

The Control of Apparent Wettability on the Efficiency of Surfactant Flooding in Tight Carbonate Rocks

Authors:

Harris Sajjad Rabbani, Yossra Osman, Ibrahim Almaghrabi, Mohammad Azizur Rahman, Thomas Seers

Date Submitted: 2019-12-09

Keywords: water flooding, surfactant flooding, core-scale, carbonate rock, apparent wettability

Abstract:

In this research, a state-of-the-art experimental core flooding setup is used to assess the efficiency of surfactant flooding as an enhanced oil recovery (EOR) technique in tight carbonate rocks. Specifically, we investigate the role of apparent wettability in governing the effectiveness of surfactant flooding. A series of flooding experiments with well-defined boundary conditions were performed on the low permeability core plug samples of Indiana Limestone (calcite-cemented carbonate grainstones). Experiments were conducted on three samples exhibiting differing apparent wetting characteristics: strongly oil-wet, moderately oil-wet and weakly oil-wet. Initially, the oil-saturated core samples were flooded with brine until the residual oil saturation was achieved, with surfactant flooding performed as a tertiary recovery technique. Interestingly, our experimental results reveal that the efficiency of surfactant flooding increases with the degree of oil-wetness of the tight carbonate rocks. The strongly oil-wet core showed the highest recovery, while the weakly oil-wet core manifested the least additional oil recovery associated with surfactant flooding. Moreover, we provided a pore-scale argument that explains the macroscopic role of surfactant flooding in tight carbonate rocks. We hypothesized that at the pore-scale the presence of thin film plays a critical role in controlling the effectiveness of surfactant flooding in the strongly oil-wet tight carbonate rocks. Overall, we believe that our macroscopic study provides novel insight into the dynamics of surfactant flooding in tight carbonate reservoirs and can aid in optimizing the field development plans for oil recovery.

Record Type: Published Article

Submitted To: LAPSE (Living Archive for Process Systems Engineering)

Citation (overall record, always the latest version):

LAPSE:2019.1275

Citation (this specific file, latest version):

LAPSE:2019.1275-1

Citation (this specific file, this version):

LAPSE:2019.1275-1v1

DOI of Published Version: <https://doi.org/10.3390/pr7100684>

License: Creative Commons Attribution 4.0 International (CC BY 4.0)

Article

The Control of Apparent Wettability on the Efficiency of Surfactant Flooding in Tight Carbonate Rocks

Harris Sajjad Rabbani *, Yossra Osman, Ibrahim Almaghrabi, Mohammad Azizur Rahman  and Thomas Seers

Department of Petroleum Engineering, Texas A&M University at Qatar, Education City, P.O. Box 23874 Doha, Qatar; yossra.osman@qatar.tamu.edu (Y.O.); ibrahim.almaghrabi@qatar.tamu.edu (I.A.); marahman@tamu.edu (M.A.R.); thomas.seers@qatar.tamu.edu (T.S.)

* Correspondence: harris.rabbani@qatar.tamu.edu; Tel.: +97-46-677-5443

Received: 3 September 2019; Accepted: 18 September 2019; Published: 2 October 2019



Abstract: In this research, a state-of-the-art experimental core flooding setup is used to assess the efficiency of surfactant flooding as an enhanced oil recovery (EOR) technique in tight carbonate rocks. Specifically, we investigate the role of apparent wettability in governing the effectiveness of surfactant flooding. A series of flooding experiments with well-defined boundary conditions were performed on the low permeability core plug samples of Indiana Limestone (calcite-cemented carbonate grainstones). Experiments were conducted on three samples exhibiting differing apparent wetting characteristics: strongly oil-wet, moderately oil-wet and weakly oil-wet. Initially, the oil-saturated core samples were flooded with brine until the residual oil saturation was achieved, with surfactant flooding performed as a tertiary recovery technique. Interestingly, our experimental results reveal that the efficiency of surfactant flooding increases with the degree of oil-wetness of the tight carbonate rocks. The strongly oil-wet core showed the highest recovery, while the weakly oil-wet core manifested the least additional oil recovery associated with surfactant flooding. Moreover, we provided a pore-scale argument that explains the macroscopic role of surfactant flooding in tight carbonate rocks. We hypothesized that at the pore-scale the presence of thin film plays a critical role in controlling the effectiveness of surfactant flooding in the strongly oil-wet tight carbonate rocks. Overall, we believe that our macroscopic study provides novel insight into the dynamics of surfactant flooding in tight carbonate reservoirs and can aid in optimizing the field development plans for oil recovery.

