  $Q_{h2}(D_1)$  possesses maximum  $Q_{h2,\max} = Q_{h2}(D_{1,\max})$ ,  $Q_{h2}(D_1)$  is strongly increasing if  $D_1 < D_{1,\max}$  and  $Q_{h2}(D_1)$  is strongly decreasing if  $D_1 > D_{1,\max}$ .

Figure 6 (left plot) presents the graphs of the input–output static characteristics  $Q_{h2}(D_1)$  for  $D_1 \in (0, D_1^{(1)})$  (solid line) and  $Q_{h2}^0(D_1)$  for  $D_1 \in (D_1^{(1)}, D_1^{(2)})$  (dash line). The function  $Q_{h2}(D_1)$  achieves its maximum at  $D_{1,\max} \approx 0.0072 \in (0, D_1^{(1)})$  and  $Q_{h2,\max} \approx 0.115$  [L/h]. The maximum of  $Q_{h2}^0(D_1)$  is achieved at  $D_{1,\max}^0 \approx 0.0439 \in (D_1^{(1)}, D_1^{(2)})$  and  $Q_{h2,\max}^0 \approx 0.169$  [L/h]. Therefore, the absolute maximum of  $Q_{h2}(D_1) \cup Q_{h2}^0(D_1)$ ,  $D_1 \in (0, D_1^{(2)})$  coincides with the maximum of  $Q_{h2}^0(D_1)$ . From a practical point of view, this means that the maximal flow rate of  $H_2$  could be achieved at a operation mode, related to wash-out of the LCFA degraders  $X_{fa}$  in  $BR_1$ .



**Figure 6.** The input–output static characteristics  $Q_{h2}(D_1) \cup Q_{h2}^0(D_1)$  for  $D_1 \in (0, D_1^{(2)})$  (left) and  $Q_{ch4}^0(D_{1,max}^0, D_2)$  for  $D_2 \in (0, D_2^{(1)})$  (right). The vertical dash-dot lines in the left plot pass through  $D_1^{(1)}$  (solid box) and  $D_1^{(2)}$  (solid circle). In the right plot,  $D_2^{(1)}$  is marked by a sold box.

A similar question of maximizing the biogas production at steady states has been discussed in [34] for the well known four-dimensional AM2 model. There, two maxima of the input-output static characteristics with respect to methane flow rate have also been determined, related to coexistence or wash-out of the acidogenic bacteria in the bioreactor.

Processes 2020, 8, 791 15 of 27

Using the methane flow rate  $Q_{ch4}$ , we compute further the input–output static characteristics  $Q_{ch4}$  and  $Q_{ch4}^0$  on the set of the steady states  $E(D_1, D_2)$  and  $E^0(D_1, D_2)$  respectively, namely:

$$Q_{ch4} = Q_{ch4}(D_1, D_2) = Y_{ch4,ac} \, \mu_{ac,ch4}(\bar{S}_{ac,ch4}) \, \bar{X}_{ac},$$

$$Q_{ch4}^0 = Q_{ch4}(D_1, D_2) = Y_{ch4,ac} \, \mu_{ac,ch4}(\bar{S}_{ac,ch4}) \, \bar{X}_{ac}^0.$$

The latter are parameterized on the two inputs  $D_1$  and  $D_2$ , since  $\bar{X}_{ac}$  and  $\bar{X}_{ac}^0$  depend on both  $D_1$  and  $D_2$ ; see (43) and (45), respectively.

With  $D_1=D^0_{1,\max}\approx 0.0439$ , we compute the equilibrium components  $\bar{S}_{ac,ch4}$  and  $\bar{X}^0_{ac}$  according to (42) and (45), as well as  $Q^0_{ch4}$ , all of them as functions of  $D_2$ .

Figure 6 (right plot) visualizes the graph of  $Q_{ch4}^0(D_{1,\text{max}}^0,D_2)$  for  $D_2 \in (0,D_2^{(1)})$ ; the latter is a unimodal function, taking its maximum at  $D_{2,\text{max}}^0 \approx 0.00991$  [1/h] with  $Q_{ch4,\text{max}}^0 = Q_{ch4}(D_{1,\text{max}}^0,D_{2,\text{max}}^0) \approx 0.0388$  [L/h].

The latter results are theoretical and predicted by the model studies. Experimental data that confirm these results by considering  $H_2$  and  $CH_4$  volumetric production under similar conditions for which our model was developed (i.e., two bioreactors in a cascade with a similar ratio of working volumes and operating in continuous mode) can be found in [35].

Using the presentation (1), we define and compute the constant  $\gamma$  and in this way the relationship between the volumes  $V_1$  and  $V_2$  of the two bioreactors:

$$\gamma = rac{D_{2, ext{max}}^0}{D_{1, ext{max}}^0} pprox 0.226 \quad \Longrightarrow \quad V_2 pprox 4.42 \, V_1.$$

The last relation implies that the volume of the second bioreactor  $BR_2$  for methane production should be at least 4.4 times greater than the volume of the first bioreactor  $BR_1$  for hydrogen production.

## 6. Dynamic Behavior of the Model Solutions

The dynamic model has been implemented as one-page application "(H2,CH4) Bioreactors ver. 1.2" in a web-based simulation software SmoWeb, see <a href="http://platform.sysmoltd.com/">http://platform.sysmoltd.com/</a>. SmoWeb is an open computational platform in *Python* and a library of applications is built on top of this platform. It does not require installation of any software by the user, since all computations are performed in a web cloud. *Python* is used as the implementation language because it is a powerful modern general purpose object-oriented language. Nowadays, *Python* has become an inseparable part of the scientific computing due to the vast abundance of open source libraries and interfaces to major software tools. A number of one-page applications have already been and are continuously being developed on top of the functionality provided by the platform, including bioprocess Modeling as well.

All figures below are developed using the one-page application "(H2,CH4) Bioreactors ver. 1.2". The plots illustrate the dynamic behavior of the model solutions when  $D_1$ ,  $D_2$  and the input concentration of the composites  $X_c^{in}$  take different values from Tables 3 and 4, respectively.

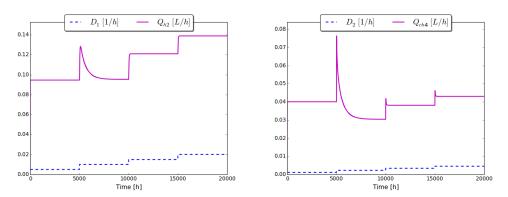
Time [h]	0-5000	5000-10,000	10,000-15,000	15,000-20,000
$D_1 [1/h]$	0.005	0.01	0.015	0.02
$D_2 [1/h]$	0.00113	0.00226	0.00339	0.00452

**Table 3.** Values of  $D_1$  and  $D_2$ .

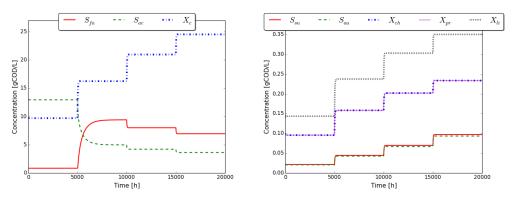
Figure 7 presents the time evolution of the flow rates  $Q_{h2}(t)$  and  $Q_{ch4}(t)$  when  $D_1$  and  $D_2$  take values from Table 3. Figures 8 and 9 display the model solutions by variable  $D_1$  and  $D_2$  given in Table 3. It is seen that, if  $D_1 \in (0, D_1^{(1)}) = (0, 0.00843)$  and  $D_2 \in (0, D_2^{(2)}) = (0, 0.0156)$ , then the model solutions tend to the corresponding components of the locally stable coexistence equilibrium

Processes 2020, 8, 791 16 of 27

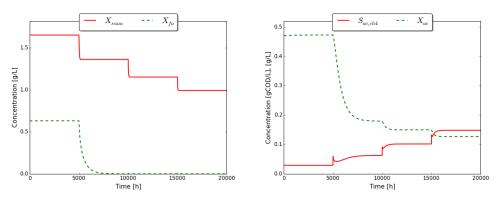
 $E(D_1,D_2)$ . When  $D_1 \in (D_1^{(1)},D_1^{(2)}) = (0.00843,0.0942)$  and  $D_2 \in (0,D_2^{(1)}) = (0,0.0152)$ , then the solutions approach the locally stable wash-out equilibrium (with  $\bar{X}_{fa}=0$ ) after sufficiently large time t [h]; in particular, one can see in the left plot of Figure 9 that  $X_{fa}(t)$  approaches 0 for sufficiently large time t [h].



