Correction: Study of n-Butyl Acrylate Self-Initiation Reaction Experimentally and via Macroscopic Mechanistic Modeling Processes 2016, 4, 15

Authors:

Ahmad Arabi Shamsabadi, Nazanin Moghadam, Sriraj Srinivasan, Patrick Corcoran, Michael C. Grady, Andrew M. Rappe, Masoud Soroush

Date Submitted: 2018-07-30

Keywords: 10.3390/pr4020015, doi

Abstract:

We wish to correct Table 5 of the published paper in Processes [1].[...]

Record Type: Published Article

Submitted To: LAPSE (Living Archive for Process Systems Engineering)

Citation (overall record, always the latest version):

LAPSE:2018.0176

Citation (this specific file, latest version):

LAPSE:2018.0176-1

Citation (this specific file, this version):

LAPSE:2018.0176-1v1

DOI of Published Version: https://doi.org/10.3390/pr4030026

License: Creative Commons Attribution 4.0 International (CC BY 4.0)





Correction

Correction: Study of *n*-Butyl Acrylate Self-Initiation Reaction Experimentally and via Macroscopic Mechanistic Modeling *Processes* 2016, *4*, 15

Ahmad Arabi Shamsabadi ¹, Nazanin Moghadam ¹, Sriraj Srinivasan ², Patrick Corcoran ¹, Michael C. Grady ³, Andrew M. Rappe ⁴ and Masoud Soroush ^{1,*}

- Department of Chemical and Biological Engineering, Drexel University, Philadelphia, PA 19104, USA; ahmad.arabishamsabadi@drexel.edu (A.A.S.); nznnmoghadam@gmail.com (N.M.); phc28@drexel.edu (P.C.)
- ² Arkema Inc., 900 1st Avenue, King of Prussia, PA 19406, USA; sriraj.s@gmail.com (S.S.)
- ³ DuPont Experimental Station, Wilmington, DE 19803, USA; michael.c.grady-1@dupont.com
- The Makineni Theoretical Laboratories, Department of Chemistry, University of Pennsylvania, Philadelphia, PA 19104-6323, USA; rappe@sas.upenn.edu
- * Correspondence: ms1@drexel.edu; Tel.: +1-215-895-1710

Academic Editor: Michael Henson

Received: 8 August 2016; Accepted: 10 August 2016; Published: 16 August 2016

We wish to correct Table 5 of the published paper in Processes [1]. In A_i in this work should be 14.86. The correct version of Table 5 is given on the following content. We apologize for any inconvenience caused to readers of Processes by this change.

Table 5. Experimentally-estimated and theoretical values of k_i (M⁻¹·s⁻¹), E_i (kJ·mol⁻¹) and A_i (M⁻¹·s⁻¹).

Temperature	This Work	Theoretical [11]	Theoretical [12]
T	k _i	k_i	k_i
413	3.30×10^{-15}	2.81×10^{-18}	1.04×10^{-14}
433	2.20×10^{-14}	2.86×10^{-17}	4.72×10^{-14}
453	4.00×10^{-13}	2.37×10^{-16}	1.95×10^{-13}
473	1.50×10^{-12}	1.64×10^{-15}	7.11×10^{-13}
493	6.80×10^{-12}	9.74×10^{-15}	2.34×10^{-12}
Parameter	This Work	Theoretical [11]	Theoretical [12]
E_i	165.51 ± 4.52	172.50	115.00
$\ln A_i$	14.86 ± 1.20	9.68	1.38

References

1. Arabi Shamsabadi, A.; Moghadam, N.; Srinivasan, S.; Corcoran, P.; Grady, M.C.; Rappe, A.M.; Soroush, M. Study of n-Butyl Acrylate Self-Initiation Reaction Experimentally and via Macroscopic Mechanistic Modeling. *Processes* **2016**, *4*, 15.



© 2016 by the authors; licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC-BY) license (http://creativecommons.org/licenses/by/4.0/).