Keywords: apparent wettability; carbonate rock; core-scale; surfactant flooding; water flooding

1. Introduction

With increasing demand for energy globally remaining largely unperturbed, the development of effective and economic enhanced oil recovery (EOR) techniques may prove to be a critical component in meeting future energy requirements. Forming a suite of processes targeted at reducing oil saturation within a reservoir below its intrinsic residual oil saturation (S_{or}), EOR techniques seek to displace greater volumes of trapped oil from within a given reservoir rock that are not readily mobilized by waterflooding alone [1]. Historically, carbonate reservoirs have formed an important focus for the deployment of EOR techniques. Constituting over half of the world's conventional oil reserves, carbonate reservoir settings are commonly characterized by poor recovery efficiency compared to conventional clastic plays [2]. The restricted producibility of many carbonate reservoirs is often due to the presence of highly heterogeneous pore networks therein. Multifarious burial and fluid charge histories, coupled with the innate chemical susceptibility of carbonate rock forming minerals (especially calcite) can often result in a high degree of pore-scale complexity in such settings, in terms of pore size distribution, pore geometry and pore wall wettability [3].

Indeed, one of the key challenges associated with carbonate reservoirs is their oil-wet nature. Owing to the surface charge that is favorable for the adsorption of polar compounds, many carbonate reservoirs exhibit oil-wet or mixed-wet tendencies [4–6]. Oil-wet conditions are highly problematic for the efficiency of conventional flooding techniques, such as gas flooding and water flooding [7,8]. As the fluid is injected into the reservoir, it moves preferentially through the path of the least capillary resistance, bypassing the oil phase and subsequently leaving it trapped in the reservoir. This behavior is attributable to the strong affinity of aged calcite pore walls within many carbonate hydrocarbon reservoirs, hence resulting in earlier water breakthrough and high water-oil ratios in comparison to equivalent clastic reservoirs (i.e., sandstones) [9].

It is widely recognized that surfactants can greatly alter fluid interfacial properties within a porous media, modifying the interfacial tension, hence reducing the strength of capillary forces [10–12]. Belhaj [13] demonstrated the control of the partitioning coefficient of a surfactant (the measure of the solubility of a surfactant at the fluid–fluid interface) on modifying surface forces at the fluid–fluid interface and the solid–fluid interface. They showed that an increase in partition coefficient results in a decrease in interfacial tension; moreover, it was concluded from their experimental results that the partition coefficient is directly related to the temperature and concentration of surfactant. Furthermore, Druetta and Picchioni [14] highlighted the impact of salinity of brine on the partition coefficient, concluding that for surfactant flooding EOR operation it is important to take account of the salinity of brine. As a consequence of the significant effect of surfactants on altering the interfacial properties, in many previous studies, surfactant flooding has been employed as an EOR technique for carbonate reservoirs [15–17]. Jarrahan et al. [18] performed a fundamental investigation to evaluate the surfactant induced wettability alteration in carbonate rocks. Their analytical results indicate that depending on the ionic structure, the surfactant will interact differently with carbonate surface. In comparison to the anionic surfactant the cationic surfactant has a more profound impact in altering the wettability of carbonate rocks from oil-wet to water-wet. Ahmadi and Shadizadeh [19] evaluated the impact of salt concentration on the adsorption of surfactant, called Zyziphus Spina Christi (ZSC) on the carbonate surface. In addition, they also delineated the effect of surfactant concentration on oil relative permeability and consequently on the efficiency oil recovery via core flooding experiments. Their results showed the promising effect of ZSC in improving oil recovery from carbonate core samples. In addition, it was shown that salinity has a detrimental impact on surfactant flooding as an increase in salinity of brine can lead to enhanced adsorption of surfactant on the carbonate surface.

In this research, we investigate the effect of the apparent wettability of tight carbonate rock on the efficiency of surfactant flooding at a fundamental level. It is important to note that in addition to surface wetting characteristics, the apparent wettability is also a function of pore geometry and surface roughness. We performed core-scale flooding experiments on low permeability (0.7–2.2 mD) Indiana Limestone core plug samples FD3H, FD4H and FD5H. The core samples displayed significantly different wetting characteristics, with FD3H being moderately oil-wet, FD4H showing weakly oil-wet characteristics and FD5H exhibiting strongly oil-wet characteristics. Our experimental results demonstrate that the efficiency of surfactant flooding is directly related to the apparent wettability of tight carbonate rocks. In summary, stronger affinity between the pore wall and oil phase results in higher recovery of hydrocarbons during surfactant flooding. In order to rationalize this intriguing behavior, we have presented a hypothesized pore-scale model that correlates Darcy scale observations with the pore-scale physics of the multiphase flow.