**Figure 7.** Time evolution of  $Q_{h2}$  and  $Q_{ch4}$  with variable  $D_1$  and  $D_2$  from Table 3.



**Figure 8.** Dynamic behavior of the model solutions with variable  $D_1$  from Table 3.



**Figure 9.** Dynamic behavior of the model solutions with variable  $D_1$  from Table 3.

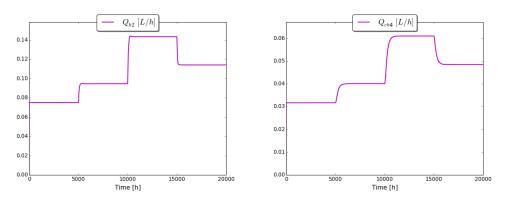
Time $[h]$	0-5000	5000-10,000	10,000-15,000	15,000-20,000
$X_c^{in}$ [gCOD/L]	40	50	75	60

**Table 4.** Values of  $X_c^{in}$ .

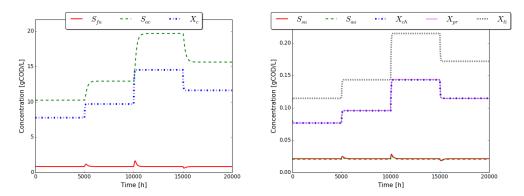
Figures 10–12 demonstrate the dynamics of  $Q_{h2}(t)$ ,  $Q_{ch4}(t)$  as well as of the model solutions, when  $D_1$  and  $D_2$  are fixed,  $D_1 = 0.005 \ [1/h]$ ,  $D_2 = 0.00113 \ [1/h]$ , but the input concentration of composites  $X_c^{in}$  varies taking values from Table 4. The model solutions tend to the corresponding

Processes 2020, 8, 791 17 of 27

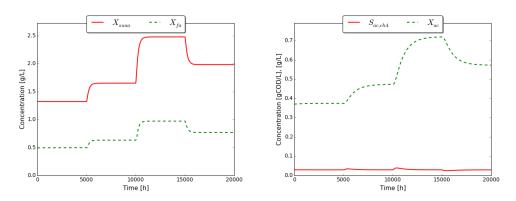
components of the coexistence equilibrium  $E(D_1, D_2)$ , which is the locally stable steady state. These numerical simulations demonstrate the fact that uncertainties in the input  $X_c^{in}$  do not affect the stability of the model dynamics.



**Figure 10.** Time evolution of  $Q_{h2}$  and  $Q_{ch4}$  with variable  $X_c^{in}$  from Table 4.



**Figure 11.** Dynamic behavior of the model solutions with variable  $X_c^{in}$  from Table 4.



**Figure 12.** Dynamic behavior of the model solutions with variable  $X_c^{in}$  from Table 4.

#### 7. Discussion

In this paper, we propose a mathematical model describing the process of simultaneous production of hydrogen and methane by anaerobic digestion of organic wastes in a cascade of two connected continuously stirred bioreactors with different volumes. The proposed model is developed by adapting and reducing the universal Anaerobic Digester Model No 1 (ADM1). The equations in our model are separated in two groups, corresponding to the processes in the two bioreactors, for hydrogen and for methane production, respectively. The simplifications made in the two models naturally lead to some limitations compared to the original ADM1. However, as already shown, the latter also does not cover all possible reactions, e.g., possible inclusion in the model of the activity of the so called

Processes 2020, 8, 791 18 of 27

syntrophicacetate oxidizers [36]. Some new, more complex models have also been developed [37], which however have the same main disadvantage as ADM1—great complexity and thus impossibility to carry out some mathematical analytical studies.

The investigation of the hydrogen and methane flow rates,  $Q_{h2}$  and  $Q_{ch4}$  respectively, on steady state operation, shows existence of maxima for some values of the dilution rates  $D_1$  and  $D_2$  in the two bioreactors  $BR_1$  and  $BR_2$ . The local maximum for  $D_1$  in  $BR_1$  reflects the maximal yield of hydrogen from lipids at very long hydraulic retention time HRT =  $1/D_{1,max} \approx 1/0.0072 \approx 139$  [hours]  $\approx 5.8$  [days]. This maximum is much less than the maximum yield of hydrogen derived from monosaccharides and aminoacids (proteins) for the accepted combination of coefficients and initial conditions reflecting the biodegradation of organic wastes. In [35] under similar conditions, results close to our theoretical model-based predictions are obtained.

The model also allows us to find the optimal ratio between the volumes  $V_1$  and  $V_2$  of the two bioreactors subject to the same optimization goal. Many studies in the literature report on TSAD processes; however, most of them are operated manually and the ratio of working volumes of CSTRs is not discussed. In some references, a ratio of 10 is accepted without comments. In [38], a TSAD system consisting of two CSTRs operating at mesophilic conditions are used to investigate the effect of HRT on hydrogen and methane production; the ratio of working volumes of the bioreactors is equal to 8 (without comments). In [39], optimization of TSAD of separately collected municipality bio-waste is carried out in pilot scale CSTRs at thermophilic temperature, using recirculation of the digestate in the second bioreactor to maintain pH in the first bioreactor at optimal value. There, the ratio of working volumes of the bioreactors is equal to 3.8 (without comments). In [40], optimal loading rate is obtained providing maximum  $H_2$  and  $CH_4$  productions in TSAD of cassava wastewaters using specific thermophilic bioreactors and constant recycling ratio of 1:1 with automatic control of pH in the hydrogenic bioreactor. The ratio of working volumes of both bioreactors is equal to 6 (without explanations). The review article [41] is the first one that combines optimization approaches for three possible products from AD—methane, hydrogen and volatile fatty acids (VFA), taking into account different process parameters and types of bioreactors, including acidogenesis and methanogenesis separation in two different bioreactors. However, the ratio of the bioreactors working volumes is not given and discussed.

Using the adapted values of the coefficients in our model, we obtain an optimal ratio of the working volumes of both bioreactors equal to 4.42 using a criterion for maximizing the bioenergy production. This value falls into the range of the above reported values; however, it is theoretically derived.