2. Experimental Procedure

The core flooding experiments were conducted using the Grace Instruments M9300 core flooding apparatus (Figure 1), with Indiana Limestone core samples with a diameter 3.8 cm and a length of 13 cm. Three samples of Indiana Limestone were utilized in these experiments, which are referred to as FD3H, FD4H and FD5H. The properties of core samples are given in Table 1. As seen in Table 1 the

core samples can be regarded as tight (i.e., low permeability) carbonate rocks, as their permeability ranges from 0.7 mD to 2.2 mD.

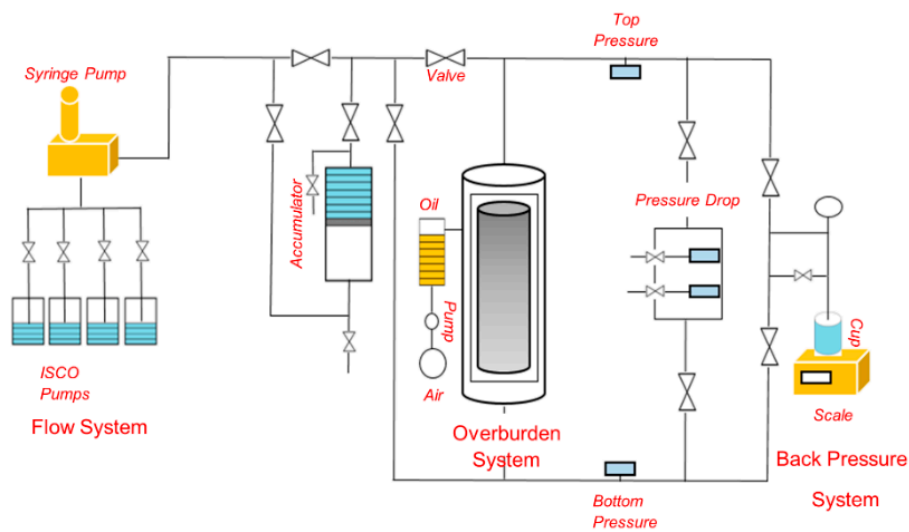


Figure 1. Schematic of M9300 core flow loop.

Table 1. Showing properties of core samples.

Sample	Porosity ϕ , (-)	Permeability K, mD
FD3H	0.13	2.2
FD4H	0.13	1.8
FD5H	0.21	0.7

Initially the core samples were saturated with brine, and later decane was injected at 2 cc/min to about 5 or 6 pore volumes (PV), to ensure that the irreducible water saturation (S_{wr}) was reached. After decane saturation, the core samples containing brine and decane were ready for waterflooding experiments. During water flooding, brine was injected at a flowrate of 1 cc/min for about 6 to 7 pore volumes to ensure that the residual oil saturation (S_{or}) was achieved. During each flooding experiment, the pressure drop across the sample was continuously monitored using paired pressure transducers.

Following the waterflooding procedure, the cores (still containing brine and decane) were prepared for surfactant flooding. The surfactant Platinum Foam Plus supplied by MI SWACO, Houston Texas, was used to perform surfactant flooding experiments. First, the surfactant mixture was prepared in the ratio of 1000 mL of water to 1 mL of liquid surfactant. The components were mixed together using a magnetic stirrer, yielding an overall concentration of 0.10 vol.%. Surfactant was injected at a flowrate of around 1 cc/min for about 7 to 8 pore volumes. Similar to decane and water flooding experiments, the pressure drop across the sample was monitored during surfactant flooding. Figure 2 provides a summary of the experimental protocol.

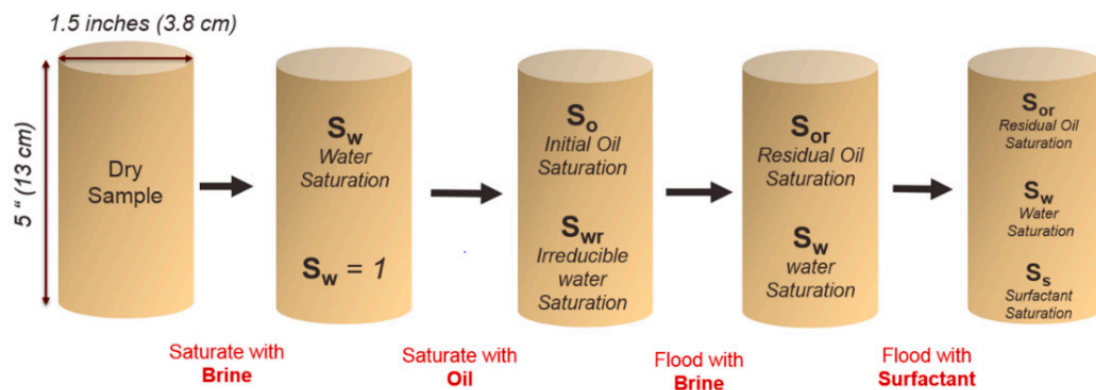


Figure 2. Experimental Procedure for flooding experiments.

3. Results and Discussion

3.1. Wettability Characterization

The pressure values recorded during decane flooding and water flooding can be seen in Figure 3.

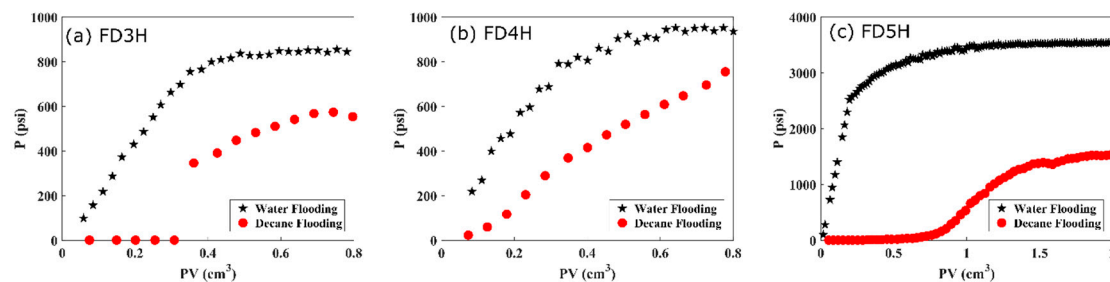


Figure 3. Pressure P (psi) during decane and water flooding as a function of pore volume PV (cm^3) injected.

As demonstrated by Figure 3, for all three core samples, pressure readings were lower during decane flooding than in the case of water flooding, indicating that it is relatively easier (lower displacement energy required) for the oil phase to enter and displace water from the pores than vice versa. Thus, Figure 3 indicates that all three core samples exhibited oil-wet characteristics.

Spontaneous imbibition ($p \approx 0$) is one of the key characteristics of multiphase flow in porous media that can aid us in distinguishing the degree of oil-wetness of core samples [20]. From Figure 3 it is clear that approximately 0.3 PV of the oil phase was spontaneously imbibed in FD3H, while in the case of FD5H, 0.6 PV was spontaneously imbibed. There was no appreciable spontaneous imbibition observed in core sample FD4H. On the basis of the quantities of oil spontaneously imbibed, it can be qualitatively deduced that FD3H, FD4H and FD5H manifest moderately oil-wet, weakly oil-wet and strongly oil-wet characteristics respectively.

3.2. Surfactant Flooding

Table 2 shows the percentage of total oil recovery during water flooding (R_w) and surfactant flooding (R_s), and the efficiency of surfactant flooding (E_s). The efficiency of surfactant flooding was computed as $E_s = R_s - R_w$.

Table 2. The results of flooding experiments.

Sample	Apparent Wettability	R_w (%)	R_s (%)	E_s (%)
FD3H	moderately oil-wet	28.78	30.23	1.45
FD4H	weakly oil-wet	33.8	34.17	0.37
FD5H	strongly oil-wet	37.58	40.22	2.64

From Table 2 it is evident that surfactant flooding has the capacity to improve oil recovery in tight carbonate rocks. It is important to note that the low percentages of efficiency of surfactant flooding in Table 2 can be ascribed to the relatively low concentrations of surfactant used for these experiments, as well as the low permeability of core samples. FD5H (strongly oil-wet) exhibits the highest efficiency of 2.64%, while FD4H (weakly oil-wet) shows the lowest improvement in surfactant induced oil recovery (0.37%). The comparison of measured pressure drop results during water flooding and surfactant flooding is shown in Figure 4.