#### 8. Conclusions

In this paper, we consider a mathematical model describing a biotechnological process of hydrogen and methane production by anaerobic digestion. The process is carried out in two connected continuously stirred bioreactors, thus the model equations are separated in two groups, describing the processes in each bioreactor. We establish existence and uniqueness of positive and uniformly bounded solutions of the model dynamics. Two equilibrium points of the model are computed and their local asymptotic stability is studied with respect to the practically important input parameters, the dilution rates  $D_1$  and  $D_2$  in the two bioreactors. The first equilibrium point corresponds to coexistence of substrates and biomass, and the second one is related to wash-out of the LCFA degraders in the first bioreactor. We show that there exists a value of the dilution rate  $D_1$ , at which the coexistence equilibrium disappears as a result of bifurcation of the steady state and the wash-out equilibrium remains the only locally asymptotically stable equilibrium. The investigation of the hydrogen and methane flow rates,  $Q_{h2}$  and  $Q_{ch4}$ , on steady state operation, shows existence of maxima at some values of the dilution rates  $D_1$  and  $D_2$ . It is worth noting that the maximal hydrogen flow rate is achieved at the steady state where the LCFA degrader  $X_{fa}$  is washed out. This is established here by taking numerical values for the model parameters from the literature [31]. Nevertheless, this fact could

Processes 2020, 8, 791 19 of 27

help in the design of an AD process to achieve maximal production of either hydrogen and methane. The future work envisages extending the model of the second bioreactor by equations describing an acetogenesis phase, where other VFA (propionate, butyrate and valeriate) will be transformed into acetate, as well as including inhibitory kinetics of the microorganisms growth. The elaborated web-based software will allow us to check different model-based optimization and control strategies that will contribute to designing and engineering a real process.

**Author Contributions:** M.B.: software development, simulation and visualization, writing and editing; N.D.: theoretical investigation of the model—existence and uniqueness of solutions, stability analysis of equilibrium points, writing and editing; I.S.: model development and validation, discussion and evaluation of the results, editing. All authors have read and agreed to the published version of the manuscript.

Funding: This research received no external funding.

**Acknowledgments:** This research has been partially supported by the Bulgarian Science Fund under contract No. DFNI–E02/13. The work of the second author has been partially supported by Grant No. BG05M2OP001-1.001-0003, financed by the Science and Education for Smart Growth Operational Program (2014–2020) in Bulgaria and co-financed by the European Union through the European Structural and Investment Funds. The authors are grateful to the anonymous reviewers for the useful comments that improved the content of this article.

Conflicts of Interest: The authors declare no conflict of interest.

### Appendix A

Appendix A.1. Proof of Theorem 1

Equation (6) is linear with respect to  $X_c$  and can be solved explicitly:

$$X_c(t) = X_c(0)e^{-(D_1 + k_{dis})t} + \frac{D_1 X_c^{in}}{D_1 + k_{dis}}.$$

The solution  $X_c(t)$  obviously exists for all  $t \ge 0$ , is positive, and is uniformly bounded. In practice, it is reasonable to consider only positive initial conditions  $X_c(0) > 0$ .

Assume that  $X_{ch}(0)=0$ . Then, Equation (7) implies  $\dot{X}_{ch}(0)=f_{ch,xc}k_{dis}X_c(0)>0$  and so there exists  $\varepsilon>0$  such that  $X_{ch}(t)>0$  for each  $t\in(0,\varepsilon)$ . Let  $X_{ch}(0)\geq0$  and suppose that there exists  $\bar{t}_{ch}>0$  such that  $X_{ch}(t)>0$  for each  $t\in(0,\bar{t}_{ch})$  and  $X_{ch}(\bar{t}_{ch})=0$ . Then,  $\dot{X}_{ch}(\bar{t}_{ch})\leq0$ , but it follows from (7) that  $\dot{X}_{ch}(\bar{t}_{ch})>0$ , a contradiction. Thus,  $X_{ch}(t)>0$  for all  $t\geq0$ .

Similar arguments as above can be applied to  $X_{pr}$  and  $X_{li}$  from Equations (8) and (9), respectively, so that  $X_{pr}(t) > 0$  and  $X_{li}(t) > 0$  for all  $t \ge 0$ .

Equation (10) implies that, if  $X_{suaa}(0) = 0$ , then  $X_{suaa}(t) = 0$  for all  $t \in [0, \infty)$  due to the uniqueness of solutions of Cauchy problems. Assuming that  $X_{suaa}(0) > 0$ , we obtain

$$X_{suaa}(t) = X_{suaa}(0) e^{\int_0^t [-D_1 + Ysuaa(\mu_{suaa,su}(S_{su}(\xi), S_{aa}(\xi)) + \mu_{suaa,aa}(S_{su}(\xi), S_{aa}(\xi)))]d\xi} > 0 \text{ for all } t \geq 0.$$

Similarly, if  $X_{fa}(0)=0$ , then it follows from Equation (11) that  $X_{fa}(t)=0$  for all  $t\in[0,\infty)$ . Thus, it is reasonable to consider only positive initial conditions  $X_{fa}(0)>0$ . Then, we have

$$X_{fa}(t) = X_{fa}(0) e^{-\int_0^t \left[D_1 - Y_{fa}\mu_{fa}(S_{fa}(\xi))\right]d\xi} > 0 \text{ for all } t \ge 0.$$

Assume now that  $X_{fa}(0) > 0$ ,  $X_{suaa}(0) > 0$  and  $X_{j}(0) \ge 0$  for  $j \in \{c, ch, pr, li\}$ . If  $S_{su}(0) = 0$ , then it follows from Equation (2) that

$$\dot{S}_{su}(0) = D_1 S_{su}^{in} + k_{hvd,ch} X_{ch}(0) + f_{su,li} k_{hvd,li} X_{li}(0) > 0,$$

thus there exists  $\varepsilon > 0$  with  $S_{su}(t) > 0$  for  $t \in (0, \varepsilon)$ . Let  $S_{su}(0) \ge 0$ . Assume that there exists  $\bar{t}_{su} > 0$  such that  $S_{su}(t) > 0$  for  $t \in (0, \bar{t}_{su})$  and  $S_{su}(\bar{t}_{su}) = 0$ . Then,  $\dot{S}_{su}(\bar{t}_{su}) \le 0$ , but, from (2), it follows that

Processes 2020, 8, 791 20 of 27

 $\dot{S}_{su}(\bar{t}_{su}) > 0$ , a contradiction. Therefore,  $S_{su}(t) > 0$  for all t > 0. Using Equations (3) and (4), one can show in a similar way that  $S_{aa}(t) > 0$  and  $S_{fa}(t) > 0$  for all  $t \ge 0$ .

Denote

$$\Sigma_1 = X_{ch} + f_{ch,xc}X_c$$
,  $\Sigma_1^{in} = f_{ch,xc}X_c^{in}$ .

Since  $X_{ch}(t) > 0$  for all  $t \ge 0$ , it follows from Equations (6) and (7) that

$$\dot{\Sigma}_{1} = \dot{X}_{ch} + f_{ch,xc}\dot{X}_{c} = D_{1}\left(f_{ch,xc}X_{c}^{in} - (X_{ch} + f_{ch,xc}X_{c})\right) - k_{hyd,ch}X_{ch} \leq D_{1}\left(\Sigma_{1}^{in} - \Sigma_{1}\right).$$

Multiplying both sides of the inequality by  $e^{D_1t}>0$ , we obtain  $e^{D_1t}\dot{\Sigma}_1+D_1e^{D_1t}\Sigma_1\leq e^{D_1t}D_1\Sigma_1^{in}$ , which is equivalent to  $\frac{d}{dt}\left(e^{D_1t}\Sigma_1\right)\leq e^{D_1t}D_1\Sigma_1^{in}$ , and further

$$\int_0^t \frac{d}{d\xi} e^{D_1 \xi} \Sigma_1(\xi) d\xi \le D_1 \Sigma_1^{in} \int_0^t e^{D_1 \xi} d\xi,$$

thus finally

$$\Sigma_1 \le e^{-D_1 t} \Sigma_1(0) + \Sigma_1^{in} \left( 1 - e^{-D_1 t} \right).$$

The latter inequality means that  $\limsup_{t\to\infty} \Sigma_1(t) \leq \Sigma_1^{in}$ . Since  $X_c(t)$  is positive and bounded, and  $X_{ch}(t)$  is positive, this implies that  $X_{ch}(t)$  is uniformly bounded and exists for all  $t\in[0,\infty)$ .