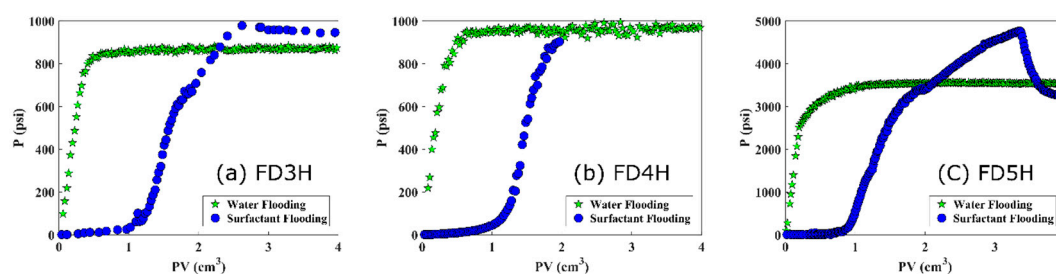


Figure 4. Pressure P (psi) during water flooding and surfactant flooding as a function of pore volume PV (cm^3) injected.

As seen in Figure 4, during surfactant flooding, in the cases of FD3H (moderately oil-wet) and FD5H (strongly oil-wet), after approximately injecting 2 PV the surfactant flood pressure exceeds the pressures previously achieved during water flood. However, for FD4H, surfactant and water flooding results in relatively consistent pressure values after injecting 2 PV of surfactant solution. This signifies that during flooding tests in FD3H (moderately oil-wet) and FD5H (strongly oil-wet), surfactants are able to alter interfacial tension, and therefore, weaken capillary forces (i.e., the forces responsible for the trapping of oil phase), as a consequence allowing the hydrocarbons to dislocate from the smaller pore channel as the surfactant flooding pressure becomes greater than water flooding pressure. However, in the case of FD4H (weakly oil-wet), as the surfactant flooding pressure and water flooding pressure at equilibrium are consistent, it demonstrates that surfactant flood's impact on oil recovery is almost negligible. Overall, the pressure results depicted in Figure 4 are closely related with the efficiency of surfactant flooding shown in Table 2.

Figure 5a presents the relationship between total oil recovery during water flooding and the permeability of the tested core samples. One can depict from Figure 5a that R_w is directly related to the permeability of the core. It is clear that oil recovery during water flooding increases as the permeability of core sample decreases. Although the trend is regular for the water flooding scenario, non-monotonic behavior can be observed in the case of E_s (Figure 5b). Figure 5b indicates that rather than the pore topological and geometric properties of porous media, there are other characteristics features that control the effectiveness of surfactant flooding. We have further shown the correlation between the E_s and the volume of oil phase spontaneously imbibed (PV_s) into the tested carbonate rock samples (Figure 6).

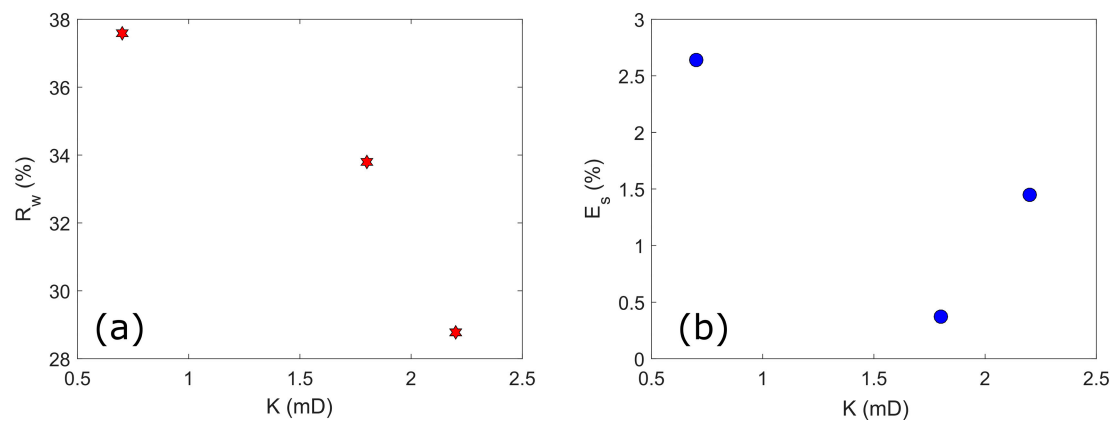


Figure 5. (a) Water flooding (R_w , %) and (b) surfactant flooding (E_s , %) against permeability of core samples K (mD).