Denote  $\Sigma_2 = X_{pr} + f_{pr,xc}X_c$ ,  $\Sigma_2^{in} = f_{pr,xc}X_c^{in}$ . Then, using the fact that  $X_{ch}(t) \ge 0$  for all  $t \ge 0$ , we obtain from Equations (6) and (9)

$$\dot{\Sigma}_2 = D_1 \left( f_{pr,xc} X_c^{in} - X_{pr} - f_{pr,xc} X_c \right) - k_{hyd,pr} X_{pr} \le D_1 \left( \Sigma_2^{in} - \Sigma_2 \right),$$

which means that  $\limsup_{t\to\infty} \Sigma_2(t) \leq \Sigma_2^{in}$ . Since  $X_c(t)$  is positive and bounded, the latter inequality implies that  $X_{pr}(t)$  is also uniformly bounded, and exists for all  $t \in [0, \infty)$ .

Define  $\Sigma_3 = X_{li} + f_{li,xc}X_c$  and  $\Sigma_3^{in} = f_{li,xc}X_c^{in}$ . Then, using Equations (6) and (9) and the fact that  $X_{li}(t) \ge 0$  for all  $t \ge 0$ , we obtain

$$\dot{\Sigma}_3 = D_1 \left( f_{li,xc} X_c^{in} - X_{li} - f_{li,xc} X_c \right) - k_{hyd,li} X_{li} \le D_1 \left( \Sigma_3^{in} - \Sigma_3 \right),$$

which implies  $\limsup_{t\to\infty} \Sigma_3(t) \leq \Sigma_3^{in}$ . Since  $X_c(t)$  is positive and bounded, the latter inequality means that  $X_{li}(t)$  is also uniformly bounded, and exists for all  $t \in [0, \infty)$ .

Consider now

$$\Sigma_4 = S_{fa} + \frac{1}{Y_{fa}} X_{fa} + f_{fa,li} X_{li} + f_{fa,li} f_{li,xc} X_c, \quad \Sigma_4^{in} = S_{fa}^{in} + f_{fa,li} f_{li,xc} X_c^{in}.$$

Then, we obtain from Equations (6), (9), and (11)

$$\dot{\Sigma}_4 = D_1 \left( \Sigma_4^{in} - \Sigma_4 \right)$$
, thus  $\Sigma_4(t) = \Sigma_4(0) e^{-D_1 t} + \Sigma_4^{in} \left( 1 - e^{-D_1 t} \right) > 0$ .

Since  $X_c(t)$  and  $X_{li}(t)$  are positive and bounded,  $S_{fa}(t)$  and  $X_{fa}(t)$  are positive, the latter inequality implies that  $S_{fa}(t)$  and  $X_{fa}(t)$  are uniformly bounded and exists for all  $t \in [0, \infty)$ .

Further consider

$$\Sigma_{5} = S_{su} + S_{aa} + \frac{1}{Y_{suaa}} X_{suaa} + X_{ch} + X_{pr} + f_{su,li} X_{li} + (f_{ch,xc} k_{dis} + f_{li,xc} f_{su,li} + f_{pr,xc}) X_{c},$$

$$\Sigma_5^{in} = S_{su}^{in} + S_{aa,in} + (f_{ch,xc}k_{dis} + f_{li,xc}f_{su,li} + f_{pr,xc}) X_c^{in}$$

Processes 2020, 8, 791 21 of 27

Then, we obtain

$$\dot{\Sigma}_5 = D_1 \left( \Sigma_5^{in} - \Sigma_5 \right) \ \Rightarrow \ \Sigma_5(t) = \Sigma_5(0) e^{-D_1 t} + \Sigma_5^{in} \left( 1 - e^{-D_1 t} \right).$$

Since  $X_c(t)$ ,  $X_{ch}(t)$ ,  $X_{li}(t)$  and  $X_{pr}(t)$  are positive and bounded, the above presentation implies that  $S_{su}(t)$ ,  $S_{aa}(t)$  and  $X_{suaa}(t)$  are also uniformly bounded, and exist for all  $t \in [0, \infty)$ .

Consider now Equation (5). Since  $S_{ac}(t)$  serves as input into the second bioreactor, see Equation (12), it is reasonable to consider only positive initial values  $S_{ac}(0) > 0$ . The positivity of the phase variables included on the right-hand side of Equation (5) imply that  $\dot{S}_{ac}(t) \geq -D_1 S_{ac}(t)$  and thus  $S_{ac}(t) \geq S_{ac}(0)e^{-D_1t} > 0$  for all  $t \in [0, \infty)$ .

Furthermore, the boundedness of the specific growth rate functions imply that there exists a constant  $\Gamma_1 > 0$  such that  $\dot{S}_{ac}(t) \leq -D_1 S_{ac}(t) + \Gamma_1$  for each  $t \geq 0$ . Then, we have consecutively

$$\frac{d}{dt} \left( e^{D_1 t} S_{ac}(t) \right) \le \Gamma_1 e^{D_1 t} 
e^{D_1 t} S_{ac}(t) \le S_{ac}(0) + \Gamma_1 \int_0^t e^{D_1 \xi} d\xi 
S_{ac}(t) \le e^{-D_1 t} S_{ac}(0) + \frac{\Gamma_1}{D_1} \left( 1 - e^{-D_1 t} \right),$$

which means that  $\limsup_{t\to\infty} S_{ac}(t) \le \frac{\Gamma_1}{D_1}$ .

Therefore, all solutions of the model (2)–(11) are positive, uniformly bounded, and thus exist for all  $t \ge 0$ .

Consider now the models (12) and (13). Obviously, if  $X_{ac}(0) = 0$ , then  $X_{ac}(t) = 0$  for all  $t \ge 0$ . Therefore, it is reasonable to consider only positive initial conditions  $X_{ac}(0) > 0$ . Then,

$$X_{ac}(t) = X_{ac}(0)e^{\int_0^t \left[-D_2 + \mu_{ac,ch4}(S_{ac,ch4}(\xi))\right]d\xi} > 0 \text{ for all } t \ge 0.$$

If  $S_{ac,ch4}(0)=0$ , then  $\dot{S}_{ac,ch4}(0)=D_2S_{ac}(0)>0$ . Assume that there exists  $\hat{t}>0$  such that  $S_{ac,ch4}(t)\geq 0$  for  $t\in (0,\hat{t})$  but  $S_{ac,ch4}(\hat{t})=0$ . Then, there exists  $\varepsilon>0$  such that  $\dot{S}_{ac,ch4}(t)>0$  for each  $t\in (\hat{t}-\varepsilon,\hat{t})$ , and so

$$S_{ac,ch4}(\hat{t}) = S_{ac,ch4}(t) + \int_{t}^{\hat{t}} \dot{S}_{ac,ch4}(\xi) d\xi > 0,$$

which is a contradiction. Thus,  $S_{ac,ch4}(t) > 0$  for all  $t \ge 0$ .