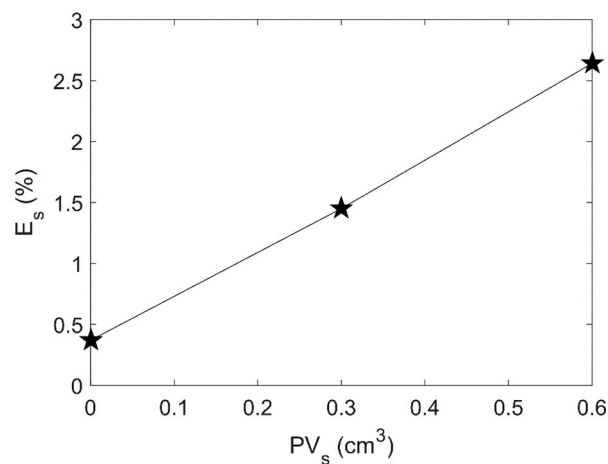


Figure 6. Linear relationship between E_s (%) and PV_s (cm³).

One can notice from Figure 6 that there is a linear trend between the E_s and PV_s . This indicates that the wetting properties of porous media have a strong impact on the efficiency of surfactant flooding. Our results demonstrate that the efficiency of surfactant flooding decreases with the degree of oil-wetness of the rock sample. The more strongly oil-wet core sample, FD5H, exhibits the highest values of E_s , while the weakly oil-wet core sample (FD4H) showed the lowest additional recovery as a result of surfactant flooding. The mathematical relationship that can be deduced from Figure 6 between E_s and PV_s can be written as $E_s = 3.78PV_s + 0.35$.

3.3. Pore Scale Hypothesis

In order to provide a rational explanation for the aforementioned experimental results, we present Figure 7, that illustrates pore scale hypothetical behavior of residual oil phase in strongly oil-wet sample FD5H and weakly oil-wet sample FD4H.

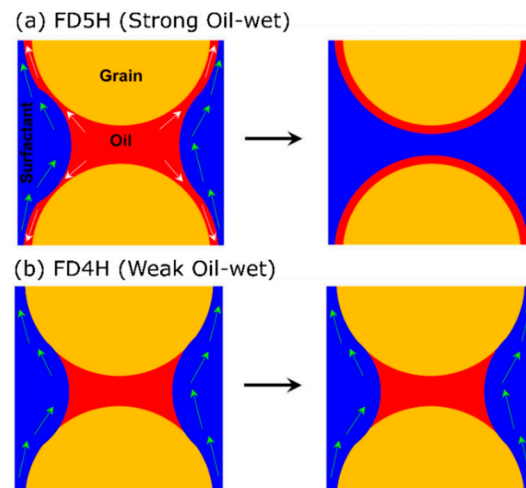


Figure 7. The hypothetical configuration of residual oil phase in (a) strong oil-wet core sample FD5H, and (b) weakly oil-wet core sample FD4H during surfactant flooding.

In core samples that have a strong affinity with the residing oil phase (illustrated in Figure 3), the oil remains behind the displacement front in small channels such as grooves, edges and pore throats during water flooding. Owing to the presence of thin film around the grains, in strongly oil wet samples, the oil phase remains connected throughout the porous medium, as shown in Figure 7a. However, in the case of the weakly oil-wet sample FD4H, there is no thin film present; therefore, the oil is trapped (remains discontinuous) within pore throats, as depicted in Figure 7b. We believe that this variation in the configuration of the oil phase, governed by the degree of oil-wetness of rock samples has a strong impact on the efficiency of surfactant flooding.

For the strongly oil-wet sample FD5H, as surfactants reduce interfacial tension (decreasing the influence of capillary forces), they allow the surrounding aqueous phase to invade the smaller pore throats; the presence of thin films facilitate the movement of the oil phase away from its initial location towards the outlet of the pore. It is important to note that in many previous studies the crucial role of thin films on the oil recovery process have been discussed [21,22]. In the case of weakly oil-wet porous media, FD4H, the oil phase remains disconnected and resides within pore throats as isolated ganglia (Figure 7b). For such conditions, in order to observe an increase in oil recovery, the isolated ganglia would have to be mobilized. As we are dealing with tight (low permeability) rock samples, both strong viscous and capillary forces inhibit the mobilization of ganglia. Though the injection of surfactants results in a decrease in interfacial tension, thus weakening the capillary forces acting at the fluid-fluid interface, it does not impact upon the viscous forces. Therefore, it can be hypothesized that the significant viscous dissipation at the three-phases' line of contact [23], in conjunction with viscous forces in the bulk volume, hinders the movement of ganglia and thus jeopardies the efficiency of surfactant flooding.

4. Summary and Conclusions

While major oil resources are global, the intricate geology associated with carbonate reservoirs makes them challenging to utilize. One of the main issues with carbonate reservoirs is their wetting characteristics: commonly oil-wet in nature, strong capillary forces trap the majority of the oil behind the displacement front in small channels, such as grooves, edges and pore throats, limiting oil recovery. The remarkable ability of surfactants to alter the interfacial tension, and hence lessen the strength of capillary forces, is well known from the literature. Therefore, surfactant flooding seems to be an ideal candidate for EOR (enhanced oil recovery) in carbonate reservoirs.

In this study, we uncovered the impact of apparent wettability of tight carbonate rocks on the efficiency of surfactant flooding. Flooding experiments were performed on three low permeability,

core samples of Indiana Limestone, referred to as FD3H, FD4H and FD5H. Although the three core samples manifested the oil-wet characteristics, the degree of oil-wetness varied between them. From the pressure drop measurements it was inferred that FD3H was moderately oil-wet, FD4H was weakly oil-wet, and FD5H was strongly oil-wet. Our core flooding experiments revealed that the apparent wettability of porous media plays a critical role in dictating the efficiency of surfactant flooding. The results demonstrate that the efficiency of surfactant flooding increases with the degree of oil-wetness of core samples. Furthermore, we forwarded a pore-scale conceptual model to rationalize the underlying behavior of surfactant flooding in tight carbonates, observed herein at the core scale.

5. Data Availability

The data presented in this manuscript will be available freely via sending a request to the corresponding author.

Author Contributions: H.S.R., Y.O., and I.A. performed the experiments. H.S.R., Y.O., and T.S. participated in writing the manuscript. H.S.R., T.S. and M.A.R. analyze the data. M.A.R. and T.S. secured funding for the project.

Funding: This publication was partly made possible by the grant NPRP10-0101-170091 from Qatar National Research Fund (a member of the Qatar Foundation).

Acknowledgments: Harris Sajjad Rabbani and Thomas Seers would like to acknowledge Total E&P. Statements made herein are solely the responsibility of the authors.

Conflicts of Interest: The authors declare no conflict of interest.

Abbreviations

EOR	Enhanced Oil Recovery
S_{wr}	Irreducible water saturation
K	Permeability
\varnothing	Porosity
PV	Pore volume
PV_s	Pore volume of oil phase spontaneously imbibed
P	Pressure
S_{or}	Residual oil saturation
R_s	Surfactant flooding oil recovery
E_s	Surfactant flooding recovery efficiency
R_w	Water flooding oil recovery

References

1. Lake, L. *Fundamentals of Enhanced Oil Recovery*; Society of Petroleum Engineers: Richardson, TX, USA, 2014.
2. Donaldson, E.; Chilingarian, G.; Yen, T. *Enhanced Oil Recovery*; Elsevier: New York, NY, USA, 1989.
3. Roehl, P.O.; Choquette, P.W. Perspectives on World-Class Carbonate Petroleum Reservoirs: ABSTRACT. *AAPG Bull.* **1985**, *69*, 148.
4. Chilingar, G.; Yen, T. Some Notes on Wettability and Relative Permeabilities of Carbonate Reservoir Rocks, II. *Energy Sources* **1983**, *7*, 67–75. [[CrossRef](#)]
5. Hirasaki, G.; Zhang, D. Surface Chemistry of Oil Recovery from Fractured, Oil-Wet, Carbonate Formations. *SPE J.* **2004**, *9*, 151–162. [[CrossRef](#)]
6. Zhang, P.; Austad, T. Wettability and oil recovery from carbonates: Effects of temperature and potential determining ions. *Colloids Surf. A Physicochem. Eng. Asp.* **2006**, *279*, 179–187. [[CrossRef](#)]
7. Rabbani, H.; Joekar-Niasar, V.; Pak, T.; Shokri, N. New insights on the complex dynamics of two-phase flow in porous media under intermediate-wet conditions. *Sci. Rep.* **2017**, *7*, 4584. [[CrossRef](#)] [[PubMed](#)]
8. Rabbani, H.; Zhao, B.; Juanes, R.; Shokri, N. Pore geometry control of apparent wetting in porous media. *Sci. Rep.* **2018**, *8*, 15729. [[CrossRef](#)] [[PubMed](#)]
9. Anderson, W. Wettability Literature Survey-Part 6: The Effects of Wettability on Waterflooding. *J. Pet. Technol.* **1987**, *39*, 1605–1622. [[CrossRef](#)]