Finally, denote  $\Sigma_6 = S_{ac,ch4} + Y_{ac}X_{ac}$ . Since  $S_{ac}(t)$  is positive and uniformly upper bounded, there exists a constant  $\Gamma_2 > 0$  such that  $S_{ac}(t) \le \Gamma_2$  for all  $t \ge 0$ . Then, it is straightforward to see that

$$\dot{\Sigma}_6 = D_2(S_{ac} - \Sigma_6) \le D_2(\Gamma_2 - \Sigma_6),$$

which means that  $\Sigma_6(t) \leq \Sigma_6(0)e^{-D_2t} + \Gamma_2(1-e^{-D_2t})$ . Thus,  $\limsup_{t\to\infty} \Sigma_6(t) \leq \Gamma_2$ . Since  $S_{ac,ch4}$  and  $X_{ac}$  are positive, it follows that they are bounded as well. This proves Theorem 1.

Processes 2020, 8, 791 22 of 27

Appendix A.2. Characteristic Polynomials of the Equilibrium Points  $E_1(D_1)$  and  $E_1^0(D_1)$ 

We introduce the following notations for simplicity and better readability:

$$\begin{array}{lll} g_{1} & = & D_{1}(S_{su}^{in} - S_{su}) + k_{hyd,ch}X_{ch} + f_{su,li}k_{hyd,li}X_{li} - \mu_{suaa,su}X_{suaa} \\ g_{2} & = & D_{1}(S_{aa}^{in} - S_{aa}) + k_{hyd,pr}X_{pr} - \mu_{suaa,aa}X_{suaa} \\ g_{3} & = & D_{1}(S_{fa}^{in} - S_{fa}) + f_{fa,li}k_{hyd,li}X_{li} - \mu_{fa}X_{fa} \\ g_{4} & = & -D_{1}S_{ac} + (1 - Y_{suaa}) \left[ f_{ac,su}\mu_{suaa,su} + f_{ac,aa}\mu_{suaa,aa} \right] X_{suaa} + 0.7(1 - Y_{fa})\mu_{fa}X_{fa} \\ g_{5} & = & D_{1}(X_{c}^{in} - X_{c}) - k_{dis}X_{c} \\ g_{6} & = & -(D_{1} + k_{hyd,ch})X_{ch} + f_{ch,xc}k_{dis}X_{c} \\ g_{7} & = & -(D_{1} + k_{hyd,pr})X_{pr} + f_{pr,xc}k_{dis}X_{c} \\ g_{8} & = & -(D_{1} + k_{hyd,li})X_{li} + f_{li,xc}k_{dis}X_{c} \\ g_{9} & = & \left[ -D_{1} + Y_{suaa}(\mu_{suaa,su} + \mu_{suaa,aa}) \right]X_{suaa} \\ g_{10} & = & -(D_{1} - Y_{fa}\mu_{fa})X_{fa}. \end{array}$$

The Jacobi matrix J related to  $g_j$ ,  $j=1,2,\ldots,10$ , with respect to  $(S_{su},S_{aa},S_{fa},S_{ac},X_c,X_{ch},X_{pr},X_{li},X_{suaa},X_{fa})$  has the form

$$J = \begin{pmatrix} \frac{\partial g_1}{\partial S_{su}} & \frac{\partial g_1}{\partial S_{aa}} & \frac{\partial g_1}{\partial S_{fa}} & \frac{\partial g_1}{\partial S_{ac}} & \frac{\partial g_1}{\partial X_c} & \frac{\partial g_1}{\partial X_{ch}} & \frac{\partial g_1}{\partial X_{pr}} & \frac{\partial g_1}{\partial X_{li}} & \frac{\partial g_1}{\partial X_{suaa}} & \frac{\partial g_1}{\partial X_{fa}} \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} & \frac{\partial g_2}{\partial S_{fa}} & \frac{\partial g_2}{\partial S_{ac}} & \frac{\partial g_2}{\partial X_c} & \frac{\partial g_2}{\partial X_{ch}} & \frac{\partial g_2}{\partial X_{pr}} & \frac{\partial g_2}{\partial X_{li}} & \frac{\partial g_2}{\partial X_{suaa}} & \frac{\partial g_2}{\partial X_{fa}} \\ \vdots & & & & & & & \\ \frac{\partial g_{10}}{\partial S_{su}} & \frac{\partial g_{10}}{\partial S_{aa}} & \frac{\partial g_{10}}{\partial S_{fa}} & \frac{\partial g_{10}}{\partial S_{ac}} & \frac{\partial g_{10}}{\partial X_c} & \frac{\partial g_{10}}{\partial X_{ch}} & \frac{\partial g_{10}}{\partial X_{pr}} & \frac{\partial g_{10}}{\partial X_{li}} & \frac{\partial g_{10}}{\partial X_{suaa}} & \frac{\partial g_{10}}{\partial X_{fa}} \end{pmatrix}$$

Using the explicit dependance of  $g_i$ , j = 1, 2, ..., 10, on the above variables, we obtain

The characteristic polynomial corresponding to the Jacobi matrix J is defined by  $det(J - \lambda I_{10})$ , where  $\lambda$  is any complex number and  $I_{10}$  is the  $(10 \times 10)$ -identity matrix.

We note that the Jacobian  $J=(J_{ij})$ ,  $i,j=1,2,\ldots,10$ , is a sparse matrix, whose fourth column and fifth row contain only one nonzero element, i.e.,  $J_{44}=\frac{\partial g_4}{S_{ac}}=-D_1$  and  $J_{55}=\frac{\partial g_5}{X_c}=-D_1-k_{dis}$ , respectively. Then, we obtain

Processes 2020, 8, 791 23 of 27

$$det(J - \lambda I_{10}) = (-D_1 - \lambda)(-D_1 - k_{dis} - \lambda) \times$$

$$\begin{vmatrix} \frac{\partial g_1}{\partial S_{su}} - \lambda & \frac{\partial g_1}{\partial S_{aa}} & 0 & k_{hyd,ch} & 0 & f_{su,li}k_{hyd,li} & -\mu_{suaa,su} & 0 \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} - \lambda & 0 & 0 & k_{hyd,pr} & 0 & -\mu_{suaa,aa} & 0 \\ 0 & 0 & \frac{\partial g_3}{\partial S_{fa}} - \lambda & 0 & 0 & f_{fa,li}k_{hyd,li} & 0 & -\mu_{fa} \\ 0 & 0 & 0 & -D_1 - k_{hyd,ch} - \lambda & 0 & 0 & 0 & 0 \\ 0 & 0 & 0 & 0 & -D_1 - k_{hyd,pr} - \lambda & 0 & 0 & 0 \\ 0 & 0 & 0 & 0 & 0 & -D_1 - k_{hyd,pr} - \lambda & 0 & 0 & 0 \\ 0 & 0 & 0 & 0 & 0 & -D_1 - k_{hyd,li} - \lambda & 0 & 0 \\ \frac{\partial g_9}{\partial S_{su}} & \frac{\partial g_9}{\partial S_{aa}} & 0 & 0 & 0 & 0 & \frac{\partial g_9}{\partial S_{suaa}} - \lambda & 0 \\ 0 & 0 & \frac{\partial g_{10}}{\partial S_{fa}} & 0 & 0 & 0 & 0 & \frac{\partial g_{10}}{\partial S_{fa}} - \lambda \end{vmatrix}$$