10. Rabbani, H.; Seers, T.; Guerillot, D. Analytical Pore-Network Approach (APNA): A novel method for rapid prediction of capillary pressure-saturation relationship in porous media. *Adv. Water Resour.* **2019**, *130*, 147–156. [[CrossRef](#)]
11. Olayiwola, S.; Dejam, M. A comprehensive review on interaction of nanoparticles with low salinity water and surfactant for enhanced oil recovery in sandstone and carbonate reservoirs. *Fuel* **2019**, *241*, 1045–1057. [[CrossRef](#)]
12. Ahmadi, M.; Zendehboudi, S.; Shafiei, A.; James, L. Nonionic Surfactant for Enhanced Oil Recovery from Carbonates: Adsorption Kinetics and Equilibrium. *Ind. Eng. Chem. Res.* **2012**, *51*, 9894–9905. [[CrossRef](#)]
13. Belhaj, A.; Elraies, K.; Alnarabiji, M.; Shuhli, J.; Mahmood, S.; Ern, L. Experimental Investigation of Surfactant Partitioning in Pre-CMC and Post-CMC Regimes for Enhanced Oil Recovery Application. *Energies* **2019**, *12*, 2319. [[CrossRef](#)]
14. Druetta, P.; Picchioni, F. Simulation of Surfactant Oil Recovery Processes and the Role of Phase Behaviour Parameters. *Energies* **2019**, *12*, 983. [[CrossRef](#)]
15. Adams, W.; Schievelbein, V. Surfactant Flooding Carbonate Reservoirs. *SPE Reserv. Eng.* **1987**, *2*, 619–626. [[CrossRef](#)]
16. Ahmadi, M.; Shadizadeh, S. Nano-surfactant flooding in carbonate reservoirs: A mechanistic study. *Eur. Phys. J. Plus* **2017**, *132*, 246. [[CrossRef](#)]
17. Firozjahi, A.M.; Derakhshan, A.; Shadizadeh, S.R. An investigation into surfactant flooding and alkaline-surfactant-polymer flooding for enhancing oil recovery from carbonate reservoirs: Experimental study and simulation. *Energy Sources Part A Recover. Util. Environ. Eff.* **2018**, *40*, 2974–2985. [[CrossRef](#)]
18. Jarrahian, K.; Seiedi, O.; Sheykhan, M.; Sefti, M.; Ayatollahi, S. Wettability alteration of carbonate rocks by surfactants: A mechanistic study. *Colloids Surf. A Physicochem. Eng. Asp.* **2012**, *410*, 1–10. [[CrossRef](#)]
19. Ahmadi, M.; Shadizadeh, S. Spotlight on the New Natural Surfactant Flooding in Carbonate Rock Samples in Low Salinity Condition. *Sci. Rep.* **2018**, *8*, 10985. [[CrossRef](#)] [[PubMed](#)]
20. Ma, S.; Zhang, X.; Morrow, N.; Zhou, X. Characterization of Wettability from Spontaneous Imbibition Measurements. *J. Can. Pet. Technol.* **1999**, *38*. [[CrossRef](#)]
21. Salathiel, R. Oil Recovery by Surface Film Drainage in Mixed-Wettability Rocks. *J. Pet. Technol.* **1973**, *25*, 1216–1224. [[CrossRef](#)]
22. Blunt, M. Pore Level Modeling of the Effects of Wettability. *SPE J.* **1997**, *2*, 494–510. [[CrossRef](#)]
23. Snoeijer, J.; Andreotti, B. Moving Contact Lines: Scales, Regimes, and Dynamical Transitions. *Annu. Rev. Fluid Mech.* **2013**, *45*, 269–292. [[CrossRef](#)]



© 2019 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<http://creativecommons.org/licenses/by/4.0/>).