$$= (-D_1 - \lambda)(-D_1 - k_{dis} - \lambda)(-D_1 - k_{hyd,ch} - \lambda)(-D_1 - k_{hyd,pr} - \lambda)(-D_1 - k_{hyd,li} - \lambda)$$

$$\begin{vmatrix} \frac{\partial g_1}{\partial S_{su}} - \lambda & \frac{\partial g_1}{\partial S_{aa}} & 0 & -\mu_{suaa,su} & 0 \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} - \lambda & 0 & -\mu_{suaa,aa} & 0 \\ \\ \times & 0 & 0 & \frac{\partial g_3}{\partial S_{fa}} - \lambda & 0 & -\mu_{fa} \\ \frac{\partial g_9}{\partial S_{su}} & \frac{\partial g_9}{\partial S_{aa}} & 0 & \frac{\partial g_9}{\partial X_{suaa}} - \lambda & 0 \\ 0 & 0 & \frac{\partial g_{10}}{\partial S_{fa}} & 0 & \frac{\partial g_{10}}{\partial X_{fa}} - \lambda \end{vmatrix}$$

$$= (-D_1 - \lambda)(-D_1 - k_{dis} - \lambda)(-D_1 - k_{hyd,ch} - \lambda)(-D_1 - k_{hyd,pr} - \lambda)(-D_1 - k_{hyd,li} - \lambda)$$
$$\times det(\Delta^{(5)} - \lambda I_5),$$

where  $I_5$  is the  $(5 \times 5)$ -identity matrix and  $\Delta^{(5)}$  denotes the matrix

$$\Delta^{(5)} = \begin{pmatrix} \frac{\partial g_1}{\partial S_{su}} & \frac{\partial g_1}{\partial S_{aa}} & 0 & -\mu_{suaa,su} & 0\\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} & 0 & -\mu_{suaa,aa} & 0\\ 0 & 0 & \frac{\partial g_3}{\partial S_{fa}} & 0 & -\mu_{fa}\\ \frac{\partial g_9}{\partial S_{su}} & \frac{\partial g_9}{\partial S_{aa}} & 0 & \frac{\partial g_9}{\partial X_{suaa}} & 0\\ 0 & 0 & \frac{\partial g_{10}}{\partial S_{fa}} & 0 & \frac{\partial g_{10}}{\partial X_{fa}} \end{pmatrix}$$

Furthermore, we obtain

Processes 2020, 8, 791 24 of 27

$$det(\Delta^{(5)} - \lambda I_{5}) = -\mu_{fa} \begin{vmatrix} \frac{\partial g_{1}}{\partial S_{su}} - \lambda & \frac{\partial g_{1}}{\partial S_{aa}} & 0 & -\mu_{suaa,su} \\ \frac{\partial g_{2}}{\partial S_{su}} & \frac{\partial g_{2}}{\partial S_{aa}} - \lambda & 0 & -\mu_{suaa,aa} \\ \frac{\partial g_{9}}{\partial S_{su}} & \frac{\partial g_{9}}{\partial S_{aa}} & 0 & \frac{\partial g_{9}}{\partial X_{suaa}} - \lambda & 0 \\ 0 & 0 & \frac{\partial g_{10}}{\partial S_{fa}} & 0 \end{vmatrix}$$

$$+ \left( \frac{\partial g_{10}}{\partial X_{fa}} - \lambda \right) \begin{vmatrix} \frac{\partial g_{1}}{\partial S_{su}} - \lambda & \frac{\partial g_{1}}{\partial S_{aa}} & 0 & -\mu_{suaa,su} \\ \frac{\partial g_{2}}{\partial S_{su}} & \frac{\partial g_{2}}{\partial S_{aa}} - \lambda & 0 & -\mu_{suaa,aa} \\ 0 & 0 & \frac{\partial g_{3}}{\partial S_{fa}} - \lambda & 0 \\ \frac{\partial g_{9}}{\partial S_{su}} & \frac{\partial g_{9}}{\partial S_{aa}} & 0 & \frac{\partial g_{9}}{\partial X_{suaa}} - \lambda \end{vmatrix}$$

$$= \left[ \mu_{fa} \frac{\partial g_{10}}{\partial S_{fa}} + \left( \frac{\partial g_{10}}{\partial X_{fa}} - \lambda \right) \left( \frac{\partial g_{3}}{\partial S_{fa}} - \lambda \right) \right] \times det(\Delta^{(3)} - \lambda I_{3}),$$

where  $I_3$  is the  $(3 \times 3)$ -identity matrix and  $\Delta^{(3)}$  denotes the matrix

$$\Delta^{(3)} = \begin{pmatrix} \frac{\partial g_1}{\partial S_{su}} & \frac{\partial g_1}{\partial S_{aa}} & -\mu_{suaa,su} \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} & -\mu_{suaa,aa} \\ \frac{\partial g_9}{\partial S_{su}} & \frac{\partial g_9}{\partial S_{aa}} & \frac{\partial g_9}{\partial X_{suaa}} \end{pmatrix}.$$

Replacing the derivatives  $\frac{\partial g_{10}}{\partial S_{fa}}$ ,  $\frac{\partial g_{10}}{\partial X_{fa}}$  and  $\frac{\partial g_{3}}{\partial S_{fa}}$  by their corresponding explicit expressions, the characteristic polynomial  $det(J-\lambda I_{10})$  is presented by

$$\begin{split} \det(J-\lambda I_{10}) &= (-D_1-\lambda)(-D_1-k_{dis}-\lambda)(-D_1-k_{hyd,ch}-\lambda)(-D_1-k_{hyd,pr}-\lambda) \\ &\times \quad (-D_1-k_{hyd,li}-\lambda) \left[ \mu_{fa} Y_{fa} X_{fa} \frac{d\mu_{fa}}{dS_{fa}} + \left(-D_1 + Y_{fa} \mu_{fa} - \lambda\right) \left(-D_1 - X_{fa} \frac{d\mu_{fa}}{dS_{fa}} - \lambda\right) \right] \\ &\times \quad \det(\Delta^{(3)}-\lambda I_3). \end{split}$$

The corresponding partial derivatives in  $\Delta^{(3)}$  are as follows:

$$\begin{split} \frac{\partial g_1}{\partial S_{su}} &= -D_1 - \frac{\partial \mu_{suaa,su}}{\partial S_{su}} \bar{X}_{suaa} \\ \frac{\partial g_2}{\partial S_{su}} &= -\frac{\partial \mu_{suaa,aa}}{\partial S_{su}} \bar{X}_{suaa} \\ \frac{\partial g_2}{\partial S_{su}} &= -\frac{\partial \mu_{suaa,aa}}{\partial S_{su}} \bar{X}_{suaa} \\ \frac{\partial g_9}{\partial S_{su}} &= Y_{suaa} \bar{X}_{suaa} \left( \frac{\partial \mu_{suaa,su}}{\partial S_{su}} + \frac{\partial \mu_{suaa,aa}}{\partial S_{su}} \right) \\ \frac{\partial g_9}{\partial S_{aa}} &= Y_{suaa} \bar{X}_{suaa} \left( \frac{\partial \mu_{suaa,su}}{\partial S_{aa}} + \frac{\partial \mu_{suaa,aa}}{\partial S_{aa}} \right) \\ \frac{\partial g_9}{\partial S_{aa}} &= Y_{suaa} \bar{X}_{suaa} \left( \frac{\partial \mu_{suaa,su}}{\partial S_{aa}} + \frac{\partial \mu_{suaa,aa}}{\partial S_{aa}} \right) \\ \frac{\partial g_9}{\partial S_{aa}} &= -(D_1 - Y_{fa} \mu (\bar{S}_{fa}). \end{split}$$

(i). The characteristic polynomial of the coexistence equilibrium

$$E_1 = E_1(D_1) = (\bar{S}_{su}, \bar{S}_{aa}, \bar{S}_{fa}, \bar{S}_{ac}, \bar{X}_c, \bar{X}_{ch}, \bar{X}_{pr}, \bar{X}_{li}, \bar{X}_{suaa}, \bar{X}_{fa}) > 0 \text{ for } D_1 \in (0, D_1^{(1)}).$$

Processes 2020, 8, 791 25 of 27

Denote by  $P(E_1; \lambda)$  the characteristic polynomial  $det(J - \lambda I_{10})$  evaluated at the equilibrium  $E_1$ , and by  $R(E_1; \lambda)$  the characteristic polynomial  $det(\Delta^{(3)} - \lambda I_3)$  evaluated at the equilibrium components  $(\bar{S}_{su}, \bar{S}_{aa}, \bar{X}_{suaa})$  for  $D_1 \in (0, D_1^{(1)})$ . Taking into account that  $-D_1 + Y_{fa}\mu_{fa}(\bar{S}_{fa}) = 0$ , we obtain for  $P(E_1; \lambda)$  the presentation

$$\begin{split} P(E_1;\lambda) &= (-D_1 - \lambda)(-D_1 - k_{dis} - \lambda)(-D_1 - k_{hyd,ch} - \lambda) \\ &\times (-D_1 - k_{hyd,pr} - \lambda)(-D_1 - k_{hyd,li} - \lambda) \\ &\times \left[\lambda^2 + \left(D_1 + \frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right)\lambda + D_1\frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right] \times R(E_1;\lambda). \end{split}$$

(ii). The characteristic polynomial of the equilibrium

$$E_1^0 = E_1^0(D_1) = (\bar{S}_{su}, \bar{S}_{aa}, \bar{S}_{fa}^0, \bar{S}_{ac}^0, \bar{X}_c, \bar{X}_{ch}, \bar{X}_{pr}, \bar{X}_{li}, \bar{X}_{suaa}, \bar{X}_{fa} = 0) \text{ for } D_1 \in \left(D_1^{(1)}, D_1^{(2)}\right).$$

Denote by  $P(E_1^0;\lambda)$  the characteristic polynomial  $det(J-\lambda I_{10})$  evaluated at the equilibrium  $E_1^0$ , and by  $R(E_1^0;\lambda)$  the characteristic polynomial  $det(\Delta^{(3)}-\lambda I_3)$  evaluated at the equilibrium components  $(\bar{S}_{su},\bar{S}_{aa},\bar{X}_{suaa})$  for  $D_1\in \left(D_1^{(1)},D_1^{(2)}\right)$ . Taking into account that  $\bar{X}_{fa}=0$  and that  $-D_1+Y_{fa}\mu_{fa}(\bar{S}_{fa}^0)\neq 0$  in this case, we obtain for  $P(E_1^0;\lambda)$  the presentation

$$P(E_1^0; \lambda) = (-D_1 - \lambda)^2 (-D_1 - k_{dis} - \lambda)(-D_1 - k_{hyd,li} - \lambda)(-D_1 - k_{hyd,pr} - \lambda) \times (-D_1 - k_{hyd,li} - \lambda)(-D_1 + Y_{fa}\mu_{fa}(\bar{S}_{fa}^0) - \lambda) \times R(E_1^0; \lambda).$$

#### References

- 1. Ahring, B. (Ed.) Biomethanation, I and II; Springer: Berlin/Heidelberg, Germany, 2003.
- 2. Deublein, D.; Steinhauser, A. *Biogas from Waste and Renewable Resources*; Wiley-VCH Verlag: Weinheim, Germany, 2008.
- 3. Musa Egieya, J.; Čuček, L.; Zirngast, K.; Isafiade, A.J.; Pahor, B.; Kravanja, Z. Synthesis of biogas supply networks using various biomass and manure types. *Comput. Chem.* **2019**, 122, 129–151. [CrossRef]
- 4. Aceves-Lara, C.-A.; Latrille, E.; Steyer, J.-P. Optimal control of hydrogen production in a continuous anaerobic fermentation bioreactor. *Intern. J. Hydrog. Energy* **2010**, *35*, I0710–I0718. [CrossRef]
- 5. Ausiello, A.; Micoli, L.; Turco, M.; Toscano, G.; Florio, C.; Pirozzi, D. Biohydrogen production by dark fermentation of Arundo donax using a new methodology for selection of H2-producing bacteria. *Intern. J. Hydrog. Energy* **2017**, *42*, 30599–30612. [CrossRef]
- 6. Guo, X.M.; Trably, E.; Latrille, E.; Carrere, H.; Steyer, J.-P. Hydrogen production from agricultural waste by dark fermentation: A review. *Int. J. Hydrog. Energy* **2010**, *36*, 10660–10673. [CrossRef]
- 7. Pakarinen, O.M.; Kaparaju, P.L.N.; Rintala, J.A. Hydrogen and methane yields of untreated, water-extracted and acid (HCl) treated maize in one- and two-stage batch assays. *Int. J. Hydrog. Energy* **2011**, *36*, 14401–144407. [CrossRef]
- 8. Ruggeri, B.; Tommasi, T.; Sanfilippo, S. *BioH2 & BioCH4 through Anaerobic Digestion (From Research to Full-Scale Applications)*; Springer: London, UK, 2015.
- 9. Gerardi, M.H. The Microbiology of Anaerobic Digesters; John Wiley & Sons, Inc.: Hoboken, NJ, USA, 2003.
- Cárdenas, E.L.M.; Zapata-Zapata, A.D.; Kim, D. Hydrogen Production from Coffee Mucilage in Dark Fermentation with Organic Wastes. *Energies* 2019, 12, 71. [CrossRef]
- 11. Xu, L.; Wang, Y.; Shah, S.A.A.; Zameer, H.; Solangi, Y.A.; Walasai, G.D.; Siyal, Z.A. Economic viability and environmental efficiency analysis of hydrogen production processes for the decarbonization of energy systems. *Processes* **2019**, *7*, 494. [CrossRef]

Processes 2020, 8, 791 26 of 27

12. Prapinagsorn, W.; Sittijunda, S.; Reungsang, A. Co-Digestion of Napier Grass and Its Silage with Cow Dung for Bio-Hydrogen and Methane Production by Two-Stage Anaerobic Digestion Process. *Energies* **2018**, *11*, 47. [CrossRef]

- 13. Al-Rubaye, H.; Karambelkar, S.; Shivashankaraiah, M.M.; Smith, J.D. Process simulation of two-stage anaerobic digestion for methane production. *Biofuels* **2019**, *10*, 181–191. [CrossRef]
- 14. Farhat, A.; Miladi, B.; Hamdi, M.; Bouallagui, H. Fermentative hydrogen and methane co-production from anaerobic co-digestion of organic wastes at high loading rate coupling continuously and sequencing batch digesters. *Environ. Sci. Pollut. Res.* **2018**, *25*, 27945–27958. [CrossRef]
- 15. Nasr, N.; Hafez, H.; Naggar, M.H.E.; Nakhla, G. Application of artificial neural networks for modeling of biohydrogen production. *Int. J. Hydrog. Energy* **2013**, *38*, 3189–3195. [CrossRef]
- 16. Wang, J.; Wan, W. Kinetic models for fermentative hydrogen production: A review. *Int. J. Hydrog. Energy* **2009**, *34*, 3313–3323. [CrossRef]
- 17. Batstone, D.J.; Keller, J.; Angelidaki, I.; Kalyuzhnyi, S.V.; Pavlostathis, S.G.; Rozzi, A.; Sanders, W.T.M.; Siegrist, H.; Vavilin, V.A. The IWA Anaerobic Digestion Model No. 1 (ADM1). *Water Sci. Technol.* **2002**, 45, 65–73. [CrossRef] [PubMed]
- 18. Dochain, D.; Vanrolleghem, P.A.V. *Dynamical Modeling and Estimation in Wasterwater Treatement Process*; IWA Publishing: London, UK, 2001.
- 19. Neba, F.A.; Asiedu, N.Y.; Addo, A.; Morken, J.; Østerhus, S.W.; Seidu, R. Biodigester rapid analysis and design system (B–RADeS): A candidate attainable region-based simulator for the synthesis of biogas reactor structures. *Comput. Chem.* **2020**, *132*, 106607. [CrossRef]
- 20. Simeonov, I. Dynamical modeling and estimation in wasterwater treatement process. In *Contemporary Approaches to Modeling, Optimisation and Control of Biotechnological Processes*; Tzonkov, S., Ed.; Prof. M. Drinov Acad. Publ. House: Sofia, Bulgaria, 2010.
- 21. Borisov, M.; Dimitrova, N.; Simeonov, I. Mathematical Modeling of anaerobic digestion with hydrogen and methane production. In Proceedings of the 6th IFAC Conference on Foundations of Systems Biology in Engineering, the International Federation of Automatic Control, Magdeburg, Germany, 9–12 October 2016; IFAC–PapersOnLine (TuPP.2: 1–8); Volume 49, pp. 231–238.
- 22. Chorukova, E.; Simeonov, I. Mathematical modeling of the anaerobic digestion in two-stage system with production of hydrogen and methane including three intermediate products. *Int. J. Hydrog. Energy* **2020**, 45, 11550–11558. [CrossRef]
- 23. Simeonov, I.; Chorukova, E. Mathematical Modeling of the Anaerobic Digestion with Production of Hydrogen and Methane. In Proceedings of the 4th International Conference on Water, Energy and Environment (ICWEE), Burgas, Bulgaria, 1–3 June 2016; pp. 32–38.
- 24. Guellout, Z.; Clion, V.; Benguerba, Y.; Dumas, C.; Ernst, B. Study of the dark fermentative hydrogen production using modified ADM1 models. *Biochem. Eng. J.* 2018, *132*, 9–19. [CrossRef]
- 25. Schievano, A.; Tenca, A.; Lonati, S.; Manzini, E.; Adani, F. Can two-stage instead of one-stage anaerobic digestion really increase energy recovery from biomass? *Appl. Energy* **2014**, *124*, 335–342. [CrossRef]
- 26. Donoso-Bravo, A.; Gajardo, P.; Sebbah, M.; Vicencio, D. Comparison of performance in an anaerobic digestion process: One-reactor vs two-reactor configurations. *Math. Biosci. Eng.* **2019**, *16*, 2447–2465. [CrossRef]
- 27. Blumensaat, F.; Keller, J. Modeling of two-stage anaerobic digestion using the IWA Anaerobic Digestion Model No. 1 (ADM1). *Water Res.* **2005**, *39*, 171–183. [CrossRef]
- 28. Najdenski, H.; Ilyin, V.; Angelov, P.; Hubenov, V.; Korshunov, D.; Kussovski, V.; Dimitrova, L.; Simeonov, I. Laboratory biodegradation of potential cellulose wastes generated during long-term manned space missions. *Ecol. Eng. Env. Prot.* **2019**, *1*, 71–78. [CrossRef]
- 29. Denchev, D.; Hubenov, V.; Simeonov, I.; Kabaivanova, L. Biohydrogen production from lignocellulosic waste with anaerobic bacteria. In Proceedings of the 4th International Conference on Water, Energy and Environment (ICWEE), Burgas, Bulgaria, 1–3 June 2016; pp. 7–12.
- 30. Simeonov, I. Mathematical modeling and parameters estimation of anaerobic fermentation process. *Bioprocess Eng.* **1999**, *21*, 377–381. [CrossRef]
- 31. Rosen, C.; Jeppsson, U. *Aspects on ADM1 Implementation within the BSM2 Framework;* Preprint CODEN:LUTEDX/(TEIE-7224); Department of Industrial Electrical Engineering and Automation, Lund University: Lund, Sweden, 2006; pp. 1–35.

Processes 2020, 8, 791 27 of 27

32. Hess, J.; Bernard, O. Design and study of a risk management criterion for an unstable anaerobic wastewater treatment process. *J. Process. Control* **2008**, *18*, 71–79. [CrossRef]

- 33. Wiggins, S. Introduction to Applied Nonlinear Dynamical Systems and Chaos; Springer: New York, NY, USA, 1990.
- 34. Bayen, T.; Gajardo, P. On the steady state optimization of the biogas production in a two-stage anaerobic digestion model. *J. Math. Biol.* **2019**, *78*, 1067–1087. [CrossRef] [PubMed]
- 35. Li, Q.; Li, Y. Coproduction of hydrogen and methane in a CSTR-IC two-stage anaerobic digestion system from molasses wastewater. *Water Sci. Technol.* **2019**, *79*, 270–277. [CrossRef] [PubMed]
- 36. Simeonov, I.; Karakashev, D. Mathematical Modeling of the anaerobic digestion including the syntrophic acetate oxidation. In Proceedings of the 7th Vienna International Conference on Mathematical Modeling (MATHMOD), Vienna, Austria, 14–17 February 2012; Volume 7, pp. 304–309.
- 37. Borisov, M.; Denchev, D.; Simeonov, I. Mathematical Modeling of a two-stage anaerobic digestion process with hydrogen and methane production using ADM1. *Ecol. Eng. Environ. Prot.* **2020**, *1*, 18–29.
- 38. Dareioti, M.A.; Kornaros, M. Effect of hydraulic retention time (HRT) on the anaerobic co-digestion of agro-industrial wastes in a two-stage CSTR system. *Bioresour. Technol.* **2014**, *167*, 407–415. [CrossRef]
- 39. Cavinato, C.; Bolzonella, D.; Faton, F.; Cecchi, F.; Pavan, P. Optimization of two-phase thermophilic anaerobic digestion of biowaste for hydrogen and methane production through reject water recirculation. *Bioresour. Technol.* **2011**, *102*, 8605–8611. [CrossRef]
- 40. Intanoo, P.; Rangsanvigit, P.; Malakul, P.; Chavadej, S. Optimization of separate hydrogen and methane production from cassava wastewater using two-stage upflow anaerobic sludge blanket reactor (UASB) system under thermophilic operation. *Bioresour. Technol.* **2014**, *173*, 256–265. [CrossRef]
- 41. Khan, M.A.; Ngo, H.H.; Guo, W.S.; Liu, Y.; Nghiem, L.D.; Hai, F.I.; Deng, L.J.; Wang, J.; Wu, Y. Optimization of process parameters for production of volatile fatty acid, biohydrogen and methane from anaerobic digestion. *Bioresour. Technol.* **2016**, 219, 738–748. [CrossRef]



© 2020 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/).